

# City of Hialeah Reverse Osmosis Water Treatment Plant Hydrogeologic Investigation of the Upper Floridan Aquifer



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### Section 1 Executive Summary

### 1.1 Executive Summary

The proposed City of Hialeah reverse osmosis water treatment plant project is a major component of the City's capital improvement program. Reverse osmosis treatment of brackish groundwater will be used to help meet the potable water demands of the City's annexation area and growth within the Miami-Dade Water and Sewer Department (MDWASD) service area. The City and MDWASD recognized the need for additional water and have partnered together to share the cost of constructing the new treatment facility. Large-scale development of fresh groundwater sources near the major population centers in South Florida is no longer feasible due to competing uses, the potential for saline water intrusion problems, and perceived impacts to wetlands or other environmentally sensitive areas. The South Florida Water Management District (SFWMD) encourages the use of alternative water supplies and has supported this project by issuing grants to help cover portions of the construction costs. The design capacity of the water treatment facility is 17.5 million gallons per day (Mgd) and it is anticipated that the initial plant capacity will be 10.0 Mgd. Raw water pumpage rates of 23.33 and 13.33 Mgd will be required based on an assumed recovery efficiency of 75% for the reverse osmosis process. The development of brackish water for public supply purposes includes three major components: a reverse osmosis treatment facility, a concentrate disposal system, and the raw water supply wellfield. This report addresses the testing and evaluations conducted in order to design the raw water supply wellfield.

Brackish water is known to occur within the Floridan Aquifer at depths of approximately 1,000 feet below land surface and greater in Dade County based on the results of previous hydrogeologic investigations conducted in the area. However, aquifer yield and water quality conditions can vary greatly over relatively small distances and with depth. A detailed program of testing and analyses was developed to: identify one or more groundwater sources suitable for reverse osmosis treatment; assess the quantity of water available for public supply use; estimate the pumpage induced, long-term changes in water quality with time; and to collect sufficient data to insure the long-term viability of the water source(s). The hydrogeologic investigation included the drilling of five deep test wells to various depths near the proposed water treatment plant site, collection of aquifer yield and water quality data, computer modeling to evaluate the impacts of the proposed pumpage, and development of wellfield design scenarios. The proposed water treatment plant and wellfield site is located near the intersection of NW 170<sup>th</sup> Street and NW 97<sup>th</sup> Avenue in Hialeah.

The Floridan Aquifer is suitable for use as a raw water supply source for a reverse osmosis treatment facility based on the results of the investigation conducted. The Floridan Aquifer was penetrated to a depth of 1,733 feet below land surface (bls) in a test/production well constructed at the plant site. The upper portion of the Floridan Aquifer was tested between the depths of approximately 1,080 feet and 1,490 feet bls during the investigation. The transmissivity was determined to range from approximately 3,700 to 6,200 ft<sup>2</sup>/day indicating the aquifer has a moderate yield capacity. The total dissolved solids (TDS) and chloride concentration of the water produced during aquifer performance testing of the Upper Floridan Aquifer were

approximately 3,500 milligrams per liter (mg/L) and 1,650 mg/L, respectively. Salinity levels increase with depth within the aquifer system. The Upper Floridan Aquifer is considered the primary source of feedwater for the proposed reverse osmosis water treatment facility. A limestone unit encountered between the approximate depths of 497 and 620 feet during drilling was also tested as part of the investigation. The unnamed limestone unit has water quality similar to the underlying Upper Floridan Aquifer, but very low yield potential so it is not considered a viable source of feedwater for the reverse osmosis plant.

Hydraulic and solute transport computer models of the Floridan Aquifer System were developed to estimate water level and water quality changes that might occur due to long term brackish water withdrawals. A proposed wellfield alignment consisting of 14 production wells producing a maximum of 23.33 Mgd was simulated for a period of 30 years. The model results indicate that aquifer drawdowns of up to 107 feet may occur due to sustained withdrawals at the maximum anticipated pumping rate. Drawdowns in the production wells will be greater due to well losses and pumping water levels of 150 feet below land surface or more may occur near the center of the wellfield alignment. The average simulated TDS of the water produced from the wellfield increases from approximately 3,500 mg/L to 4,310 mg/L over a 30 year period based on the model results. Sensitivity analyses were conducted in order to evaluate how variations of the model input parameters impact the model results and thus estimates of future water quality from the wellfield. A range of estimated water quality changes that might occur from the wellfield over a 30 year period were determined based on the sensitivity analyses results. The upper end scenario indicates that average TDS values from the wellfield may increase to as high as 6,420 mg/L over a 30 year period due to sustained pumpage at a rate of 23.33 Mgd.

The water level drawdown and water quality projections presented in this report are based primarily on data collected during a hydrogeologic investigation conducted in the vicinity of the proposed reverse osmosis plant. The investigation was limited in size and scope due to the typical time and budget restraints associated with a project of this type. Existing data from other areas in South Florida were also utilized but the amount of existing information available is not extensive. The designers of the membrane process and plant operators should consider the anticipated range of water quality from the wellfield. It would be prudent to specify piping and other equipment capable of treating water with the highest salinity anticipated. The model results should be compared to actual water level and water quality data obtained from the wellfield after it has been in operation for approximately two years. The model should be updated and calibrated to the actual operational data at that time and additional simulations conducted to predict future wellfield performance, if necessary.

### 1.2 Conclusions

• An investigation of the Upper Floridan Aquifer was successfully performed at the future City of Hialeah reverse osmosis water treatment plant site. Brackish groundwater from the Upper Floridan Aquifer will adequately supply the future raw water demands of the water treatment plant. The reverse osmosis treatment process is anticipated to be 75 percent efficient and will require a raw water supply of 13.33 Mgd to produce a finished water flow capacity of 10 Mgd and a raw water supply of 23.33 Mgd to produce a finished water flow capacity of 17.5 Mgd.

- One test-production well (TP-1) and three monitor wells (F-1, F-2, and F-3) were constructed in the Upper Floridan Aquifer at the project site. The wells were constructed and tested to evaluate yield and quality of groundwater from the Upper Floridan Aquifer. The test-production well was designed to enable conversion to a future production well.
- An unnamed aquifer was identified in the Intermediate Confining Unit during testproduction and monitor well construction at the project site. Recognizing the potential value of a freshwater source to supplement brackish groundwater, a test well was subsequently constructed and cased with an open hole ranging from 497 to 620 feet below land surface (bls). The aquifer had low yield characteristics and similar water quality to the Upper Floridan Aquifer. The aquifer is not considered a viable raw water source for the water plant due to its low yield potential.
- The Upper Floridan Aquifer consists primarily of porous limestone and dolomite. The top of the aquifer occurs at an approximate depth of 1,050 feet bls at the project site. Dissolved chloride and total dissolved solids (TDS) were measured at concentrations of approximately 1,650 milligrams per liter (mg/L) and 3,500 mg/L, respectively, in the test-production well. The overall testing results indicate that the transmissivity of the production zone at the project site ranges from approximately 3,700 to 6,200 ft<sup>2</sup>/d and that the storage coefficient is roughly 5 X 10<sup>-4</sup>. The calculated leakance values ranged from 1.5 X 10<sup>-4</sup> to 4.1 X 10<sup>-4</sup> d<sup>-1</sup>.
- The potentiometric surface in the Upper Floridan Aquifer ranged between an approximate altitude of 47 and 52 feet referenced to the National Geodetic Vertical Datum of 1929, which is equivalent to approximately 39 to 44 feet above land surface. Wells constructed in the Upper Floridan Aquifer are artesian and will flow freely at land surface.
- Hydraulic and solute transport computer models of the Upper Floridan Aquifer were developed to simulate aquifer response in terms of drawdown and water quality changes that may result from a raw water withdrawal rate of 13.33 and 23.33 Mgd. A proposed design for the wellfield was evaluated using a groundwater flow model to estimate the amount of drawdown that would occur in the aquifer based on the proposed withdrawal rates. A solute transport model was used to estimate salinity changes over a 30-year period. A maximum drawdown of approximately 107 feet is anticipated in the aquifer based on the modeling results with a 23.33 Mgd withdrawal rate. The solute modeling indicated that TDS concentrations from the wellfield may increase from an average initial value of 3,500 mg/L to approximately 4,310 mg/L over the 30-year period at a withdrawal rate of 23.33 Mgd.
- A sensitivity analysis was performed to quantify the uncertainty in the calibrated model caused by uncertainty in the estimates of aquifer parameters, stress, and boundary conditions. The sensitivity analysis was performed by varying longitudinal and vertical dispersivities, effective porosity, and the vertical hydraulic conductivity value of the Middle Confining Unit that directly underlies the proposed pumping zone. The sensitivity analysis results indicate that the model is most sensitive to the vertical

hydraulic conductivity of the Middle Confining Unit. Based on the results of sensitivity analysis, the average TDS concentration of raw water withdrawn from the wellfield could reach 6,420 mg/L. The maximum drawdown may reach 117 ft according to one sensitivity analysis run in which the vertical hydraulic conductivity of the underlying confining unit was reduced by 50%.

### 1.3 Recommendations

- A total of 14 production wells should be utilized to meet a raw water demand of 23.33 Mgd. Two of the 14 production wells should serve as backup supply wells that provide a redundant supply capacity in the event one or more of the production wells are not operational. All of the production wells and backup wells should be used on a rotational basis. The individual wells should be constructed with a minimum spacing of approximately 1,250 feet and a configuration that maximizes separation to the extent possible. A proposed wellfield alignment is provided in this report.
- The production wells should be constructed with a cased depth of approximately 1,080 feet bls with an open hole to approximately 1,490 feet bls. The depth to the top of the Upper Floridan Aquifer will vary across the proposed wellfield site. Therefore, the final casing and open hole depths should be determined by a geologist based on actual lithology at each individual production well site. A cost effective well design should be considered that enables installation of a submersible pump while minimizing casing diameter below the submersible pump. A common design includes a larger diameter (e.g., 16-inch diameter) casing installed to approximately 200 feet, in which the submersible pump is installed, and below which the casing size is reduced (e.g., 12-inch diameter).
- Step-drawdown pump tests should be performed on each newly constructed production well. Pumping rates during the tests should range between approximately 1,000 and 1,500 gallons per minute (gpm). Specific capacity values calculated based on the test results can be used to assess individual well yields and confirm withdrawal rates. All of the production wells should be disinfected following development and step-drawdown pump testing. Submersible pumps equipped with electric motors should be installed in the wells with the intakes set at depths determined based on the specific capacity test results. Variable frequency drive controllers should be used to adjust production rates as needed. Intake setting depths of 100 to 150 feet bls are anticipated. Recommended withdrawal rates for the Upper Floridan Aquifer production wells when the reverse osmosis water treatment plant is in full operation may range between 1.5 to 2.0 Mgd.
- A letter modification of the existing South Florida Water Management District water use permit should be submitted to document the number and locations of proposed production and monitor wells. The existing Miami-Dade water use permit (SFWMD No. 13-00017-W) includes the proposed withdrawals from the Upper Floridan aquifer at the project site.

- The monitor wells constructed during this investigation should be preserved and be used to monitor future water levels and water quality, if possible. Monitoring may be required by limiting conditions of the SFWMD water use permit. An additional monitoring well located distant from the wellfield should be designated as a background well.
- The project design and management strategy should be sufficiently flexible to accommodate the maximum salinity change predicted by the model. Based on the results of sensitivity analysis, the average TDS concentration of raw water withdrawn from 14 proposed wells could reach 6,420 mg/L after 30 years of pumpage ranging from 13.33 to 23.33 Mgd. It is recommended that the design of the 17.5 Mgd RO plant accommodate raw water with a TDS concentration of at least 6,420 mg/L.
- Additional sensitivity analyses and model calibration may be necessary when more operational data from the water treatment plant become available. The hydraulic and solute transport models should be updated and recalibrated when more data, especially the actual drawdown and water quality data, become available after the proposed RO plant is in operation, if necessary. Actual monitoring data add to the reliability of the predicted changes in water quality.

### Section 2 Introduction

### 2.1 Background

The Biscayne Aquifer is an unconfined, highly permeable aquifer, underlying approximately 4,000 square miles of Miami-Dade, Broward, and southeastern Palm Beach counties (Miller, 1986). Due to the aquifer productivity and the relative ease with which this water can be withdrawn and treated, it is the dominant source of public water supply for many utilities in southeast Florida. In addition, the aquifer is widely used for commercial, industrial, and agricultural purposes. The high permeability and unconfined nature of the aquifer causes it to be susceptible to contamination by salt-water intrusion and infiltration of contaminants from the extensive system of canals used for drainage and flood control and other anthropogenic sources (Miller, 1986).

Due to the large demands from this aquifer, as well as potential impacts accompanying continued and additional withdrawals from the Biscayne Aquifer, the SFWMD continues to encourage utilities to seek alternative water supplies, including RO treatment of brackish water. The SFWMD, along with funds from the State Water Protection and Sustainability Program, provide funding to off-set the cost of alternative water supply projects. To meet the alternative use objective, the City of Hialeah and Miami-Dade County entered into a joint participation agreement in December 2007 to develop a 17.5 Mgd RO water treatment plant (WTP). A site located near the Florida Turnpike was selected for the RO plant and wellfield (**Figure 2-1**). The plant will treat brackish water obtained from production wells completed in the Upper Floridan Aquifer. The RO treatment process is anticipated to be approximately 75% efficient and therefore, a raw water supply of approximately 23.3 Mgd will be required at buildout. A 10 Mgd RO WTP capacity will initially be built, which will require a 13.33 Mgd raw water supply.

Hydrogeologic data for the Upper Floridan Aquifer near the proposed RO WTP is relatively limited. Therefore, an investigative program was conducted to obtain information on the Upper Floridan Aquifer and identify the preferred production zone for the wellfield. A test-production well (TP-1) was constructed to serve this purpose. In addition, three Upper Floridan Aquifer monitor wells were constructed, at variable distances from TP-1, to function as monitor wells during aquifer performance testing to obtain additional water quality data, and to assess aquifer heterogeneity. The construction and testing of these wells is documented in this report.

### 2.2 Scope of Work

The scope of work completed for this project included: 1) a compilation and review of available geologic, hydraulic, and water quality data, 2) construction of test wells in the Upper Floridan Aquifer, 3) collection and analysis of lithologic, geophysical, and water quality data during well construction and testing, 4) aquifer performance testing to determine pertinent aquifer hydraulic characteristics for the Upper Floridan Aquifer, 5) groundwater modeling to estimate drawdown and water quality impacts associated with wellfield pumpage, and 6) preparation of a report summarizing the results of the investigation. In addition to testing of the Upper Floridan Aquifer, an unnamed aquifer in the Intermediate Aquifer System was identified during this



Schlumberger WATER SERVICES Figure 2-1 City of Hialeah Site Location Map investigation. Recognizing the potential value of an alternative source to blend with the brackish groundwater from the Upper Floridan Aquifer, approval to construct a test well in the unnamed aquifer was granted by the City of Hialeah.

Additional tasks that remain to be completed for wellfield development as part of the overall RO WTP expansion project include: 1) preparation of technical specifications for construction of the RO WTP production wells and assistance in the bid process, 2) modification of the water use permit to include the new wellfield, 3) provision of construction services during production well construction, and 4) start-up services for the raw water supply wellfield. The Miami-Dade County water use permit contains an allocation for this wellfield. It is anticipated that a minor modification must be obtained to show a revised configuration for the wellfield.

### Section 3 Methods of Investigation

### 3.1 Existing Data Compilation and Review

Geologic and hydrologic data available near the project site were compiled and evaluated as part of a preliminary assessment of the Upper Floridan Aquifer. The data sources included publications from the United States Geological Survey (USGS), Florida Geological Survey (FGS), and SFWMD, which are referenced in Section 6 of this report. Information obtained from the preliminary assessment was used to design the hydrogeologic investigation and prepare technical specifications for test-production well (TP-1) and monitor wells (F-1, F-2 and F-3).

### 3.2 Test Drilling

Diversified Drilling Corporation was selected by the City of Hialeah, based on a competitive request for proposals evaluation, to construct the test-production and monitor wells. Schlumberger Water Services (SWS) prepared the well designs and technical specifications, and supervised all drilling operations, casing installation and cementing, water quality testing, geophysical logging, and aquifer performance testing. The wells were designed and constructed in accordance with the requirements of Chapter 62-532, Florida Administrative Code and generally in accordance with the American Water Works Association Standards for Water Wells (AWWA A100-06). An aerial photograph showing the locations of the test-production well and monitor wells is provided as **Figure 3-1**.

All wells were constructed using conventional closed-circulation mud rotary drilling through the Surficial Aquifer System and the Intermediate Confining Unit, and open-circulation reverse air drilling through the Upper Floridan Aquifer and Middle Confining Unit. Discharge water generated during open-circulation reverse air drilling was conveyed through a settling tank, filter, and piped to an existing mine pit located to the west of the project site. A permit was obtained from Miami-Dade Department of Environmental Management (DERM) prior to discharging to the existing mine pit.

### 3.2.1 Upper Floridan Aquifer Test-Production Well (TP-1)

Construction of test-production well TP-1 was initiated on March 29, 2009 and completed on June 7, 2009. The test-production well was constructed to evaluate the Upper Floridan Aquifer and allow for later conversion of the well to a production well to supply raw water to the RO WTP.

A 42-inch borehole was drilled to 19 feet below land surface (bls) and a 34-inch diameter (0.375inch wall steel) pit casing was grouted at a depth of 18 feet bls. A 32-inch diameter borehole was then drilled using mud-rotary drilling to a depth of 220 feet bls. A 26-inch diameter (0.375inch wall steel) surface casing was then pressure grouted in place at a depth of 212 feet bls. A 25-inch diameter borehole was then drilled using the mud-rotary drilling method and a 17.4-inch outside diameter Certa-Lok SDR-17 PVC casing was pressure grouted in place with neat cement to a depth of 1,082 feet bls using a narrow diameter pipe that was run through the well casing.





The remaining annular space was grouted in about 10 stages using a collarless tremie pipe. The number of stages was necessary to avoid damage to the PVC pipe from heat of hydration created during grouting. Some bentonite was allowed in the tremie installed grout stages to reduce heat of hydration. The drilling rig was converted to reverse air and a 12.25-inch diameter pilot hole was completed to a depth of 1,733 feet bls. Discharge from reverse air drilling was conveyed through a settling tank, filter, and piped to an existing mine pit located to the west of the project site.

The full pilot hole was logged using advanced borehole geophysics to obtain the best quality hydraulic and water quality data possible. The pilot hole was then back-plugged with cement to a depth of 1,490 feet bls and then reamed using a 15-inch diameter bit to a total depth of 1,490 feet bls. The test-production well was purposely overdrilled to obtain water quality data from the zone below the anticipated open hole portion of the production well. Water quality data obtained from the overdrilled pilot hole was used in a three-dimensional density-dependent groundwater flow and solute transport model developed for this project. The final well depth was determined by an on-site SWS hydrogeologist based on lithology, geophysical data, water quality, and well yield. An as-built diagram for the test-production well are provided in **Table 3-1**. The well was air developed for several hours from within the casing until the produced water was clear and free of sediment. Rossum Sand Sampler and Silt Density Index (SDI) tests were performed during the aquifer performance test.

### 3.2.2 Upper Floridan Aquifer Monitor Wells (F-1, F-2, and F-3)

Three Upper Floridan Aquifer monitor wells (F-1, F-2, and F-3) were constructed on the project site (**Figure 3-1**) at varying distances from test-production well (TP-1). The monitor wells were constructed to measure water level changes in response to pumping of the test-production well and obtain spatial water quality and hydrogeologic data. The monitor wells were constructed by Diversified Drilling Corporation under the supervision of SWS staff.

The monitor wells were constructed using similar methods for each well. A 25-inch diameter borehole was constructed to approximately 20 feet bls, using mud-rotary drilling. A 20-inch diameter pit (0.375-inch steel wall) casing was installed to a depth of 19 feet bls in each monitor well. An 18.75-inch diameter borehole was drilled and a 12-inch (0.250-inch steel wall) casing was grouted to a depth of 267 feet bls in F-1, 210 feet bls in F-2, and 209 feet bls in F-3. A 12inch nominal diameter borehole was then drilled to approximately 1,080 feet bls or greater to the top of the Upper Floridan Aquifer using mud-rotary drilling. The variation in the borehole depth was based on localized changes in the geology as observed by the on-site geologist. A 6.625inch Certa-Lok SDR-17 PVC casing was grouted to a depth of 1,082, 1,080, and 1,085 feet bls in monitor wells F-1, F-2, and F-3, respectively. A nominal 6-inch diameter open borehole was drilled using reverse air to a depth of 1,489, 1,358, and 1,368 feet bls for monitor wells F-1, F-2, and F-3, respectively. The final well depth was determined by an on-site SWS hydrogeologist based on lithology, geophysical data, water quality, and well yield. The initial cement stage was installed by pressure grouting using a narrow diameter PVC pipe that was run through the well casing, while the remaining annular space was grouted in stages using a collarless tremie pipe. As-built diagrams for the monitor wells are provided as Figure 3-3. As-built construction details





#### Table 3-1 Well Construction Details

Well ID	Lat/Long	Distance from TP-1	<b>Construction Date</b>	Pit Casing	Surface Casing	Final Casing	Open Hole
TP-1	25.9241/80.3699		3/29/09-6/7/09	34-in, 18 ft	26-in, 212 ft	17.4-in, 1082 ft	15-in, 1490 ft
F-1	25.9239/80.3700	93	5/14/09-6/29/09	20-in, 19 ft	12-in, 210 ft	6.625-in, 1083 ft	6-in, 1488 ft
F-2	25.9239/80.3679	731	3/27/09-5/14/09	20-in, 19 ft	12-in, 210 ft	6.625-in, 1080 ft	6-in, 1358 ft
F-3	25.9275/80.3699	1266	3/30/09-5/21/09	20-in, 19 ft	12-in, 209 ft	6.625-in, 1085 ft	6-in, 1368 ft
T-2	25.9238/80.3677	732	8/19/09-9/3/9	12.75-in, 22 ft		6.625-in, 497 ft	6-in, 620 ft





for the monitor wells are provided in **Table 3-1**. The wells were air developed for several hours from within the casing until the produced water was relatively clear and free of sediment.

### 3.2.3 Unnamed Aquifer Test Well (T-2)

An unnamed aquifer in the Intermediate Confining Unit was observed at a depth of approximately 480 to 633 feet bls during construction of the test-production and monitor wells. The aquifer consists of limestone characterized as a moderately hard, sandy, wackestone with medium apparent moldic macroporosity. Recognizing the potential value of an alternative source to blend with brackish groundwater from the Upper Floridan Aquifer, the City of Hialeah granted approval to construct a test well in the unnamed aquifer. A test well (T-2) was subsequently installed and tested to determine the hydraulic characteristics and water quality in the unnamed aquifer and its potential use as a raw water source.

A 12.75-inch diameter steel surface casing was set to 22 feet bls. A 6-inch diameter PVC casing was then grouted to 497 feet bls with an open hole constructed to 620 feet bls. An as-built diagram for the test well is provided as **Figure 3-4**. As-built construction details for the test well are provided in **Table 3-1**. The test well was air developed for several hours from within the casing until the produced water was clear and free of sediment.

### 3.3 Lithological Logging

Drill cuttings were collected at 10-foot intervals, or at changes in lithology, and described by an on-site SWS geologist. The descriptions included lithology, color, hardness, and apparent porosity and permeability. Geologist's logs of the sediments encountered during drilling of test-production well TP-1 are provided in **Appendix A** of this report. The classification system of Dunham (1962) was used to classify limestone intervals. Colors of the drill cuttings were described using the Geological Society of America's Rock Color Chart, which is based on the Munsell system.

### 3.4 Water Quality

Groundwater samples were collected every 30 feet during reverse-air drilling of the testproduction (TP-1) and monitor wells (F-1, F-2, and F-3). The groundwater samples were field tested for specific conductance (conductivity) using a calibrated conductivity meter. Eight-ounce samples were also retained to conduct laboratory analysis for dissolved chloride concentration and conductivity. Dissolved chloride and conductivity measurements were also made during aquifer performance testing. The reverse-air water quality data for a given depth is not necessarily representative of the formation water quality at that particular depth due to mixing with water produced higher in the borehole. However, changes in the composition of the reverse-air discharge can provide qualitative information regarding formation water quality. Samples were collected from the test-production well near the completion of the aquifer performance test for laboratory analytical testing of primary and secondary drinking water standards and water quality parameters critical to the RO WTP design.



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### 3.5 Geophysical Logging

The borehole geophysical logging program implemented during the construction and testing of test-production well TP-1 was designed to collect information on the hydrogeology and water quality of the penetrated strata and information on borehole geometry and volume that would assist in the setting and cementing of casing strings.

Two suites of geophysical logs were run on the test-production well. The first set of geophysical logs was completed by MV Geophysical Surveys, Inc. on April 9, 2009 and included X-Y caliper, gamma ray, and dual induction. The logs were run in a 12.25-inch diameter mudded pilot hole that was completed to 1,113 feet bls. Copies of the geophysical logs are provided in **Appendix C**.

A second suite of geophysical logs was completed by Schlumberger on June 8, 2009 and included caliper, gamma ray, combinable resonance tool, dipole sonic imager, elemental capture sonde, fullbore micro imager, hostile natural gamma ray sonde, platform express (array induction/ MCFL/ spontaneous potential/ neutron/ density/ and gamma ray), three detector lithodensity, compensated neutron, and flow meter log. The logs were run in a 12.25-inch diameter pilot hole completed to 1,733 feet bls. A SWS field geologist was present during geophysical logging. Copies of the geophysical logs are provided on a DVD in **Appendix C**.

### 3.6 Aquifer Hydraulic Testing

### 3.6.1 Short-Term Aquifer Performance Test

A short-term aquifer performance test (APT) was performed in the 12.25-inch diameter pilot hole of the test-production well at a depth between 1,082 and 1,208 feet bls. The well was allowed to flow at a rate of approximately 150 gpm for approximately 2 hours. A pressure transducer/datalogger was set inside the drill pipe to monitor water level change in response to pumping. The water level was allowed to stabilize and a two hour constant rate APT was initiated. At the end of the pumping phase, a groundwater sample was collected and analyzed for specific conductance and chloride. The recovery of water levels was monitored for one hour after the pump was turned off using the pressure transducer/datalogger.

### 3.6.2 Step-Drawdown Test

A step-drawdown test was performed on test-production well TP-1 to obtain aquifer and well yield information and to aid in selecting appropriate pumping rates for the constant rate aquifer performance test. The well had been back plugged to a depth of 1,490 feet bls and reamed to a nominal diameter of 15 inches prior to the step-drawdown test. The step-drawdown test was performed on June 24, 2009 using a 100-horsepower submersible pump set at 100 feet bls. The test consisted of three steady (1,000, 1,200, and 1,400 gpm) steps lasting 2 hours each. Water levels were recorded with pressure transducer/dataloggers installed in the test-production well. Discharge was measured using a flowmeter calibrated to within 2 percent of full scale. Discharge was piped to the existing mine lake located to the west of the project site.

### 3.6.3 Long-Term Constant Rate Aquifer Performance Test

A long-term constant rate APT was completed to determine the hydraulic coefficients for the Upper Floridan Aquifer. Test-production well TP-1 was pumped at a constant rate of 1,150 gpm using a 100 horsepower submersible pump for 6,954 minutes (4.8 days). The APT was completed between June 30, 2009 and July 5, 2009 under the supervision of SWS personnel. Time and potentiometric surface data from the test-production well TP-1 and monitor wells F-1, F-2, and F-3 were measured and recorded using pressure transducers/dataloggers. Discharge from the APT was piped to the existing mine lake to the west of the project site. Silt density index (SDI) tests were performed on the discharge water along with testing of the sand concentration using a Rossum Sand Sampler. Prior to and after conducting the APT, background potentiometric surface data were recorded for a minimum of 24 hours in order to measure natural fluctuations of the potentiometric surface.

A constant rate APT was also performed using a single test well (T-2) constructed in the unnamed aquifer. The APT was performed using a centrifugal pump at a constant rate of 23 gpm for 1.5 hours. Water levels in the pumping well were recorded using pressure transducers/dataloggers installed above and below the pump intake.

### Section 4 Investigation Results

### 4.1 Geology

The geology and hydrogeology of Miami-Dade County have been described in a number of investigations conducted by the USGS, FGS, SFWMD, and various consultants. A list of key references is provided in Section 6.0 of this report. The geologic descriptions provided below are based on a combination of the above sources and the analysis of drill cuttings collected during drilling. The classification scheme of Dunham (1962) was used to describe the limestones encountered during well construction. The stratigraphic and hydrogeologic terminology used in this report conforms to that recommended by the Southeastern Geological Society Ad Hoc Committee (1986).

A description of the geologic formations, aquifers, and confining beds encountered during the drilling of test-production well TP-1 is provided below in order from youngest to oldest. **Figure 4-1** shows a hydrostratigraphic column of test-production well TP-1.

### <u>Fill</u>

The shallow subsurface (0-3 feet bls) near and surrounding the test-production well TP-1 and monitor wells (F-1, F-2, and F-3) is characterized as fill consisting of loose sand, peat and silt along with demolition debris.

Miami (Oolite) Limestone/Fort Thompson Formation/Key Largo Limestone (Undifferentiated) The thin fill is underlain by Pleistocene-aged strata consisting of interfingered and often discontinuous bodies of shallow-water deposits. Fish and Stewart (1991) assigned these strata to the Miami (Oolite) Limestone, Fort Thompson Formation and Key Largo Limestone. However, these subsurface strata are lumped together in this discussion because they bear no real significance to this investigation. According to Fish and Stewart (1991), the largest component of the very highly permeable units in eastern Miami-Dade County is the Fort Thompson Formation, which is partly or completely replaced with the Miami (Oolite) Limestone or Key Largo Limestone near the coast. The undifferentiated strata extend to an approximate depth of 100 feet bls at the project site.

### Tamiami Formation

The Pliocene Tamiami Formation is characterized by Parker et al (1955) as a creamy white limestone and green-gray clayey and calcareous marl locally hardened to limestone, silty sands, and shelly sands. Fish and Stewart (1991) point out that the upper part of the Tamiami Formation along coastal Miami-Dade County consists of highly permeable limestone, calcareous sandstone, and sand. The highly permeable portions of the Tamiami Formation form the lower part of the Biscayne Aquifer at the project site.

The upper portion (100 to 130 feet bls) of the Tamiami Formation consists of light to medium gray sandy fossiliferous packstone with medium to high moldic macroporosity. The porosity of the Tamiami Formation decreases with depth as the formation grades into a wackestone at 130 to 140 feet bls.



#### Hawthorn Group

The Hawthorn Group is regionally extensive and lies unconformably beneath the Tamiami Formation. The Hawthorn Group is characterized by a predominant greenish color and higher clay content than the Tamiami Formation. It is a lithologically diverse unit that contains varying sequences of limestones, sands, sandstones, marls, clays, and phosphates. The commonly high phosphate concentration of numerous beds within the Hawthorn Group results in these beds having a distinctive high gamma ray log response.

The contact between the Tamiami Formation and the Hawthorn Group occurs at approximately 140 feet bls at the project site and is marked by a lithological transition downward from a light gray wackestone to a light olive gray limestone/calcareous sandstone with very fine sand-sized phosphate grains. All phosphatic clays, silts, sands, and limestones are considered part of the Hawthorn Group in this investigation. The base of the Hawthorn Group is identified at a depth of 1,155 feet bls based on a decrease in visible phosphate and natural gamma ray activity in geophysical logs. The basal portion (1,080 to 1,155 feet bls) of the Hawthorn Group is characterized by phosphatic limestones with medium to high macroporosity. It is not possible to separate the Peace River Formation from the Arcadia Formation, as defined by Scott (1988), which may not be present beneath the site.

#### Suwannee Limestone

A thin interval of very light colored fine-grained packstone with good porosity was identified below the Hawthorn Group at a depth between 1,155 and 1,177 feet bls. The interval exhibited low natural radioactivity measured by gamma ray logs and no visible phosphate. The interval lacks the diagnostic fossils characteristic of the subjacent Avon Park Formation.

There is a lack of consensus among local workers as to the presence or absence of the Suwannee Limestone in southeastern Florida because of marine erosion by the Florida current (Reese and Richardson, 2008). The Suwannee Limestone is regionally overlain by the Ocala Limestone except in the southern part of southeastern Florida, including most of Miami-Dade County (Reese and Richardson, 2008). Based on the documented absence of the Ocala, lack of visible phosphate and natural radioactivity, and lack of diagnostic fossils that would place the interval in the Avon Park Formation, the interval is placed in the early Oligocene-aged Suwannee Limestone.

### Avon Park Formation

The top of the middle Eocene-aged Avon Park Formation is located at an approximate depth of 1,177 feet bls at the project site. The boundary between the Avon Park Formation and overlying limestones is subtle. Limestones of the Avon Park Formation are characterized by the presence of the distinctive cone-shaped dictyoconid foraminifera, which were first encountered in cuttings at a depth of 1,177 feet bls. The centimeter-sized echinoid *Neolaganum dali* was found by Vernon (1951) to be very abundant in the upper 50 feet of the Avon Park Formation in Florida peninsula wells. *Neolaganum dali* is abundant in the cuttings obtained from 1,192 to 1,218 feet bls at the project site.

The Avon Park Formation is a lithologically diverse unit. The bulk of the Avon Park Formation penetrated in the test-production well consists of very pale orange (10YR 8/2) to grayish orange

(10YR 7/4) colored limestones that are classified as grainstones. Dolomitic limestone beds are thin (<5 foot), dark gray, and exhibit medium permeability. The entire thickness of the Avon Park Formation was not penetrated at the project site. The FGS placed the base of the Avon Park Formation at a depth of 2,743 feet bls based on cuttings from a SFWMD test well located approximately 6 miles to the west of the project site (Lukasiewicz, 2003).

### 4.2 Hydrogeology

There are two major aquifer systems underlying the project site from land surface to a depth of approximately 3,500 feet bls; the Surficial Aquifer System and the deeper, artesian Floridan Aquifer System. These two aquifer systems are separated by a confining sequence referred to as the Intermediate Confining Unit. The Intermediate Confining Unit contains aquifers suitable for freshwater or brackish water supply in some areas of Florida (where it is referred to as the Intermediate Aquifer System). The Floridan Aquifer System is underlain by low transmissivity carbonate and evaporite strata. A hydrostratigraphic column of TP-1 is provided as **Figure 4-1**.

### 4.3 Surficial Aquifer System

The Surficial Aquifer System in Florida is defined as the "permeable hydrogeologic unit contiguous with land surface that is comprised principally of unconsolidated clastic deposits" (Southeastern Geological Society Ad Hoc Committee, 1986). The Surficial Aquifer System consists predominantly of Pleistocene to late Pliocene-aged sands, sandstones, and fossiliferous limestones that were mostly deposited in shallow-water depositional environments. The base of the Surficial Aquifer System is marked by a significant decrease in the average hydraulic conductivity.

The Surficial Aquifer System in northeastern Miami-Dade County contains one major aquifer, the Biscayne Aquifer. Another aquifer referred to as the Gray Limestone Aquifer is found deeper than the Biscayne Aquifer and is not found at the project site. The Gray Limestone Aquifer thins towards eastern Miami-Dade County and pinches out to west of the project site.

### 4.3.1 Biscayne Aquifer

The Biscayne Aquifer was defined by Parker (1951) as the hydrologic unit of water-bearing rock that carries unconfined groundwater in southeastern Florida. Parker et al., (1955) later amended the definition of the Biscayne aquifer to specifically consist of water-bearing rock of Pleistocene to later Miocene age that includes all or parts of the following formations: Tamiami Formation (uppermost part only), Caloosahatchee Marl, Fort Thompson Formation, Anastasia Formation, Key Largo Limestone and Pamlico Sand. The "Biscayne Aquifer", as originally defined is synonymous with "Surficial Aquifer System". Fish and Stewart (1991) restrict the term "Biscayne Aquifer" to only those areas where there is at least 10 feet of section that has a hydraulic conductivity of 1,000 feet per day (ft/d) or more. Modern revisions of ages and formations have limited the Biscayne Aquifer to the Fort Thompson, the Miami (Oolite) Limestone, the Key Largo Limestone, and the Tamiami Limestone ranging in age from Late Pleistocene to Pliocene.

The Biscayne Aquifer has been designated as a sole source aquifer and is the principal potable water source in Miami-Dade and Broward Counties. The Biscayne Aquifer in general, is wedge-shaped, increasing in thickness towards the coast where it is 300 feet or more thick. The Biscayne Aquifer is absent in westernmost Miami-Dade and Broward County (Fish, 1988; Fish and Stewart, 1991). The regional groundwater flow direction in the project site vicinity is towards the east or southeast (Fish and Stewart, 1991).

The Biscayne aquifer is approximately 140 feet thick in the project site area and consists mostly of fossiliferous limestones, which Fish and Stewart (1991) in their regional cross-sections assigned Miami (Oolite) Limestone, Fort Thompson Formation, and locally to the Key Largo Limestone. Fish and Steward (1991) also indicate that along coastal Miami-Dade County, the upper part of the Tamiami Formation is highly permeable and is included in the Biscayne Aquifer. The upper portion (100 to 130 feet bls) of the Tamiami Formation at the project site consists of light to medium gray sandy fossiliferous packstone with medium to high moldic macroporosity and forms the base of the Biscayne Aquifer at the project site.

### 4.4 Intermediate Confining Unit

The Intermediate Confining Unit is defined as including "all rocks that lie between and collectively retard the exchange of water between the overlying Surficial Aquifer System and the underlying Floridan Aquifer System" (Southeastern Geological Society Ad Hoc Committee, 1986). In eastern Miami-Dade County, the boundary between the Surficial Aquifer System and Intermediate Confining Unit essentially coincides with the boundary between the Tamiami Formation and underlying Hawthorn Group (Fish and Stewart, 1991), which is located at approximately 140 ft bls at the project site.

The Intermediate Confining Unit consists of phosphatic clays, silts, marls, and limestones of the Hawthorn Group. The base of the Intermediate Confining Unit is located at an approximate depth of 1,042 to 1,048 feet bls based on lithologic log from the test-production and monitor wells. The difference between the depths of the Intermediate Confining Unit is due to different land surface elevations across the project site. Therefore, the base of the Intermediate Confining Unit is flat across the project site. Additional information on the lithology of the Intermediate Confining Unit near the project site is available from the aquifer storage and recovery (ASR) wells installed near the Hialeah WTP on Okeechobee Road and West 3<sup>rd</sup> Avenue (Merritt, 1997). The Intermediate Confining Unit was reported by Merritt (1997) to consist of the Hawthorn Group and approximately the upper 25 feet of the Suwannee Limestone. Merritt (1997) placed the boundary between the Intermediate Confining Unit and Floridan Aquifer System at approximately 975 ft bls. Lukasiewicz (2003) identified the base of the Intermediate Confining Unit at a depth of 1,135 feet bls at the SFWMD test site six miles to the west.

### 4.4.1 Unnamed Aquifer

An unnamed aquifer was identified within the Intermediate Confining Unit during construction of the test-production well and monitor wells at the project site. This aquifer is technically considered part of the Intermediate Aquifer System. Recognizing the potential value of an alternative water source to blend with brackish groundwater from the Upper Floridan Aquifer, the City of Hialeah granted approval to construct a test well in the aquifer. A test well (T-2) was subsequently installed and tested to determine the hydraulic characteristics and water quality of the aquifer and its potential use as a raw water source.

### 4.4.1.1 Aquifer Description

The top of the unnamed aquifer occurs at a depth of approximately 480 feet bls and is separated from the Biscayne Aquifer by approximately 340 feet of clays, silts, and sands of the Hawthorn Group. The aquifer consists of limestones characterized as a moderately hard sandy wackestone with low to medium apparent moldic macroporosity. The base of the aquifer occurs at a depth of approximately 620 feet bls at the project site. Lukasiewicz (2003) identified this same permeable zone during construction of a SFWMD test well in the Floridan Aquifer approximately 6 miles to the west of the project site and referred to it as the mid-Hawthorn interval. Permeability indicators in the SFWMD test well included the loss of some drilling fluids while drilling, rapid drill bit penetration and relatively pure limestones in the return cuttings. The SFWMD constructed a monitoring interval (DF-3: 516-620 feet bls) in the aquifer as part of a tri-zone monitor well.

### 4.4.1.2 Water Levels

Static water level in test well (T-2) was observed at the land surface on September 3, 2009. The estimated land surface and water level is approximately 7 feet referenced to NGVD-29.

### 4.4.1.3 Water Quality

Water quality samples were collected from test well (T-2) on September 14, 2009 and analyzed for dissolved chloride concentration and specific conductance. Dissolved chloride concentrations ranged between 1,560 and 1,580 milligrams per liter (mg/L) and specific conductance ranged between 5,270 and 5,280 micromhos/cm (umhos/cm).

### 4.4.1.4 Aquifer Hydraulics

Water levels in the unnamed aquifer declined approximately 16.9 feet in response to a constant pumping rate of 23 gpm resulting in a specific capacity of 1.4 gallons per minute per foot (gpm/ft). A transmissivity of 375 ft<sup>2</sup>/day was estimated for the unnamed aquifer based on the specific capacity. The SFWMD tested the same zone in their test well (DF-3) and measured a specific capacity of 1.9 gpm/ft.

### 4.5 Floridan Aquifer System

The Floridan Aquifer System is one of the most productive aquifers in the United States and underlies all of Florida and parts of Georgia and South Carolina for a total area of about 100,000 square miles. The Southeastern Geological Society Ad Hoc Committee of Florida Hydrostratigraphic Unit Definition (1986) defines the Floridan Aquifer System as a thick carbonate sequence which includes all or part of the Eocene to middle Miocene Series and functions regionally as a water-yielding hydraulic unit. The system in northeastern Miami-Dade County consists of the following formations in ascending order: Oldsmar Formation of early Eocene age, Avon Park Formation of middle Eocene age, Suwannee Limestone of Oligocene age, and the basal portion of the Hawthorn Group of late Oligocene to Miocene age. The Ocala Limestone of late Eocene age appears to be absent at the project site. The base of the Floridan Aquifer System is generally placed at the top of the uppermost evaporite (anhydrite) bed in the Cedar Keys Formation, which ranges from about 3,500 to 3,700 ft bls in depth in eastern Miami-Dade County (Miller, 1986).

The Floridan Aquifer System is quite heterogeneous as far as hydraulic conductivity (Bush and Johnston, 1988). Flowmeter log data show that the aquifer consists of a number of zones with very high hydraulic conductivities, which are commonly either solution riddled or fractured, separated by confining or semi-confining intervals of rock with low hydraulic conductivities (Miller, 1986). Confining units within the Floridan Aquifer System in south Florida vary greatly in thickness and vertical continuity. Thin clay beds may provide a much higher degree of confinement than much thicker marly and/or dense limestones. Some dolomitic intervals may provide very effective vertical confinement within the Floridan Aquifer System of South Florida because the matrix permeability of dolomite is often an order of magnitude or more less than that of limestone (Maliva and Walker, 1998, 2000; Maliva et al., 2007). Dolostone beds with common vugs or small cavities may be very effective confining units or may have high permeabilities depending upon the degree to which the vugs or cavities are interconnected. Vertical fractures and solution features are locally present within apparent confining units, which may result in high degrees of connections between aquifers.

The Floridan Aquifer System can be subdivided into three main units based on their relative permeabilities; the Upper Floridan Aquifer, the Middle Confining Unit, and the Lower Floridan Aquifer (Miller, 1986). This discussion is limited to the Upper Floridan Aquifer and the Middle Confining Unit. The Lower Floridan Aquifer was not penetrated during test well construction or tested during this investigation.

### 4.5.1 Upper Floridan Aquifer

### 4.5.1.1 Aquifer Description

The top of the Upper Floridan Aquifer was identified at an approximate depth of 1,042 to 1,048 feet bls based on lithologic logs from the test-production and monitor wells. The range in depth of the top of the Upper Floridan Aquifer is due to different land surface elevations across the project site. The Upper Floridan Aquifer is located within limestones of the basal portion of the Hawthorn Group, Suwannee Limestone and upper Avon Park Formation.

The Upper Floridan Aquifer in southeastern Florida is often interpreted to include only a relatively thin Suwannee Limestone and the upper part of the Avon Park Formation (Reese and Alvarez-Zarikian, 2007). An alternative interpretation described by Reese and Alvarez-Zarikian (2007) is that the Suwannee Limestone is absent in parts of southeastern Florida (Miller, 1986; Reese and Memberg, 2000) or equivalent to the lower part of the basal Hawthorn unit (Reese, 2004), and that the Upper Floridan Aquifer begins in the basal Hawthorn Group. While this investigation agrees that the Upper Floridan Aquifer begins in the basal Hawthorn Group, there

is also evidence that a thin interval (1,155 to 1,177 foot bls) below the Hawthorn Group consists of the Suwannee Limestone.

Reese (2008) points out that commonly, one or two major flow zones (typically <20 feet thick) provide most of the productive capacity and that these zones occur within the upper part of the Upper Floridan Aquifer. Flowmeter logs performed in test-production well TP-1 confirm an apparent flow zone in the upper part of the Upper Floridan Aquifer at a depth of 1,112 to 1,140 feet bls. In addition, Miller (1986) identifies unconformities at the top of the Suwannee Limestone and Avon Park Limestone, which Meyer (1989) points out are associated with zones of dissolution and increased permeability. The identified flowmeter zone in the test-production well is located near the top of the Suwannee Limestone.

The basal boundary of the Upper Floridan Aquifer is difficult to define objectively and appears to be gradational (Reese, 2008). According to Reese (2008), the basal boundary is placed above a thick limestone unit that shows gradual but substantial borehole enlargement on caliper logs that is characteristic of fine-grained, poorly cemented limestone of relatively low permeability. For the purposes of this investigation, the base of the Upper Floridan Aquifer is set at a depth of 1,489 feet bls below which a thin, dense, limestone layer was observed. This limestone layer may minimize the upward flow of poorer quality groundwater. The overdrilled pilot hole section of test-production well TP-1 was subsequently backplugged to a depth of 1,489 feet bls. Thus, test-production well TP-1 fully penetrates the Upper Floridan Aquifer. For comparison, Lukasiewicz (2003) placed the base of the Upper Floridan Aquifer at 1,370 feet bls in the SFWMD test well based on the flow logs, which indicate flow stops entering the borehole below this depth.

### 4.5.1.2 Water Levels

Water levels in the test-production well and monitor wells were measured using a pressure gauge installed on the wellhead and by adding the height of the gauge from land surface. The land surface elevation was estimated to the nearest foot referenced to NGVD-29 using a USGS topographic map. Water levels ranged between approximately 47 and 52 feet NGVD-29 during June and July of 2009. Water levels are higher than land surface and therefore the wells will flow freely at land surface. Fluctuations of water levels in the Upper Floridan Aquifer are minimal. The SFWMD recorded a water level fluctuation of 1.5 feet over a period of two years in the test well located six miles to the west (Lukasiewicz, 2003). The potentiometric surface does vary slightly with tidal fluctuations.

### 4.5.1.3 Water Quality

Water samples were collected from the test-production well during drilling, step drawdown testing, and during the APT. SWS staff analyzed the samples for dissolved chloride concentration using an argentometric titration technique (Standard Methods 4500-Cl<sup>-</sup>B, 1997) and specific conductance using a calibrated conductivity meter. Results of the water quality testing provide a general assessment of salinity conditions within the aquifer.

Water samples were collected every 30 feet during open-circulation reverse air well drilling of the open hole of test-production well TP-1 and monitor wells F-1, F-2, and F-3. A plot of dissolved chloride concentrations in mg/L versus depth is provided as **Figure 4-2**. A plot of specific conductance versus depth is provided as **Figure 4-3**.

Dissolved chlorides remain at a nearly constant concentration of 1,210 to 1,260 mg/L between the depths of 1,115 to 1,300 feet bls in test-production well TP-1. Dissolved chlorides increase gradually to a concentration of 1,700 mg/L at a depth of 1,515 feet bls in test-production well TP-1. Chloride concentrations then increase 250 mg/L in the next 30 foot interval ending at 1,547 feet bls and steadily increase to a concentration of 2,170 mg/L at a maximum drilled depth of 1,733 feet bls. Specific conductance shows a similar increasing trend with depth.

Samples collected during the step-drawdown test of TP-1 showed a dissolved chloride concentration of 1,288 mg/l during the first step completed at 1,000 gpm. Dissolved chloride concentrations then increased and remained nearly constant with a range of 1,620 to 1,648 mg/L for the remaining steps completed at 1,200 and 1,400 gpm. Likewise, water samples collected periodically during the duration of the 4.8-day APT show a nearly constant dissolved chloride concentration ranging between 1,600 and 1,630 mg/L. Conductivity was also measured during the APT and results show a result ranging between 5,850 and 5,990 umhos/cm. The lower chloride concentration measured in the first step is likely due to the introduction of freshwater during drilling of the pilot hole.

Water quality samples were collected from test-production well TP-1 near the end of the APT on July 5, 2009. The samples were analyzed for primary and secondary drinking water quality standards. The results are summarized in **Table 4-1**. In addition, water quality samples were collected from TP-1 on September 3, 2009 and subsequently analyzed to determine the concentration of chemical parameters critical to the RO WTP design. Total dissolved solids (TDS) of the samples obtained on July 5, 2009 were measured by the analytical laboratory at a concentration of 3,416 mg/L. The SFWMD measured a similar TDS concentration of 3,460 mg/L in an Upper Floridan Aquifer well constructed, approximately six miles to the west, with an open hole interval between 1,140 and 1,230 feet bls (Lukasiewicz, 2003). Four silt density index tests were performed between July 2, 2009 and July 5, 2009 during the long-term APT. The results ranged between 1.60 and 1.86. Copies of the laboratory analytical results are provided in **Appendix B**.

### 4.5.1.4 Aquifer Hydraulics

An APT was performed to obtain site-specific data on key aquifer hydraulic parameters; transmissivity (T), leakance (L), and storage coefficient (S; storativity). The APT was accomplished by pumping the test-production well TP-1 while recording changes in water levels in the three observation wells (F-1, F-2, and F-3). Wells F-1, F-2, and F-3 are located approximately 93, 730, and 1,265 feet, respectively, from well TP-1. Water levels were recorded using self-contained pressure transducers and data loggers (Schlumberger MicroDivers<sup>TM</sup>). In Situ LevelTroll<sup>®</sup>500 dataloggers were also installed in each well for back-up data recording purposes. The data from both devices yielded a similar overall trend of water levels during the APT.



Figure 4-2. Reverse Air Drilling Dissolved Chloride Concentrations





Figure 4-3. Reverse Air Drilling Specific Conductance



#### Table 4-1 Laboratory Analytical Data for Test Production Well TP-1

Coliform-Total (E-Coli)         Absent         -           Specific Conductance (Field)(grab)         5560         usits           Temperature (Field)         22.1         Degree C           Turbicity (Field)         1.4         NTU           Oxygen, Dissolved (field)         7.9         mg/L           Carbofuran         U         ug/L           Oxarnyl (Vydate)         U         ug/L           Olygosate         U         ug/L           Diquat         U         ug/L           Total Dissolved Solids (TDS)         3416         mg/L           Fluoride         1.11         mg/L           Nitrate (as N)         U         mg/L           Nitrate (as N)         U         mg/L           Nitrate (as N)         U         mg/L           Sulfate         460         mg/L           Sulfate         0.16         mg/L           Sulfate         3.61         mg/L           Sulfate         3.61         mg/L           Carbonate         0.22         mg/L           Carbonate         0.12         mg/L           Sulfate         3.61         mg/L           Carbonate         0.12         mg/L	Parameter	Result	Units
Specific Conductance (Field) (grab)         5560         US/cm           pH (field)         6.70         units           Temperature (Field)         22.1         Degree C           Turbidity (field)         1.4         NTU           Oxygen, Disolved (field)         7.9         mg/L           Carbofuran         U         ug/L           Carbofuran         U         ug/L           Diquat         U         ug/L           Diquat         U         ug/L           Choride         1.11         mg/L           Fluoride         1.11         mg/L           Nitrate (as N)         U         mg/L           Nitrate (as N)         U         mg/L           Sulfate         460         mg/L           Alkalinity, Total (CaCO3) Endpoint 4.3         122         mg/L           Sulfate         0.16         mg/L           Alkalinity, Total (CaCO3) Endpoint 4.3         122         mg/L           Sulfide         3.61         mg/L           Carbonate         122         mg/L           Sulfide         3.61         mg/L           Carbonate         0.16         mg/L           Ovaride, Total         2.0 <td>Coliform-Total (E-Coli)</td> <td>Absent</td> <td>-</td>	Coliform-Total (E-Coli)	Absent	-
PH (field)         6.70         units           Temperature (Field)         22.1         Degree C           Turbicity (field)         1.4         NTU           Oxagen, Dissolved (field)         7.9         mg/L           Carbofuran         U         ug/L           Okamyl (Vydate)         U         ug/L           Oljouat         U         ug/L           Chloride         14300         mg/L           Fluoride         1.11         mg/L           Nitrate (as N)         U         mg/L           Sulfate         460         mg/L           Garbonate         122         mg/L           Corbonate         0.16         mg/L           Sulfate         460         mg/L           Sulfate         0.42         mg/L           Garbonate         0.22         mg/L           Color/pH (Lab)         U         Pt-Co           Odor (Lab) at 40 Degrees C         8         TON           Cyanide, Total         0.0043         mg/L	Specific Conductance (Field)(grab)	5560	uS/cm
Temperature (Field)         22.1         Degree C           Turbidity (field)         1.4         NTU           Oxygen, Dissolved (field)         7.9         mg/L           Carbofuran         U         ug/L           Oxamyl (Vydate)         U         ug/L           Bigsolved Solids (TDS)         3416         mg/L           Chloride         1430         mg/L           Chloride         1.11         mg/L           Fluoride         1.11         mg/L           Nitrate (as N)         U         mg/L           Nitrate (as N)         U         mg/L           Onho-Phosphate (as P)         0.42         mg/L           Onho-Phosphate (as P)         0.42         mg/L           Sulfate         460         mg/L           Atkalinity, Total (CaCO3) Endpoint 4.3         122         mg/L           Carbonate         0.16         mg/L           Sulfade         3.61         mg/L           Sulfade         3.61         mg/L           Odor (Lab) at 40 Degrees C         8         TON           Cyanide, Total         2.0         mg/L           Organic Carbon, Disolved         1.5         mg/L           Organic	pH (field)	6.70	units
Turbidity (field)         1.4         NTU           Oxygen, Dissolved (field)         7.9         mg/L           Carbofuran         U         ug/L           Oxamy (Vydate)         U         ug/L           Bipposate         U         ug/L           Diquat         U         ug/L           Total Dissolved Solids (TDS)         3416         mg/L           Chloride         1.430         mg/L           Fluoride         1.11         mg/L           Nitrate (as N)         U         mg/L           Nitrate (as N)         U         mg/L           Ortho-Phosphate (as P)         0.42         mg/L           Sulfate         460         mg/L           Carbonate         122         mg/L           Carbonate         122         mg/L           Carbonate         0.16         mg/L           Color/pH (Lab)         U         Pt-Co           Odor (Lab) at 40 Degrees C         8         TON           Organic Carbon, Dissolved         1.5         mg/L           Organic Carbon, Dissolved         1.2         mg/L           Organic Carbon, Total         0.021         mg/L           MBAS Surfactants <t< td=""><td>Temperature (Field)</td><td>22.1</td><td>Degree C</td></t<>	Temperature (Field)	22.1	Degree C
Oxygen, Dissolved (field)         7.9         mg/L           Carbofuran         U         ug/L           Oxamyl (Vydate)         U         ug/L           Bipposate         U         ug/L           Diquat         U         ug/L           Total Dissolved Solids (TDS)         3416         mg/L           Chloride         1430         mg/L           Fluoride         1.11         mg/L           Nitrate (as N)         U         mg/L           Ortho-Phosphate (as P)         0.42         mg/L           Ortho-Phosphate (as P)         0.42         mg/L           Sulfate         460         mg/L           Alkalinity, Total (CaCO3) Endpoint 4.3         122         mg/L           Sulfate         0.16         mg/L           Carbonate         122         mg/L           Carbonate         0.16         mg/L           Sulfide         3.61         mg/L           Sulfide         3.61         mg/L           Color/pH (Lab)         U         Pt-Co           Organic Carbon, Total         0.02         mg/L           Organic Carbon, Total         0.12         mg/L           Organic Carbon, Total         <	Turbidity (field)	1.4	NTU
Carbofuran         U         ug/L           Oxamyl (Vydate)         U         ug/L           Olyphosate         U         ug/L           Diquat         U         ug/L           Total Dissolved Solids (TDS)         3416         mg/L           Fluoride         1430         mg/L           Fluoride         1.11         mg/L           Nitrate (as N)         U         mg/L           Nitrate (as N)         U         mg/L           Ortho-Phosphate (as P)         0.42         mg/L           Sulfate         460         mg/L           Garbonate         0.16         mg/L           Carbonate         0.16         mg/L           Sulfate         3.61         mg/L           Color/pH (Lab)         U         Pt-Co           Oddy         mg/L         Sulfate         3.61           Sulfate         3.0043         mg/L         Sulfate           Color/pH (Lab)         U         Pt-Co         O           Oddy         mg/L         Sulfate         3.61         mg/L           Silica         0.0043         mg/L         Sulfate         3.61         mg/L           Organic Carbon, Total	Oxygen, Dissolved (field)	7.9	mg/L
Dxamyl (Vydate)         U         ug/L           Glyphosate         U         ug/L           Glyphosate         U         ug/L           Total Dissolved Solids (TDS)         3416         mg/L           Chloride         1430         mg/L           Fluoride         1.11         mg/L           Nitrate (as N)         U         mg/L           Nitrate (as N)         U         mg/L           Onto-Phosphate (as P)         0.42         mg/L           Sulfate         460         mg/L           Bicarbonate         0.16         mg/L           Carbonate         0.16         mg/L           Sulfate         460         mg/L           Sulfate         0.42         mg/L           Carbonate         0.16         mg/L           Sulfate         440         mg/L           Sulfate         0.411         mg/L           Sulfate         0.411         mg/L           Sulfate         0.16         mg/L           Sulfate         0.41         mg/L           Sulfate         0.41         mg/L           Sulfate         0.41         mg/L           Color(ptL(tab) <t< td=""><td>Carbofuran</td><td>U</td><td>ug/L</td></t<>	Carbofuran	U	ug/L
Glyphosate         U         ug/L           Diquat         U         ug/L           Diquat         U         ug/L           Chloride         1430         mg/L           Fluoride         1.11         mg/L           Fluoride         1.11         mg/L           Nitrate (as N)         U         mg/L           Nitrate (as N)         U         mg/L           Nitrate (as N)         U         mg/L           Sulfate         460         mg/L           Sulfate         461         mg/L           Carbonate         0.16         mg/L           Color/pH (Lab)         U         Pt-Co           Odor (Lab) at 40 Degrees C         8         TON           Cyanide, Total         0.043         mg/L           Silica         10.8         mg/L           Organic Carbon, Dissolved         1.5         mg/L           Organic Carbon, Dissolved         1.5         mg/L           Organic Carbon, Total         2.0         mg/L           Aluminum         0.021         mg/L           Calcium         94.4         mg/L           Copper         U         mg/L           Iton <td>Oxamyl (Vydate)</td> <td>U</td> <td>ug/L</td>	Oxamyl (Vydate)	U	ug/L
Diquat         U         ug/L           Total Dissolved Solids (TDS)         3416         mg/L           Chloride         1430         mg/L           Fluoride         1.11         mg/L           Nitrate (as N)         U         mg/L           Nitrate (as N)         U         mg/L           Nitrate (as N)         U         mg/L           Ortho-Phosphate (as P)         0.42         mg/L           Sulfate         460         mg/L           Bicarbonate         0.16         mg/L           Carbonate         0.16         mg/L           Sulfate         0.411         mg/L           Carbonate         0.16         mg/L           Color/pH (Lab)         U         Pt-Co           Oddr (Lab) at 40 Degrees C         8         TON           Cyanide, Total         0.0043         mg/L           Gilica         10.8         mg/L           Organic Carbon, Dissolved         1.5         mg/L           Organic Carbon, Dissolved         1.5         mg/L           Aluminum         0.021         mg/L           Colorium         94.4         mg/L           Copper         U         mg/L	Glyphosate	U	ug/L
Total Dissolved Solids (TDS)         3416         mg/L           Chloride         1430         mg/L           Nitrate (as N)         U         mg/L           Nitrate (as N)         U         mg/L           Ortho-Phosphate (as P)         0.42         mg/L           Sulfate         460         mg/L           Alkalinity, Total (CaCO3) Endpoint 4.3         122         mg/L           Bicarbonate         0.16         mg/L           Carbonate         0.16         mg/L           Sulfide         3.61         mg/L           Sulfide         3.61         mg/L           Sulfide         0.043         mg/L           Sulfide         0.0043         mg/L           Color/pH (Lab)         U         Pt-Co           Oddr (Lab) at 40 Degrees C         8         TON           Cyanide, Total         0.0043         mg/L           Silica         0.12         mg/L           Organic Carbon, Dissolved         1.5         mg/L           Organic Carbon, Total         7.92         mg/L           Aluminum         0.021         mg/L           Calcium         94.4         mg/L           Capper         U	Diquat	U	ug/L
Chloride         1430         mg/L           Fluoride         1.11         mg/L           Nitrate (as N)         U         mg/L           Sulfate         460         mg/L           Sulfate         460         mg/L           Bicarbonate         0.16         mg/L           Carbonate         0.16         mg/L           Sulfide         3.61         mg/L           Sulfide         3.61         mg/L           Color/pH (Lab)         U         Pt-Co           Odd         0.0043         mg/L           Organic Carbon, Dissolved         1.5         mg/L           Organic Carbon, Total         0.0043         mg/L           Organic Carbon, Total         0.12         mg/L           Hardness, Total         792         mg/L           Calcium         94.4         mg/L           Copper         U         mg/L           Iron         0.023         mg/L           Manganese         U         mg/L           Stortum	Total Dissolved Solids (TDS)	3416	mg/L
Fluoride         1.11         mg/L           Nitrate (as N)         U         mg/L           Nitrate (as N)         U         mg/L           Nitrate (as N)         U         mg/L           Sulfate         460         mg/L           Sulfate         460         mg/L           Sulfate         460         mg/L           Bicarbonate         122         mg/L           Carbonate         0.16         mg/L           Sulfide         3.61         mg/L           Sulfide         3.61         mg/L           Color/pH (Lab)         U         Pt-Co           Oddr (Lab) at 40 Degrees C         8         TON           Cyanide, Total         0.0043         mg/L           Silica         10.8         mg/L           Organic Carbon, Dissolved         1.5         mg/L           Organic Carbon, Total         2.0         mg/L           Aluminum         0.021         mg/L           Calcium         94.4         mg/L           Copper         U         mg/L           Iton         0.023         mg/L           Magnesium         135         mg/L           Solium	Chloride	1430	mg/L
Nitrate (as N)         U         mg/L           Nitrate (as N)         U         mg/L           Ortho-Phosphate (as P)         0.42         mg/L           Sulfate         460         mg/L           Alkalinity, Total (CaCO3) Endpoint 4.3         122         mg/L           Bicarbonate         122         mg/L           Carbonate         0.41         mg/L           Carbonate         0.41         mg/L           Sulfide         3.61         mg/L           Color/pH (Lab)         U         Pt-Co           Odor (Lab) at 40 Degrees C         8         TON           Cyanide, Total         0.0043         mg/L           Silica         10.8         mg/L           Organic Carbon, Dissolved         1.5         mg/L           Organic Carbon, Total         2.0         mg/L           MBAS Surfactants         0.12         mg/L           Aluminum         0.021         mg/L           Calcium         94.4         mg/L           Copper         U         mg/L           Iron         0.023         mg/L           Magnesium         135         mg/L           Sodium         958         mg/L <td>Fluoride</td> <td>1.11</td> <td>mg/L</td>	Fluoride	1.11	mg/L
Nitrate Nitrate (as N)         U         mg/L           Nitrate (as N)         U         mg/L           Ortho-Phosphate (as P)         0.42         mg/L           Sulfate         460         mg/L           Bicarbonate         122         mg/L           Bicarbonate         0.16         mg/L           Carbonate         0.16         mg/L           Sulfide         3.61         mg/L           Carbonate         0.16         mg/L           Sulfide         3.61         mg/L           Color/pH (Lab)         U         Pt-Co           Odor (Lab) at 40 Degrees C         8         TON           Cyanide, Total         0.0043         mg/L           Silica         10.8         mg/L           Organic Carbon, Dissolved         1.5         mg/L           Organic Carbon, Total         2.0         mg/L           MASS Surfactants         0.12         mg/L           Hardness, Total         792         mg/L           Calcium         94.4         mg/L           Copper         U         mg/L           Iton         0.023         mg/L           Solium         95.5         mg/L	Nitrate (as N)	U	mg/L
Nitrate (as N)         U         mg/L           Chrho-Phosphate (as P)         0.42         mg/L           Sulfate         460         mg/L           Alkalinity, Total (CaCO3) Endpoint 4.3         122         mg/L           Bicarbonate         122         mg/L           Carbonate         0.16         mg/L           Sulfide         3.61         mg/L           Sulfide         3.61         mg/L           Color/pH (Lab)         U         Pt-Co           Odor (Lab) at 40 Degrees C         8         TON           Cyanide, Total         0.0043         mg/L           Silica         10.8         mg/L           Organic Carbon, Dissolved         1.5         mg/L           Organic Carbon, Total         2.0         mg/L           MBAS Surfactants         0.12         mg/L           Hardness, Total         792         mg/L           Copper         U         mg/L           Copper         U         mg/L           Iron         0.023         mg/L           Magnesium         135         mg/L           Sodium         958         mg/L           Sodium         958         mg/L	Nitrate+Nitrate (as N)	U	mg/L
Ortho-Phosphate (as P)         0.42         mg/L           Sulfate         460         mg/L           Alkalinity, Total (CaCO3) Endpoint 4.3         122         mg/L           Bicarbonate         122         mg/L           Carbonate         0.16         mg/L           Carbonate         0.41         mg/L           Carbonate         0.41         mg/L           Color/pH (Lab)         U         Pt-Co           Odor (Lab) at 40 Degrees C         8         TON           Cyanide, Total         0.0043         mg/L           Organic Carbon, Dissolved         1.5         mg/L           Organic Carbon, Total         2.0         mg/L           MBAS Surfactants         0.12         mg/L           Hardness, Total         792         mg/L           Calcium         94.4         mg/L           Copper         U         mg/L           Iron         0.023         mg/L           Nanganesium         135         mg/L           Solium         958         mg/L           Solium         958         mg/L           Solium         0.0002         mg/L           Solium         0.00015         mg/L	Nitrate (as N)	U	mg/L
Sulfate         460         mg/L           Alkalinity, Total (CaCO3) Endpoint 4.3         122         mg/L           Bicarbonate         122         mg/L           Bicarbonate         0.16         mg/L           Nitrogen (Ammonium, NH4+)         0.41         mg/L           Sulfide         3.61         mg/L           Color/pH (Lab)         U         Pt-Co           Oddr (Lab) at 40 Degrees C         8         TON           Cyanide, Total         0.0043         mg/L           Silica         10.8         mg/L           Organic Carbon, Dissolved         1.5         mg/L           Organic Carbon, Total         2.0         mg/L           MBAS Surfactants         0.12         mg/L           Hardness, Total         792         mg/L           Calcium         94.4         mg/L           Copper         U         mg/L           Iron         0.023         mg/L           Naganese         U         mg/L           Solum         958         mg/L           Silver         U         mg/L           Solum         958         mg/L           Silver         0.0015         mg/L <td>Ortho-Phosphate (as P)</td> <td>0.42</td> <td>mg/L</td>	Ortho-Phosphate (as P)	0.42	mg/L
Alkalinity, Total (CaCO3) Endpoint 4.3         122         mg/L           Bicarbonate         0.16         mg/L           Carbonate         0.16         mg/L           Sulfide         3.61         mg/L           Sulfide         3.61         mg/L           Color/pH (Lab)         U         Pt-Co           Odor (Lab) at 40 Degrees C         8         TON           Older (Lab)         0.0043         mg/L           Silica         10.8         mg/L           Organic Carbon, Dissolved         1.5         mg/L           Organic Carbon, Total         2.0         mg/L           MBAS Surfactants         0.12         mg/L           Aluminum         0.021         mg/L           Calcium         94.4         mg/L           Copper         U         mg/L           Iron         0.023         mg/L           Magnesium         135         mg/L           Magnese         U         mg/L           Silver         U         mg/L           Silver         U         mg/L           Silver         0.0015         mg/L           Sodium         958         mg/L           Le	Sulfate	460	mg/L
Bicarbonate         122         mg/L           Carbonate         0.16         mg/L           Nitrogen (Ammonium, NH4+)         0.41         mg/L           Sulfide         3.61         mg/L           Color/pH (Lab)         U         Pt-Co           Odor (Lab) at 40 Degrees C         8         TON           Cyanide, Total         0.0043         mg/L           Silica         10.8         mg/L           Organic Carbon, Dissolved         1.5         mg/L           Organic Carbon, Total         2.0         mg/L           Hardness, Total         792         mg/L           Aluminum         0.021         mg/L           Calcium         94.4         mg/L           Copper         U         mg/L           Iron         0.023         mg/L           Magnesium         135         mg/L           Soliver         U         mg/L           Strontium         958         mg/L           Solium         958         mg/L           Strontium         0.0100         mg/L           Zalasium         0.0015         mg/L           Strontium         0.0100         mg/L	Alkalinity, Total (CaCO3) Endpoint 4.3	122	mg/L
Carbonate         0.16         mg/L           Nitrogen (Ammonium, NH4+)         0.41         mg/L           Sulfide         3.61         mg/L           Color/pH (Lab)         U         Pt-Co           Odor (Lab) at 40 Degrees C         8         TON           Cyanide, Total         0.0043         mg/L           Silica         10.8         mg/L           Organic Carbon, Dissolved         1.5         mg/L           Organic Carbon, Total         2.0         mg/L           MBAS Surfactants         0.12         mg/L           Aluminum         0.021         mg/L           Calcium         94.4         mg/L           Copper         U         mg/L           Iron         0.023         mg/L           Magnesium         135         mg/L           Manganese         U         mg/L           Potassium         59.5         mg/L           Soliver         U         mg/L           Strontium         11.3         mg/L           Solium         958         mg/L           Solium         0.0015         mg/L           Strontium         0.0015         mg/L	Bicarbonate	122	mg/L
Nitrogen (Ammonium, NH4+)         0.41         mg/L           Sulfide         3.61         mg/L           Color/pH (Lab)         U         Pt-Co           Odor (Lab) at 40 Degrees C         8         TON           Cyanide, Total         0.0043         mg/L           Silica         10.8         mg/L           Organic Carbon, Dissolved         1.5         mg/L           Organic Carbon, Total         2.0         mg/L           Hardness, Total         792         mg/L           Aluminum         0.021         mg/L           Calcium         94.4         mg/L           Copper         U         mg/L           Iron         0.023         mg/L           Magnesium         135         mg/L           Manganese         U         mg/L           Sodium         958         mg/L           Sodium         9588         mg/L           Strontium         0.015         mg/L           Zinc         0.0015         mg/L           Arsenic         0.0015         mg/L           Barium         0.0100         mg/L           Chromium         U         mg/L           Steintiu	Carbonate	0.16	mg/L
Sulfide         3.61         mg/L           Color/pH (Lab)         U         Pt-Co           Odor (Lab) at 40 Degrees C         8         TON           Cyanide, Total         0.0043         mg/L           Silica         10.8         mg/L           Organic Carbon, Dissolved         1.5         mg/L           Organic Carbon, Total         2.0         mg/L           MBAS Surfactants         0.12         mg/L           Hardness, Total         792         mg/L           Aluminum         0.021         mg/L           Calcium         94.4         mg/L           Copper         U         mg/L           Iron         0.023         mg/L           Magnesium         135         mg/L           Magnese         U         mg/L           Solium         958         mg/L           Solium         958         mg/L           Strontium         11.3         mg/L           Zinc         0.0015         mg/L           Arsenic         0.0015         mg/L           Earlum         0.0100         mg/L           Chromium         U         mg/L           Earlum         <	Nitrogen (Ammonium, NH4+)	0.41	mg/L
Color/pH (Lab)         U         Pi-Co           Odor (Lab) at 40 Degrees C         8         TON           Cyanide, Total         0.0043         mg/L           Silica         10.8         mg/L           Organic Carbon, Dissolved         1.5         mg/L           Organic Carbon, Total         2.0         mg/L           MBAS Surfactants         0.12         mg/L           Hardness, Total         792         mg/L           Aluminum         0.021         mg/L           Calcium         94.4         mg/L           Copper         U         mg/L           Iron         0.023         mg/L           Magnesium         135         mg/L           Manganese         U         mg/L           Soliver         U         mg/L           Soliver         U         mg/L           Solium         958         mg/L           Strontium         11.3         mg/L           Zinc         0.0015         mg/L           Barium         0.0021         mg/L           Cadmium         U         mg/L           Chromium         0.0021         mg/L           Lead         U	Sulfide	3.61	mg/L
Odor (Lab) at 40 Degrees C         8         TON           Cyanide, Total         0.0043         mg/L           Silica         10.8         mg/L           Organic Carbon, Dissolved         1.5         mg/L           Organic Carbon, Total         2.0         mg/L           MBAS Surfactants         0.12         mg/L           Hardness, Total         792         mg/L           Aluminum         0.021         mg/L           Calcium         94.4         mg/L           Copper         U         mg/L           Iron         0.023         mg/L           Magnesium         135         mg/L           Manganese         U         mg/L           Silver         U         mg/L           Sodium         958         mg/L           Strontium         11.3         mg/L           Zinc         0.0015         mg/L           Arsenic         0.0010         mg/L           Earium         0.0100         mg/L           Strontium         1         mg/L           Earium         0.0021         mg/L           Earium         0.0021         mg/L           Chromium         <	Color/pH (Lab)	U	Pt-Co
Cyanide, Total         0.0043         mg/L           Silica         10.8         mg/L           Organic Carbon, Dissolved         1.5         mg/L           Organic Carbon, Total         2.0         mg/L           MBAS Surfactants         0.12         mg/L           Hardness, Total         792         mg/L           Aluminum         0.021         mg/L           Calcium         94.4         mg/L           Copper         U         mg/L           Iron         0.023         mg/L           Magnesium         135         mg/L           Manganese         U         mg/L           Potassium         59.5         mg/L           Silver         U         mg/L           Sodium         958         mg/L           Strontium         0.002         mg/L           Arsenic         0.0015         mg/L           Barium         0.00100         mg/L           Cadmium         U         mg/L           Chromium         0.0021         mg/L           Lead         U         mg/L           Nickel         U         mg/L           Selenium         U <td< td=""><td>Odor (Lab) at 40 Degrees C</td><td>8</td><td>TON</td></td<>	Odor (Lab) at 40 Degrees C	8	TON
Silica         10.8         mg/L           Organic Carbon, Dissolved         1.5         mg/L           Organic Carbon, Total         2.0         mg/L           MBAS Surfactants         0.12         mg/L           Hardness, Total         792         mg/L           Aluminum         0.021         mg/L           Calcium         94.4         mg/L           Copper         U         mg/L           Iron         0.023         mg/L           Magnesium         135         mg/L           Manganese         U         mg/L           Potassium         59.5         mg/L           Silver         U         mg/L           Sodium         958         mg/L           Strontium         11.3         mg/L           Zinc         0.002         mg/L           Arsenic         0.0015         mg/L           Barium         0.0100         mg/L           Cadmium         U         mg/L           Chromium         0.0021         mg/L           Read         U         mg/L           Nickel         U         mg/L           Selenium         U         mg/L	Cyanide, Total	0.0043	mg/L
Organic Carbon, Dissolved         1.5         mg/L           Organic Carbon, Total         2.0         mg/L           MBAS Surfactants         0.12         mg/L           Hardness, Total         792         mg/L           Aluminum         0.021         mg/L           Calcium         94.4         mg/L           Calcium         94.4         mg/L           Copper         U         mg/L           Iron         0.023         mg/L           Magnesium         135         mg/L           Magnesium         135         mg/L           Potassium         59.5         mg/L           Silver         U         mg/L           Sodium         958         mg/L           Strontium         11.3         mg/L           Zinc         0.0015         mg/L           Arsenic         0.0015         mg/L           Cadmium         U         mg/L           Nickel         U         mg/L           Selenium         U         mg/L           Nickel         U         mg/L           Martin         U         mg/L           Martin         U         mg/L <td>Silica</td> <td>10.8</td> <td>mg/L</td>	Silica	10.8	mg/L
Organic Carbon, Total         2.0         mg/L           MBAS Surfactants         0.12         mg/L           Hardness, Total         792         mg/L           Aluminum         0.021         mg/L           Calcium         94.4         mg/L           Calcium         94.4         mg/L           Copper         U         mg/L           Iron         0.023         mg/L           Magnesium         135         mg/L           Manganese         U         mg/L           Silver         U         mg/L           Sodium         958         mg/L           Sodium         958         mg/L           Strontium         11.3         mg/L           Zinc         0.0015         mg/L           Arsenic         0.0010         mg/L           Cadmium         U         mg/L           Chromium         0.0021         mg/L           Lead         U         mg/L           Nickel         U         mg/L           Selenium         U         mg/L           Antimony         U         mg/L           Beryllium         U         mg/L	Organic Carbon, Dissolved	1.5	mg/L
MBAS Surfactants         0.12         mg/L           Hardness, Total         792         mg/L           Aluminum         0.021         mg/L           Calcium         94.4         mg/L           Copper         U         mg/L           Iron         0.023         mg/L           Magnesium         135         mg/L           Manganese         U         mg/L           Potassium         59.5         mg/L           Silver         U         mg/L           Sodium         958         mg/L           Strontium         11.3         mg/L           Zinc         0.0015         mg/L           Arsenic         0.0015         mg/L           Cadmium         U         mg/L           Chromium         0.0021         mg/L           Lead         U         mg/L           Nickel         U         mg/L           Selenium         U         mg/L           Marganese         U         mg/L	Organic Carbon, Total	2.0	mg/L
Hardness, Total         792         mg/L           Aluminum         0.021         mg/L           Calcium         94.4         mg/L           Copper         U         mg/L           Iron         0.023         mg/L           Magnesium         135         mg/L           Manganese         U         mg/L           Potassium         59.5         mg/L           Silver         U         mg/L           Sodium         958         mg/L           Strontium         11.3         mg/L           Zinc         0.002         mg/L           Arsenic         0.0015         mg/L           Barium         0.0100         mg/L           Cadmium         U         mg/L           Chromium         0.0021         mg/L           Lead         U         mg/L           Nickel         U         mg/L           Selenium         U         mg/L           Mercoury         U         mg/L	MBAS Surfactants	0.12	mg/L
Aluminum         0.021         mg/L           Calcium         94.4         mg/L           Copper         U         mg/L           Iron         0.023         mg/L           Magnesium         135         mg/L           Manganese         U         mg/L           Potassium         59.5         mg/L           Silver         U         mg/L           Sodium         958         mg/L           Strontium         11.3         mg/L           Zinc         0.002         mg/L           Arsenic         0.0015         mg/L           Barium         0.0100         mg/L           Cadmium         U         mg/L           Lead         U         mg/L           Nickel         U         mg/L           Selenium         U         mg/L           Antimony         U         mg/L           Mercury         U         mg/L	Hardness, Total	792	mg/L
Calcium         94.4         mg/L           Copper         U         mg/L           Iron         0.023         mg/L           Magnesium         135         mg/L           Manganese         U         mg/L           Potassium         59.5         mg/L           Silver         U         mg/L           Sodium         958         mg/L           Strontium         11.3         mg/L           Zinc         0.002         mg/L           Arsenic         0.0015         mg/L           Barium         0.0100         mg/L           Cadmium         U         mg/L           Steel U         U         mg/L           Barium         0.0021         mg/L           Cadmium         U         mg/L           Chromium         0.0021         mg/L           Lead         U         mg/L           Nickel         U         mg/L           Selenium         U         mg/L           Antimony         U         mg/L           Beryllium         U         mg/L           Mercury         U         mg/L	Aluminum	0.021	mg/L
CopperUmg/LIron0.023mg/LMagnesium135mg/LManganeseUmg/LPotassium59.5mg/LSilverUmg/LSodium958mg/LStrontium11.3mg/LZinc0.002mg/LArsenic0.0015mg/LBarium0.0100mg/LCadmiumUmg/LLeadUmg/LNickelUmg/LSeleniumUmg/LBerylliumUmg/LHanimonyUmg/LBerylliumUmg/LHanimonyUmg/LHercuryUmg/LHercuryUmg/LHercuryUmg/LHercuryUmg/LHercuryUmg/L	Calcium	94.4	mg/L
Iron0.023mg/LMagnesium135mg/LManganeseUmg/LPotassium59.5mg/LSilverUmg/LSodium958mg/LStrontium11.3mg/LZinc0.002mg/LArsenic0.0015mg/LBarium0.0100mg/LCadmiumUmg/LLeadUmg/LSelenium0.0021mg/LSeleniumUmg/LItedUmg/LMickelUmg/LSeleniumUmg/LMitenUmg/LMercuryUmg/LMercuryUmg/LMercuryUmg/LMercuryUmg/LMercuryUmg/LMercuryUmg/LMercuryUmg/L	Copper	U	mg/L
Magnesium135mg/LManganeseUmg/LPotassium59.5mg/LSilverUmg/LSodium958mg/LStrontium11.3mg/LZinc0.002mg/LArsenic0.0015mg/LBarium0.0100mg/LCadmiumUmg/LLeadUmg/LNickelUmg/LSeleniumUmg/LBariumUmg/LChromium0.0021mg/LLeadUmg/LSeleniumUmg/LSeleniumUmg/LHanimonyUmg/LBerylliumUmg/LMercuryUmg/LMercuryUmg/LMercuryUmg/L	Iron	0.023	mg/L
ManganeseUmg/LPotassium59.5mg/LSilverUmg/LSodium958mg/LStrontium11.3mg/LZinc0.002mg/LArsenic0.0015mg/LBarium0.0100mg/LCadmiumUmg/LLeadUmg/LNickelUmg/LSeleniumUmg/LAntimonyUmg/LBerylliumUmg/L	Magnesium	135	mg/L
Potassium59.5mg/LSilverUmg/LSodium958mg/LStrontium11.3mg/LZinc0.002mg/LArsenic0.0015mg/LBarium0.0100mg/LCadmiumUmg/LCadmium0.0021mg/LLeadUmg/LNickelUmg/LSeleniumUmg/LAntimonyUmg/LBerylliumUmg/LHarmonick AlexandreUmg/LMercuryUmg/LMercuryUmg/LMercuryUmg/LMercuryUmg/L	Manganese	U	mg/L
SilverUmg/LSodium958mg/LStrontium11.3mg/LZinc0.002mg/LArsenic0.0015mg/LBarium0.0100mg/LCadmiumUmg/LCadmium0.0021mg/LLeadUmg/LNickelUmg/LSeleniumUmg/LBerylliumUmg/LMercuryUmg/L	Potassium	59.5	mg/L
Sodium958mg/LStrontium11.3mg/LZinc0.002mg/LArsenic0.0015mg/LBarium0.0100mg/LCadmiumUmg/LCadmium0.0021mg/LLeadUmg/LNickelUmg/LSeleniumUmg/LBerylliumUmg/LMercuryUmg/LMercuryUmg/L	Silver	U	mg/L
Strontium11.3mg/LZinc0.002mg/LArsenic0.0015mg/LBarium0.0100mg/LCadmiumUmg/LCadmium0.0021mg/LLeadUmg/LNickelUmg/LSeleniumUmg/LAntimonyUmg/LBerylliumUmg/LMercuryUmg/L	Sodium	958	mg/L
Zinc0.002mg/LArsenic0.0015mg/LBarium0.0100mg/LCadmiumUmg/LChromium0.0021mg/LLeadUmg/LNickelUmg/LSeleniumUmg/LAntimonyUmg/LBerylliumUmg/LMercuryUmg/L	Strontium	11.3	mg/L
Arsenic0.0015mg/LBarium0.0100mg/LCadmiumUmg/LChromium0.0021mg/LLeadUmg/LNickelUmg/LSeleniumUmg/LAntimonyUmg/LBerylliumUmg/LMercuryUmg/L	Zinc	0.002	mg/L
Barium         0.0100         mg/L           Cadmium         U         mg/L           Chromium         0.0021         mg/L           Lead         U         mg/L           Nickel         U         mg/L           Selenium         U         mg/L           Antimony         U         mg/L           Beryllium         U         mg/L           Mercury         U         mg/L	Arsenic	0.0015	mg/L
Cadmium         U         mg/L           Chromium         0.0021         mg/L           Lead         U         mg/L           Nickel         U         mg/L           Selenium         U         mg/L           Antimony         U         mg/L           Beryllium         U         mg/L           Mercury         U         mg/L	Barium	0.0100	mg/L
Chromium         0.0021         mg/L           Lead         U         mg/L           Nickel         U         mg/L           Selenium         U         mg/L           Antimony         U         mg/L           Beryllium         U         mg/L           Thallium         U         mg/L           Wercury         U         mg/L	Cadmium	U	mg/L
Lead         U         mg/L           Nickel         U         mg/L           Selenium         U         mg/L           Antimony         U         mg/L           Beryllium         U         mg/L           Thallium         U         mg/L           Mercury         U         mg/L	Chromium	0.0021	mg/L
Nickel         U         mg/L           Selenium         U         mg/L           Antimony         U         mg/L           Beryllium         U         mg/L           Thallium         U         mg/L           Mercury         U         mg/L	Lead	U	mg/L
Selenium         U         mg/L           Antimony         U         mg/L           Beryllium         U         mg/L           Thallium         U         mg/L           Mercury         U         mg/L	Nickel	U	mg/L
Antimony     U     mg/L       Beryllium     U     mg/L       Thallium     U     mg/L       Mercury     U     mg/L	Selenium	U	mg/L
Beryllium         U         mg/L           Thallium         U         mg/L           Mercury         U         mg/L	Antimony	U	mg/L
Thallium     U     mg/L       Mercury     U     mg/L	Beryllium	U	mg/L
Mercury U mg/L	Thallium	U	mg/L
	Mercury	U	mg/L
Ultraviolet Absorption Method 0.070 II/cm	Ultraviolet Absorption Method	0.070	l/cm
Gross Alpha 21.8 ± 5.1 pCi/L	Gross Alpha	21.8 ± 5.1	pCi/L
Radium-226 5.2 ± 0.6 pCi/L	Radium-226	$5.2 \pm 0.6$	pCi/L
Radium-228 0.8U ± 0.5 pCi/L	Radium-228	0.8U ± 0.5	pCi/L

Sample collected after pumping TP-1 for 6,954 minutes (4.8 days) at 1,150 gallons per minute U - Analyzed for but not detected
The APT test was conducted for approximately 4.8 days between June 30, 2009 and July 5, 2009. The average pumping rate was 1,150 gpm (1.7 Mgd), which was measured using a totalizing flow meter. The pumping rate was adjusted using a gate valve as needed to maintain a near constant rate. The recovery of water levels was recorded after the termination of pumping for approximately 18 hours.

The time-drawdown data were analyzed using the Hantush-Jacob (1955) and Walton (1960, 1962) modification of the Theis non-equilibrium equation (Hantush-Williams solution), which is a curve matching procedure. The time-drawdown data were also analyzed using the Cooper-Jacob (1946) modification of the Theis non-equilibrium equation (also known as the 'straight-line' method). Copies of the APT analyses for F-1, F-2, and F-3 are provided as **Figure 4-4, 4-5, and 4-6**, respectively. The time-recovery data were analyzed using a method similar to the Cooper-Jacob method with the exception that residual drawdown (s') is plotted against equivalent time (t\*), rather than drawdown versus time (Theis, 1935). Analysis of the APT recovery data for the three monitor wells is provided as **Figure 4-7**.

The results of the APT are summarized in **Table 4-2**. Calculated transmissivity values increased with the distance of the observation well from the pumped well, which is a common pattern for APTs in the Floridan Aquifer System. The overall testing results indicate that the transmissivity of the production zone at the project site is approximately 5,500 to 6,500 ft<sup>2</sup>/d and that the storage coefficient is roughly 5 X  $10^{-4}$ . For comparison, Lukasiewicz (2003) calculated a transmissivity of approximately 7,100 ft<sup>2</sup>/d from an APT performed on a flow zone identified between 1,140 and 1,230 feet bls in a well constructed approximately 6 miles to the west.

The calculated leakance values ranged from  $1.5 \times 10^{-4}$  to  $4.1 \times 10^{-4} d^{-1}$ . The leakance value is critical as it is a measure of the potential for vertical fluid migration into the production zone, such as more saline water from underlying strata. The leakance value reflects the total leakage of water into the production zone from both underlying and overlying strata during the APT. However, inasmuch as the strata that overlies the production zone is significantly less permeable than the underlying strata, the leakance value likely reflects primarily vertical leakage upward into the production zone from below.

A short-term APT was performed in the 12.25-inch diameter pilot hole of the test-production well at a depth between 1,082 and 1,208 feet bls. The APT was performed at 150 gpm for two hours on May 27, 2009. The maximum drawdown of 30.4 feet was recorded during the test resulting in a specific capacity of 5 gpm/ft. A transmissivity of approximately 1,300 ft<sup>2</sup>/day was estimated from the specific capacity.

# 4.5.2 Middle Confining Unit

The top of the Middle Confining Unit is placed at the approximate depth of 1,489 feet bls and is difficult to distinguish from the overlying Upper Floridan Aquifer. The Middle Confining Unit predominantly consists of soft to hard very pale orange to grayish orange limestones with low to medium macroporosity. Dolomitic limestones and dolomite are also located in the confining zone. The Middle Confining Unit consists of the middle and lower parts of the Avon Park Formation and upper part of the Oldsmar Formation. The porosity and permeability of the



Figure 4-4: Analysis of APT Drawdown Data from Monitor Well F-1



Figure 4-5: Analysis of APT Drawdown Data from Monitor Well F-2



Figure 4-6: Analysis of APT Drawdown Data from Monitor Well F-3





Figure 4-7: Analysis of APT Recovery Data

Ν	/lethod/Well	F-1	<b>F-2</b>	<b>F-3</b>	
Pumping Phase					
Hantush-Walton	Transmissivity	$3,700 \text{ ft}^2/\text{d}$	5,460 ft <sup>2</sup> /d	6,170 ft <sup>2</sup> /d	
	Storage coefficient	5.7 X 10 <sup>-5</sup>	4.5 X 10 <sup>-4</sup>	2.9 X 10 <sup>-4</sup>	
	Leakance	$1.6 \text{ X } 10^{-4} \text{ d}^{-1}$	4.1 X 10 <sup>-4</sup> d <sup>-1</sup>	$1.5 \text{ X } 10^{-4} \text{ d}^{-1}$	
Cooper-Jacob	Transmissivity	4,300 ft <sup>2</sup> /d	6,410 ft <sup>2</sup> /d	7,340 ft <sup>2</sup> /d	
	Storage coefficient	9.95 X 10 <sup>-5</sup>	4.3 X 10 <sup>-4</sup>	6.7 X 10 <sup>-4</sup>	
<b>Recovery Phase</b>	-				
Theis	Transmissivity	$3,300 \text{ ft}^2/\text{d}$	5,880 ft <sup>2</sup> /d	6,340 ft <sup>2</sup> /d	

# Table 4-2 Aquifer Performance Test Summary

individual beds of the Middle Confining Unit are variable, but the overall vertical hydraulic conductivity of the unit is low. Reese (2007) indicates that the Middle Confining Unit provides leaky confinement. The geophysical logs completed on test-production well TP-1 indicate that the hydraulic conductivity decreases with depth below the Upper Floridan Aquifer and into the Middle Confining Unit.

The base of the Middle Confining Unit was not penetrated during the construction of testproduction well TP-1. The pilot hole was advanced to a total depth of 1,733 feet bls to record water quality below the proposed open hole section of the future production wells. The test well was backplugged using neat cement to a depth of 1,489 feet bls. According to Lukasiewicz (2003), the base of the Middle Confining Unit and top of the Lower Floridan Aquifer was identified at a depth of 2,510 feet bls at the SFWMD test well site.

# Section 5 Groundwater Model

# 5.1 Introduction

A three-dimensional groundwater flow model with variable density and solute transport was developed to study the long-term changes of water quality at the proposed wellfield constructed in the Upper Floridan Aquifer. The model was run using a withdrawal rate of 13.33 Mgd and 23.33 Mgd. This model was developed based on a calibrated regional model (Golder Associates, 2008), and updated with local hydraulic parameters and water quality data collected during the field study. The model was developed using the USGS SEAWAT code that is capable of simulating flow with variable density.

Local calibration was performed so that the model could mimic the drawdown observed in the APTs completed at the project site. The local model calibration was based on the five-day APT conducted at the site between June 30, 2009 and July 5, 2009. During model calibration, hydraulic parameters were adjusted so that simulated drawdowns matched the drawdowns observed in the on-site monitor wells.

After the model calibration, the model was used to predict the long-term water quality changes of mixed raw water from 14 proposed production wells tapping the Upper Floridan Aquifer at the site. A series of sensitivity analysis model runs were also performed. Dispersivity, effective porosity and vertical hydraulic conductivity were tested during sensitivity analysis.

# 5.2 Code Selection

The target aquifer, the Upper Floridan Aquifer contains mostly brackish water (Myers, 1989; Reese, 1994; Reese and Richardson, 2004). The groundwater movement in the Floridan Aquifer System is impacted by the fluid density. Therefore, a computer code capable of simulating flow with variable density is required.

SEAWAT is a computer program that couples two popular codes, MODFLOW (McDonald and Harbaugh 1988) and MT3DMS (Zheng and Wang 1998) for flow with variable density. SEAWAT, widely used throughout the world (Guo and Bennett, 1998; Guo and Langevin 2002; Langevin et al., 2003) was used to construct the model.

SEAWAT solves two coupled partial differential equations (Guo and Langevin 2002; Langevin et al., 2003). The governing equation for the flow in terms of freshwater head is:

$$\nabla \cdot \rho K_f \left( \nabla h_f + \frac{\left(\rho - \rho_f\right)}{\rho_f} \nabla z \right) = \rho S_f \frac{\partial h_f}{\partial t} + n \frac{\partial \rho}{\partial C} \frac{\partial C}{\partial t} - \rho_s q_s$$

where  $h_f$  is the equivalent freshwater head [L],  $K_f$  the hydraulic conductivity [LT<sup>-1</sup>];  $\rho$  the fluid density [ML<sup>-3</sup>];  $\rho_f$  the freshwater density [ML<sup>-3</sup>];  $S_f$  the storage coefficient in terms of freshwater head;  $\rho_s$  [ML<sup>-3</sup>]  $q_s$  represents the volumetric flow rate per unit volume of aquifer representing source and/or sink terms [T<sup>-1</sup>]; *C* the salt concentration [ML<sup>-3</sup>], and t represents time [T].

The governing equation for solute transport in porous media is:

$$\frac{\partial C}{\partial t} = -\nabla \cdot \left( \vec{v} C \right) + \nabla \cdot \left( D \cdot \nabla C \right) - \frac{q_s}{\theta} C_s + \sum_{k=1}^N R_k$$

where *D* is the hydrodynamic dispersion coefficient tensor  $[L^2T^{-1}]$ ;  $\vec{v}$  the flow velocity  $[LT^{-1}]$ ; *Cs* the source concentration and  $\theta$  the effective porosity.

The fluid density is defined as a linear function of salt concentration (Guo and Langevin, 2002). Under most natural conditions, salt concentration is represented as the concentration of total dissolved solids (TDS):

$$\rho = \rho_f + \frac{\partial \rho}{\partial C} (C - C_o)$$

where *Co* is the salt concentration for freshwater  $[ML^{-3}]$ . Practically *Co* is equal to zero. Typically, the fluid density of seawater is 1,025 kg/m<sup>3</sup> and the density of fresh water is 1,000 kg/m<sup>3</sup>. The change of water viscosity due to salt concentration change is not considered in this study.

#### 5.3 Regional Model ECFAS

There are many groundwater models that cover the study area, but most of these models were developed for the Surficial Aquifer System. For example, a variable density flow model was developed by the USGS to study the freshwater discharge to Biscayne Bay (Langevin, 2003). The SFWMD has developed a number of MODFLOW-based groundwater flow models for this area (SFWMD, 2006).

Recently, several large-scale groundwater flow models with variable density were developed for the Upper Floridan Aquifer System. A regional scale groundwater flow and solute transport model, the East Coast Floridan Aquifer System Model (ECFAS) was recently developed for the SFWMD (Golder Associates, 2008) using the computer code SEAWAT 2000 (Guo and Langevin 2002; Langevin et al., 2003). It is a three-dimensional groundwater flow and solute transport model. In addition, the U.S. Army Corps of Engineers (USACE, 2006) recently finished the first phase of development of a regional scale groundwater flow model using both SEAWAT2000 and WASH123D codes for the feasibility study of aquifer storage and recovery (ASR) wells as one of the key components of the Comprehensive Everglades Restoration Plan (CERP).

The development of the ECFAS model was completed in two phases. Phase I, LEC Floridan Aquifer System model (HydroGeologic, 2006) was the SFWMD's first attempt to develop a density-dependent groundwater flow model for the Floridan Aquifer System. The Phase I model covered the area of the Lower East Coast (LEC). The Phase II modeling effort included both the Lower East (LE) and the Upper East (UE) regions and the modeling tasks were undertaken by Golder Associates, Inc. (Golder Associates, 2008). Both the Phase I and Phase II models used the USGS computer code SEAWAT 2000 (Guo and Langevin, 2002; Langevin et al., 2003) allowing simulation of density-dependent flow so that the movement of freshwater and brackish water within the aquifer as well as the seawater boundary conditions of the Atlantic Ocean and the underlying Boulder Zone could be simulated.

The model domain includes all or part of Okeechobee, Indian River, St. Lucie, Martin, Palm Beach, Broward, and Miami-Dade Counties, Florida (as shown in **Figure 5-1**). The model has 542 rows and 192 columns. A regular grid spacing of 2,400 feet was applied in both the row and the column directions.

The model has 14 layers that were constructed using the hydrogeological unit geometry and properties compiled by the USGS and the SFWMD. The 14-layer model simulates the FAS only. The deeper so-called "Boulder Zone" was simulated as a constant head and constant concentration boundary. Similarly, the overlying Surficial Aquifer System was also simulated as a constant head and constant concentration boundary condition representing the average head and concentration in the Surficial Aquifer (Golder Associates, 2008).

The structure and typical values of some hydraulic parameters in the vicinity of the project site are summarized in **Table 5-1**. The aquifer named SAS stands for the "Surficial Aquifer System", ICU for the "Intermediate Confining Unit", UFA for the "Upper Floridan Aquifer"; MCU1 and MCU2 for the "First and Second Middle Confining Unit", respectively, APPZ stands for "Avon Park Permeable Zone"; LF1 stands for "the Lower Floridan Aquifer", LFCU1 for the "Lower Floridan Confining Unit" and BZ for the "Boulder Zone".

The model includes specified head boundaries along the northern and western sides for all FAS aquifers and general head boundaries at the Atlantic Ocean outcrop. The initial conditions for the model were established using available observed water levels and water quality data from the monitoring wells as well as results from multiple model simulations.

The model calibration included long-term transient model runs (approaching pseudo-steady-state conditions) over a 365 year period, long-term transient runs from 1999 to 2004 and local-scale calibration at six selected aquifer test locations.

# 5.4 Development of the Site-Specific Model

#### 5.4.1 TMR Approach and Local Model Development

The regional groundwater flow and solute transport model, as described in the previous section, covers a large area of southeast Florida. In order to focus on the hydraulic impact analysis of proposed groundwater withdrawals, the zoom approach was utilized so that a small portion of the regional groundwater model was selected as the local model. The proposed wellfield site is located at the center of this local model. This approach is often referred to as the telescope mesh refinement method (TMR) (Ward et al, 1987; Anderson and Woessner, 1992). This approach does not only help to more accurately locate wells, but also significantly saves computational time and data storage. Typically, the local and regional models are connected by setting up either constant heads or constant fluxes along the local model borders whose values are obtained from the regional model at a specific time. For this study, constant head and constant concentration boundary conditions were specified along the parameters of the local model. The values of heads and concentrations were derived from the regional model.





#### Table 5-1 ECFAS Model Structure and Hydraulic Parameters at Proposed Wellfield

Model Layer	Top Elevation	<b>Bottom Elevation</b>	Thickness	Aquifer	Boundary Conditions	H. Hydraulic conductivity	V. Hydraulic conductivity	Specific storativity	Effective Porosity
	(ft, NGVD)	(ft, NGVD)	(ft)			(ft/day)	(ft/day)	(1/ft)	
1	10	-194	204	SAS	constantHead	10	10	0.00125	0.25
2	-194	-1072	878	ICU	variable	0.006	0.0006	9.00E-07	0.35
3	-1072	-1207	135	UFA	variable	90	9	5.25E-07	0.18
4	-1207	-1341	134	UFA	variable	90	9	5.25E-07	0.18
5	-1341	-1494	153	MCU1	variable	0.01	0.002	9.00E-07	0.35
6	-1494	-1647	153	MCU1	variable	0.01	0.002	9.00E-07	0.35
7	-1647	-1721	74	APPZ	variable	450	45	7.50E-07	0.18
8	-1721	-1795	74	APPZ	variable	450	45	7.50E-07	0.18
9	-1795	-2000	205	MCU2	variable	0.3	0.0015	9.00E-07	0.35
10	-2001	-2207	206	MCU2	variable	0.3	0.0015	9.00E-07	0.35
11	-2207	-2412	205	MCU2	variable	0.3	0.0015	9.00E-07	0.35
12	-2412	-2514	102	LF1	variable	300	30	7.50E-07	0.18
13	-2514	-2977	463	LFCU1	variable	0.002	0.0002	9.00E-07	0.35
14	-2977	-3177	200	BZ	const Head	10000	10000	7.50E-07	0.18

The site-specific local model developed for this study is shown in **Figure 5-2a.** In order to reduce the uncertainty associated with boundary conditions, the model borders of the local model were set relatively far away from the site. The local model has 106 rows and 112 columns. Irregular grid spacing was used for the local model, as also shown in **Figure 5-2b**, so that the monitor and production wells used in the APT tests could be accurately located. The grid spacing applied in the local model varies from 75 feet to 2,400 feet.

The 14 model layers from the original regional model were maintained. However, the depths of some layers were adjusted based on the field data obtained for this study.

# 5.4.2 Local Model Modification

During the development of the local model from the regional model, the hydraulic parameters were translated from the regional model and preserved in the local model. After the local model was created, the properties, including the layer top and bottom elevations, and hydraulic parameters, including the hydraulic conductivity and specific storativity, were revised based on the field data collected for this study. In the original model, the bottom of the Upper Floridan Aquifer is located at approximately -1,341 ft NGVD-29 while the field drill data indicates the bottom of Upper Floridan Aquifer is located at an approximately 1,470 ft NGVD-29. The model layer thickness was changed to reflect this.

The local model was updated with the water quality data collected during the field program. A constant value of TDS concentration 3,500 mg/L was applied for the Upper Floridan Aquifer in the vicinity of proposed wellfield based on the field data. The TDS concentration for the confining unit (MCU or model layer 6) is 4,750 mg/L, which was obtained from the regional model and verified by the field data.

# 5.4.3 Local Model Calibration

An APT was conducted at the site between June 30, 2009 and July 5, 2009. The duration of the APT was 6,954 minutes (4.8 days). Test-production well (TP-1) was pumped continuously at a constant rate of 1,150 gpm. Drawdowns at three monitor wells (F-1, F-2 and F-3) were measured during the APT. The location of the test-production well (TP-1) and the three monitor wells are shown in **Figure 3-1**. The maximum drawdown at these three monitor wells are shown in **Table 5-2**. A discussion on the APT was provided in Section 4.5.1.4 of this report. The local model calibration information is presented here.

During the local model calibration, some of the hydraulic parameters including horizontal and vertical hydraulic conductivity and specific storativity values, were adjusted manually in order to minimize the difference between simulated and observed drawdown at the three monitor wells. The adjustment of hydraulic parameters was made within a patch representing the area where the APT data should be valid. It should be noted that a relatively small area, called TP1\_Zone, was created around well TP-1 during the model calibration. The location of the patch and TP1\_Zone are shown in **Figure 5-3**.



Figure 5-2a: Location of the Site-Specific Local Model





Well	Туре	X (ft)	Y(ft)	Distance (ft)	Time (min)	Drawdown (ft)
TP-1	Test Production Well	579605.69	863248.44			
F-1	Monitor Well	578512.78	863248.44	93	6954	35.61
F-2	Monitor Well	578574.67	863978.3	731	6951	11.85
F-3	Monitor Well	579871.38	863253.85	1266	6941	4.82

Table 5-2 Locations and Maximum Drawdown Observed in Monitor Wells



Figure 5.3: Locations of Patch and TP1\_Zone.



**Figures 5-4** through **5-6** display simulated drawdown and observed drawdown at the three monitor wells, respectively. From the results of the local model calibration, the simulated and observed drawdowns are in general agreement, although the model tends to predict smaller drawdown at early times. The difference between simulated drawdown and observed drawdown becomes smaller towards the end of the five-day aquifer test. It should also be noted that the model likely over-predicts the drawdown in a long-term simulation. Therefore, the results would likely be more conservative in terms of both drawdown and water quality change.

**Table 5-3** shows the hydraulic parameters in the vicinity of the project site after model calibration. The horizontal hydraulic conductivity in the vicinity of the test-production well TP-1 and monitor well F-1 has a lower value of 10 ft/day. The test zone is approximately 400 ft thick, so the transmissivity value in the close vicinity of these two wells is about 4,000 ft<sup>2</sup>/day. That is close to the value of 3,701 ft<sup>2</sup>/day derived using graphic methods in aquifer test analysis.

For the area where F-2 and F-3 are located, the horizontal hydraulic conductivity value is 32 ft/day (or 8,840 ft<sup>2</sup>/day for transmissivity) for the upper portion of the UFA (Layers 3 and 4) and 10 ft/day for the lower portion of FAS (Layer 5). This calibrated value is slightly higher than the transmissivity values derived using graphic methods in the aquifer test analysis (5,500 ft<sup>2</sup>/day for F-2 and 6,168 ft<sup>2</sup>/day for F-3, respectively). However, using the transmissivity values derived from APT analysis would generate much higher drawdown at wells F-2 and F-3.

# 5.5 Model Predictions

After the model was calibrated, it was used to evaluate the long-term changes of water quality and drawdown impacts associated with the proposed wellfield withdrawals. The proposed wellfield includes 14 wells (including 12 primary production wells and 2 backup wells) with a total pumping capacity of 23.33 Mgd. In this study, pumpage was distributed among all of the 14 wells. All the proposed production wells are constructed in the Upper Floridan Aquifer, which is located approximately between 1,080 ft to 1,480 feet bls (corresponding to the model layers 3 through 5). Therefore, a total number of 42 well cells were used in the model to represent the proposed 14 production wells. The locations of these 14 proposed production wells are shown in **Figure 5-7**.

All the prediction simulations were run for 30 years. The total proposed pumping rate for the wellfield increases with time as shown below:

Years	Pumping Rates (MGD
0-6	13.33
7-17	20.00
18-30	23.33

The total proposed pumpage was evenly distributed among the 14 proposed wells. These wells were assumed to open to the whole Upper Floridan Aquifer (model layers 3 through 5). The distribution of pumpage in each well, however, depends on the transmissivity ratio of each layer



Figure 5-4: Simulated and Observed Time-Drawdown Curves at F-1







Figure 5-5: Simulated and Observed Time-Drawdown Curves at F-2



Figure 5-6: Simulated and Observed Time-Drawdown Curves at F-3



Layer	Bottom elevation	Kx (Ky) (TP1_Zone)	Kz (TP1_Zone)	Kx (ky) (Patch)	Kz (Patch)	Ss	Effective Porosity	Initial Concentration
	(ft NGVD)	(ft/day)	(ft/day)	(ft/day)	(ft/day)	(1/ft)		(TDS, mg/l)
1	-196	10	10	10	10	0.00125	0.25	350
2	-1080	0.006	0.0006	0.006	0.0006	9.00E-07	0.35	1520
3	-1210	10	2	32	4	3.00E-06	0.1	3500
4	-1300	10	2	32	4	1.00E-07	0.1	3500
5	-1480	10	10	10	10	1.00E-07	0.1	3500
6	-1550	0.01	0.01	0.01	0.01	1.00E-07	0.1	3900
7	-1721	450	45	450	45	7.50E-07	0.18	4600
8	-1795	450	45	450	45	7.50E-07	0.18	4600
9	-2000	0.3	0.0015	0.3	0.0015	9.00E-07	0.35	18410
10	-2207	0.3	0.0015	0.3	0.0015	9.00E-07	0.35	18410
11	-2412	0.3	0.0015	0.3	0.0015	9.00E-07	0.35	18410
12	-2514	300	30	300	30	7.50E-07	0.18	35000
13	-2977	0.002	0.0002	0.002	0.0002	9.00E-07	0.35	35000
13	-3177	10000	10000	10000	10000	7.50E-07	0.18	35000





Figure 5.7: Locations of Proposed Production Wells

to the total transmissivity of the entire pumping zone. **Table 5-4** shows the pumpage distribution of the proposed pumping wells.

An additional prediction simulation was run with a constant pumping rate of 13.33 Mgd for the entire duration of 30 years.

Based on the field data, the initial head in the vicinity of the proposed wellfield was set as 48 ft NGVD and the initial TDS concentration was set as 3,500 mg/L for the UFA.

# 5.5.1 Simulated Drawdown

**Figures 5-8a, 5-8b, and 5-8c** show the model calculated drawdown (ft) in model layers 3, 4 and 5, respectively, after 30 years of simulation, due to proposed withdrawals from the Upper Floridan Aquifer. The maximum drawdown, (107 ft), appears at test-production well TP-1 in model layer 5 after 30 years with variable pumping rates. The drawdown in all three layers are relatively similar. **Figures 5-8d, 5-8e, and 5-8f** show the model calculated drawdown (ft) in model layers 3, 4, and 5, respectively, after 30 years of simulation, due to a constant pumping rate of 13.33 Mgd. The maximum drawdown, (65.02 ft), appears at test-production well TP-1 in model layer 5 after 30 years at a constant pumping rate.

# 5.5.2 Long-Term TDS Concentration

The solute transport model, as part of SEAWAT model, was run in predictive mode with the 14 wells from the proposed brackish water wellfield. The model uses mass fraction of total dissolved solids (TDS) as the primary variable for solute transport simulation and fluid density calculation.

The TDS concentration of mixed water from the proposed production wells is calculated as:

$$C_{ave} = \Sigma C_i Q_i / \Sigma Q_i$$

where  $C_{ave}$  is the TDS concentration of mixed water,  $C_i$  is the simulated TDS concentration of well cell i, and  $Q_i$  is the production rate of well cell i. There are 42 well cells to represent the 14 proposed production wells.

**Figure 5-9a** shows the model predicted average TDS concentration of mixed raw water from the 14 proposed wells with variable pumping rates. The results indicate that the average TDS concentration from the wells will gradually increase with time, from an initial value of 3,500 mg/L to approximately 4,310 mg/L over a 30-year period with variable pumping rates. **Figure 5-9b** shows the model predicted average TDS concentration of 3,987 mg/L resulting from a 13.33 Mgd pumping rate over 30 years. There is no apparent stabilization of the long-term change in TDS concentration. The major reason for the increase of salinity is likely due to the vertical fluxes from the layers below the production zone that have a higher chloride concentration.

Well name	Layer	Row	Column	Q (ft3/day)
TP-1	3	52	54	-72141.561
TP-1	4	52	54	-50322.974
TP-1	5	52	54	-100320.46
PW-10	3	51	77	-105325.59
PW-10	4	51	77	-73050.048
PW-10	5	51	77	-44409.364
PW-14	3	38	47	-104489.75
PW-14	4	38	47	-72887.736
PW-14	5	38	47	-45407.513
PW-11	3	38	77	-105325.59
PW-11	4	38	77	-73050.048
PW-11	5	38	77	-44409.364
PW-13	3	38	65	-104860.43
PW-13	4	38	65	-73014.839
PW-13	5	38	65	-44909.731
PW-12	3	38	73	-104860.43
PW-12	4	38	73	-73014.839
PW-12	5	38	73	-44909.731
PW-6	3	26	76	-106491.12
PW-6	4	26	76	-71822.11
PW-6	5	26	76	-44471.772
PW7	3	29	77	-106076.73
PW7	4	29	77	-72245.494
PW7	5	29	77	-44462.778
PW-2	3	72	62	-101783.04
PW-2	4	72	62	-75717.999
PW-2	5	72	62	-45283.958
PW-3	3	72	72	-101777.29
PW-3	4	72	72	-76316.281
PW-3	5	72	72	-44691.432
PW-4	3	72	76	-101792.03
PW-4	4	72	76	-76900.179
PW-4	5	72	76	-44092.791
PW-9	3	64	77	-103125.39
PW-9	4	64	77	-75443.131
PW-9	5	64	77	-44216.474
PW-8	3	70	77	-101792.03
PW-8	4	70	77	-76900.179
PW-8	5	70	77	-44092.791
PW-5	3	24	76	-106491.12
PW-5	4	24	76	-71822.11
PW-5	5	24	76	-44471.772

#### Table 5-4 Pumping Distribution at 23.33 Mgd



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Figure 5-8a: Simulated drawdown (ft) in Model Layer 3 due to Proposed Pumping up to 23.33 mgd from the UFA after 30 Years





Figure 5-8b: Simulated drawdown (ft) in Model Layer 4 due to Proposed Pumping up to 23.33mgd from the UFA after 30 Years



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Figure 5-8c: Simulated drawdown (ft) in Model Layer 5 due to Proposed Pumping up to 23.33 mgd from the UFA after 30 Years



Figure 5-8d: Simulated drawdown (ft) in Model Layer 3 due to Pumpage of 13.33 MGD from the UFA after 30 Years



Figure 5-8e: Simulated drawdown (ft) in Model Layer 4 due to Pumpage of 13.33 MGD from the UFA after 30 Years



Figure 5-8f: Simulated drawdown (ft) in Model Layer 5 due to Pumpage of 13.33 MGD from the UFA after 30 Years.





Figure 5-9a: Simulated Long-term Changes of TDS Concentration (Mg/L) from Proposed Wellfield Pumping up to 23.33 mgd (Baserun)





Figure 5-9b: Simulated Long-term Changes of TDS Concentration (Mg/L) from Proposed Wellfield Pumping 13.33 MGD

#### 5.6 Model Sensitivity Analysis

All numerical model simulations are subject to some degree of uncertainty. Hydraulic and transport parameters and boundary conditions are never known in sufficient detail. The flow and solute transport models were built based on the available hydraulic and salinity data, and calibrated to known conditions. These data were collected from specific locations and in some cases at different times. To populate all of the model cells, the available data were estimated through a kriging process. The distributions of initial chloride concentration, transmissivity, and layer thicknesses obtained from the kriging process may generate distortion in areas where data are not available. Like any other model, assumptions and simplifications of the natural system were made during model development. Therefore, there is some degree of uncertainty involved in the model prediction results.

The purpose of sensitivity analysis is to quantify the uncertainty in the calibrated model caused by uncertainty in the estimates of aquifer parameters, stress, and boundary conditions. During a sensitivity analysis, calibrated values for initial chloride concentration, longitudinal and vertical dispersivities, vertical hydraulic conductivity of confining units underlying the proposed pumping zone, etc. are systematically changed within a plausible range.

A number of model simulations were performed during the sensitivity analysis for this study by changing one parameter value at a time. The parameters tested include the longitudinal dispersivity and vertical dispersivity, effective porosity of the pumping zone, and the vertical hydraulic conductivity values of model layer 6 that is directly underlying the pumping zone. The results of the sensitivity analysis indicate that the predicted TDS concentration is very sensitive to these parameters. The model sensitivity analysis was performed using a variable pumping rates.

Due to the limitation of data availability, time and budget constraints, not all the parameters used in the model were tested for their sensitivity to the model results. The sensitivity of the spatial distribution of initial TDS, horizontal and vertical hydraulic conductivity was not assessed in this study.

# 5.6.1 Longitudinal Dispersivity

Longitudinal dispersivity is one of the key input parameters used in the solute transport model. This parameter is rarely measured in the field, but often estimated through model calibration or simply by literature review. The values of longitudinal, transverse and vertical dispersivity used in the base run were 30 ft, 3 ft and 3 ft, respectively.

**Figure 5-10** shows the model results using different values of longitudinal dispersivity. For the run with longitudinal dispersivity value of 100 ft and transverse and vertical dispersivity of 10 ft, the TDS concentration would reach 5,109 mg/L after 30 years of pumping. For the run with longitudinal dispersivity value of 3 ft and transverse and vertical dispersivity of 0.3 ft, the TDS concentration would reach 3,916 mg/L after 30 years of pumping. The results indicate that the greater the longitudinal dispersivity used the higher the modeled salinity concentrations would be.



Figure 5-10: Sensitivity Analysis 1: Longitudinal Dispersivity



# 5.6.2 Vertical Dispersivity

For the sensitivity analysis of vertical dispersivity ratio, two simulations were conducted to evaluate the uncertainty of this parameter on the model results. In the first run, the ratio of vertical dispersivity to longitudinal dispersivity was reduced from 0.1 in the base-run case to 0.01. In the second simulation, this ratio was increased from 0.1 to 0.5.

**Figure 5-11** shows the results using different values of vertical dispersivity. For the run with longitudinal dispersivity value of 30 ft, transverse dispersivity of 3 ft and vertical dispersivity of 0.3 ft, the TDS concentration would reach 3,917 mg/L after 30 years of pumping. This is expected since a smaller value of vertical dispersivity will likely reduce the vertical movement of solute from the Middle Confining Unit. For the run with longitudinal dispersivity value of 30 ft and transverse and vertical dispersivity of 15 ft, the TDS concentration would reach 5,294 mg/L after 30 years of pumping. This result is also expected since higher vertical dispersivity will likely bring more salt from the Middle Confining Unit below that has a higher TDS concentration.

The results shown in **Figure 5-11** indicate that the greater the vertical dispersivity ratio used, the higher the modeled salinity concentration would be because the major source of salinity increase is due to leakage from deeper layers.

# 5.6.3 Effective Porosity

Effective porosity is another important parameter used in solute transport simulation. It is most commonly considered to represent the porosity of a rock or sediment that is available to actually contribute to groundwater movement through the rock or sediment. The larger the value of effective porosity, the slower the advective solute transport, because flux is reversely proportional to the value of effective porosity as shown in the equation below:

$$v = \frac{q}{n_a}$$

where v is the groundwater flow velocity or pore water velocity (L/T); q is the Darcy velocity (L/T) simulated by the MODFLOW portion of SEAWAT and  $n_e$  is the effective porosity (Batu, 2005).

Similar to dispersivity values, effective porosity is rarely measured in the field but often estimated from literature review or model calibration. Two simulations were conducted to evaluate the uncertainty of this parameter on the model results. In the base run, a value of effective porosity 0.1 was used. This value is relatively low, therefore higher values of effective porosity were used in the sensitivity analysis.

In the first run, the effective porosity was increased from 0.1 in the base-run to 0.15. In the second simulation, this ratio increased from 0.1 to 0.25. **Figure 5-12** shows the results using different values of effective porosity. The results indicate that the higher effective porosity value was used, the lower the simulated TDS concentration. In the run with an effective porosity of 0.15, the simulated TDS concentration is 4,148 mg/L after 30 years. In the run with an effective porosity of 0.25, the simulated TDS concentration is 3,925 mg/L. These results would be




Figure 5-11: Sensitivity Analysis 2: Vertical Dispersivity





Figure 5-12: Sensitivity Analysis 3: Effective Porosity

expected since the higher the effective porosity, the slower the groundwater seepage velocity and thus a reduction in the advective solute transport.

#### 5.6.4 Vertical Hydraulic Conductivity

Model layer 6 represents a semi-confining unit (Middle Confining Unit) that directly underlies the proposed production zone in the model. The vertical hydraulic conductivity of this confining unit does not only have a strong impact on the long-term water quality changes by limiting upward migration of the water with higher TDS concentrations from deeper layers, but also has a direct impact on the drawdown corresponding to the proposed maximum pumpage rate of 23.33 Mgd from the Upper Floridan Aquifer.

The vertical hydraulic conductivity was 0.01 ft for the Middle Confining Unit. To assess the sensitivity of vertical hydraulic conductivity to the simulation results, different values were used in the sensitivity analysis. In one run, the value was increased from 0.01 ft/day used in the base run to 0.05 ft/day; in another run, the value was reduced from 0.01 used in the base run to 0.005 ft/day.

The simulation results, together with the result from the base run, are shown in **Figure 5-13**. As expected, the higher the vertical hydraulic conductivity for the Middle Confining Unit, the higher predicted TDS concentration, because the vertical hydraulic conductivity controls the vertical upward movement of groundwater in response to the proposed pumping in the Upper Floridan Aquifer. When the value of vertical hydraulic conductivity of the Middle Confining Unit was reduced from 0.01 ft/day to 0.005 ft/day, the simulated TDS concentration was 3,712 mg/L after 30 years of simulation, while the maximum drawdown increases to 117 ft at well TP-1. It is clear that a lower value of the vertical hydraulic conductivity of the underlying confining unit will increase the drawdown in the production and reduce the upward migration of saltier water from the lower layers to the production zone so the TDS of mixed water will remain low. When the value of vertical hydraulic conductivity to 6,420 mg/L after 30 years of simulation. It is clear that the simulation results are very sensitive to the value of vertical hydraulic conductivity.

**Figures 5-14** and **5-15** show that simulated drawdown, in feet, in the lower portion of the proposed production zone (model layer 5) resulting from the runs using vertical hydraulic conductivity values within model layer 6 of 0.005 ft/day and 0.05 ft/day, respectively. The maximum predicted drawdown for the run with a vertical hydraulic conductivity of 0.005 ft/day is 117 ft and the maximum predicted drawdown for the run with vertical hydraulic conductivity of 0.05 ft/day is 85 ft

#### 5.6.5 Overall Prediction of Long-term Changes of TDS Concentration

**Figure 5-16** is a summary of all of the sensitivity simulation runs. The shaded area shown suggests the possible ranges of simulated TDS concentration based on the sensitivity analysis. As discussed earlier, among the parameters tested, the vertical hydraulic conductivity of the Middle Confining Unit has the most significant influence on the simulation results. **Figure 5-17** shows the possible ranges of simulated chloride concentrations, based on the assumption that





Figure 5-13: Sensitivity Analysis 4: Vertical Hydraulic Conductivity of MCU





Figure 5-14: Simulated Drawdown (ft) in the Proposed Production Zone after 30 Years with Vertical Hydraulic Conductivity in Layer 6 Set at 0.005 ft/d(Run 7C)



Schlumberger WATER SERVICES Figure 5-15: Simulated Drawdown (ft) in the Proposed Production Zone after 30 Years with Vertical Hydraulic Conductivity in Layer 6 Set at 0.05 ft/d (Run 7D)





Figure 5-16:Simulated Ranges of TDS Concentration (Mg/L) vs. Time (Years)





Figure 5-17:Simulated Ranges of Chloride Concentration (Mg/L) vs. Time (Years)

chloride/TDS ratio is 47%, which was determined from the field data collected during field investigation at this site.

It should be noted that the groundwater modeling is an iterative process (Anderson and Woessner, 1992). The prediction results presented here were based on the best available data and understanding of the geology and hydrogeology of the project site. This model should be updated and recalibrated when new data become available. It is recommended that the model prediction be checked against the actual water quality changes approximately two years after the proposed RO plant is in operation.

## Section 6 References

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Appendix A

Geologic Log of Test-Production Well TP-1

# TABLE A-1. Geological Log Well TP-1City of Hialeah Hydrogeologic Testing and Analyses

### Location: SE ¼, NW ¼, Sec. 17, Township 52 South, Range 40 East Miami-Dade County, Florida Lat. 25° 55.451', Long. 80° 22.196'

Depth (ft bls)	Lithology
0 – 3	SAND, quartz, pale yellowish brown (10YR 8/2), fine, some silt, moderately well sorted, abundant debris-Styrofoam, roof tiles, pieces of wood, medium hydraulic conductivity.
3 - 10	LIMESTONE, yellowish gray (5Y 7/1 to 5Y 8/1), wackestone to slightly sandy wackestone (with 10-15% quartz sand), moderately hard, some gastropod molds, medium to low moldic macroporosity; some wackestone/mudstone, hard, with low macroporosity.
10 - 20	LIMESTONE, very pale orange to pale yellowish brown (10YR 7/2 to 10YR 8/2), sandy wackestone to wackestone, variable quartz content – up to 30%, hard to very hard, low moldic and intergranular macroporosity.
20 - 30	LIMESTONE, yellowish gray to light olive gray (5Y 7/1 to 5Y 8/1), fossiliferous wackestone/packstone, very hard, abundant rig chatter, abundant bivalve and gastropod molds, high to medium moldic macroporosity, some intraclasts, becoming slightly sandy and softer with depth.
30 – 39	LIMESTONE, yellowish gray (5Y 8/1), sandy, fossiliferous wackestone, hard, external bivalve molds, low to medium moldic macroporosity, noticeably harder 32 feet bls.
39 - 60	LIMESTONE/CALCAREOUS SANDSTONE, very pale yellowish brown (10YR 7/2), very sandy wackestone, bordering on calcareous sandstone, moderately hard –getting softer with depth, trace bivalve shell fragments, low intergranular macroporosity, drilling rate increasing through interval.
60 – 90	LIMESTONE, very pale yellowish brown (10YR 7/2), very sandy wackestone, as above, except more bivalve shell fragments- 5-10%, barnacles quick rate of penetration, harder limestone from 63 – 66 feet bls.
90 – 97	LIMESTONE, very pale yellowish brown (10YR 7/2), sandy fossiliferous wackestone, hard, less sandy than above, bivalve/gastropod molds, medium moldic macroporosity.

Depth (ft bls)	Lithology
97 – 100	LIMESTONE, very pale yellowish brown (10YR 7/2), sandy wackestone/packstone, moderately hard, 15-20% shell fragments, other fossils include bryozoans, barnacles, high to medium moldic macroporosity.
100 - 110	LIMESTONE, very light gray to light gray (N7 to N8), sandy fossiliferous packstone, moderately hard to soft, abundant aragonitic shell fragments and molds, medium to high moldic macroporosity.
110 – 130	LIMESTONE, medium gray to medium light gray (N5 to N6) and yellowish gray (5Y 8/1), sandy, fossiliferous packstone, hard, fossils include bryozoans, bivalves, medium moldic and intergranular macroporosity.
130 - 140	LIMESTONE, yellowish gray to light gray (5Y 7/1), wackestone/packstone, moderately hard to soft, bivalve fragments, low to medium macroporosity.
140 – 174	LIMESTONE to CALCAREOUS SANDSTONE, light olive gray (5Y 6/1), very sandy wackestone to calcareous sandstone, moderately hard to soft, 10-15% bivalve fragments, trace (<5%) very fine sand-size phosphate grains, low to medium moldic macroporosity.
174 – 181	LIMESTONE to CALCAREOUS SANDSTONE (50%), as above. CLAYEY SILT/CLAYEY, SILTY SAND (50%), light olive gray (5Y 6/1), very fine quartz sand, well sorted, slightly sticky, abundant bivalve fragments, very fine sand-size phosphate grains.
181 – 201	SILT CLAY/CLAYEY SILT, pale olive (10Y 5/2), slightly sticky to stiff, well sorted, less abundant bivalve shell fragments than above, low permeability.
201 – 212	CLAYEY SILT, pale olive (10Y 5/2), sandy –very fine quartz sand, well sorted, abundant silt-size phosphate grains, low permeability.
212 – 225	<ul> <li>CLAYEY, SANDY, PHOSPHORITIC SILT (60-70%), pale olive (10YR 5/2), very sticky, cohesive (sticking to shaker screen), very fine to fine quartz sand, trace shell fragments (getting more numerous with depth), abundant very fine grained phosphate.</li> <li>LIMESTONE (30-40%), interbedded with clayey silt, yellowish gray (5Y 8/1), fossiliferous wackestone, hard, external bivalve molds, medium to low moldic macroporosity.</li> </ul>

Depth (ft bls)	Lithology
225 - 230	SAME AS ABOVE? Lots of cement returns from drilling through cement plug at bottom of casing, drilling mud is extremely thick due to cement – taking a very long time to come up hole.
230 - 240	SILTY CLAY, light olive gray (5Y 5/2), well sorted, some very fine grained quartz sand, few returns, mud extremely thick due to drilling through cement plug, will empty mud tank at kelly down (244 ft bls).
240 - 250	SILTY SAND, light olive gray (5Y 5/2), minor clay, sticky, contains very fine grained quartz sand, moderately well sorted, 5-10% very fine to silt-size phosphate, trace lithified siliceous mudstone, some coarse-sand size grains.
250 - 260	SILTY SAND, as above, except slightly more phosphatic – 10-15%, also trace bivalve fragments and bone fragments.
260 - 280	SILTY SAND, light olive gray (5Y 5/2), minor clay, phosphatic – similar to above, except no shell or fossil fragments; abundant cement returns in sample (from above).
280 - 290	SILTY SAND, light olive gray (5Y 5/2), minor clay, very soft, plastic, 10% silt-size phosphate, mostly very fine grained quartz sand and silt.
290 - 300	SILTY/CLAYEY SAND, light olive gray (5Y 5/2), stiffer than above, but still relatively soft, 10% silt-size phosphate grains, trace of mudstone/siltstone – soft, low macroporosity.
300 - 310	SILTY/CLAYEY SAND, as above, except slightly more phosphatic (10-15%).
310 - 320	CLAYEY SILT, light olive gray (5Y 5/2), sticky – but pliable, cohesive, abundant fine sand, some siliceous mudstone/siltstone – moderately hard to soft, however, majority is unlithified sediments.
320 - 330	SILTY SAND, light olive gray (5Y 5/2), less clay than above, less cohesive, very wet, phosphatic, minor siliceous mudstone/siltstone.
330 - 340	CLAYEY, SILTY SAND, light olive gray (5Y 5/2), very fine quartz sand, phosphatic, 5% silty sandstone- coarse sand-size fragments - same composition as unlithified portion of sample, trace of bone fragments.
340 - 350	CLAYEY, SILTY SAND, light olive gray (5Y 5/2), very fine grained quartz sand, 15-20% silt-size phosphate, well sorted.

Depth (ft bls)	Lithology
350 - 360	SANDY, CLAYEY SILT, light olive gray (5Y 5/2), more silt/clay than above, variable (5-15%) silt-size phosphate grains, <10% siliceous mudstone/siltstone, moderately hard.
360 - 380	CLAYEY, SANDY SILT, light olive gray (5Y 5/2), slightly stiff, cohesive, 5-10% silt-size phosphate, stiffer clay unit from 361-361.5, low permeability.
380 - 390	CLAYEY, SANDY SILT, similar to above, except slightly stiffer and more cohesive than above.
390 - 410	SILTY, SANDY CLAY, grayish olive (10Y 4/2), stiffer than above, more clay and less very fine quartz sand than above, 5% silt-size phosphate, low permeability.
410 - 433	SANDY, SILTY CLAY, grayish olive (10Y 4/2), slightly more phosphatic than above (15% silt-size phosphate), also some more silty, sandy lenses, overall low permeability.
433 – 446	SANDY, SILTY CLAY and LIMESTONE. Sandy, silty clay, as above, interbedded with yellowish gray (5Y 8/1), sandy, fossiliferous wackestone, hard, medium moldic macroporosity.
446 – 460	<ul> <li>LIMESTONE (60-70%), similar to above, except slightly darker in color (pale light olive gray (5Y 7/1).</li> <li>MARL (30-40%), pale grayish olive (10Y 6/2), phosphatic, silty and samdy, trace of white bivalve shell fragments, some poorly lithified mudstone, friable to moderately hard – same composition as marl, low intergranular macroporosity.</li> </ul>
460 – 469	CALCAREOUS SILTSTONE/SILTY LIMESTONE, light olive gray (5Y 7/1), wackestone, phosphatic, moderately hard to soft, some hard – mostly associated with bivalve fragments, small pebble-size phosphate grains, medium moldic macroporosity.
469 – 480	SILTY LIMESTONE/CALCAREOUS SILTSTONE, light olive gray (5Y 7/1) to olive gray (5Y 6/1), silty wackestone, hard to very hard, bivalve shell fragments, low moldic macroporosity.
480 - 490	CALCAREOUS SANDSTONE/SANDY LIMESTONE, light olive gray (5Y 7/1), sandy wackestone, moderately hard to soft, very similar to

Depth (ft bls)	Lithology				
	above, except higher moldic macroporosity, Also some medium dark gray (N4), limestone, very hard, low macroporosity.				
490 - 500	LIMESTONE, light olive gray (5Y 7/1), sandy wackestone, phosphatic, moderately hard to soft, medium moldic macroporosity.				
500 - 520	LIMESTONE, light olive gray (5Y 7/1), sandy wackestone, hard to moderately hard, phosphatic, few very hard layers, abundant bivalve fragments and molds (internal and external), medium moldic macroporosity. Some light olive gray (5Y 6/1) siltstone/sandstone – moderately hard to soft, from above?				
520 - 540	LIMESTONE, light olive gray (5Y 7/1), fossiliferous wackestone/packstone, moderately hard, abundant bivalve molds - internal/external, casts, foraminifera, medium macroporosity.				
540 - 550	LIMESTONE, light olive gray (5Y 7/1), fossiliferous, sandy wackestone, moderately hard (softer than above), phosphatic, bivalve fragments, echinoderm fragments (urchin spine), medium moldic macroporosity.				
550 - 582	LIMESTONE, yellowish gray (5Y 8/1), sandy fossiliferous wackestone, mostly moderately hard, some hard, abundant external bivalve molds, less shell fragments than above, medium moldic macroporosity, trace of sandy marl, yellowish gray (5Y 8/1).				
582 – 593	LIMESTONE, yellowish gray (5Y 8/1), sandy fossiliferous wackestone, moderately hard to soft, rate of penetration has increased from previous interval, 5-10% silty to very fine sand-size phosphate, trace shell fragments, some internal/external bivalve molds, medium moldic and intergranular macroporosity.				
593 - 606	LIMESTONE, as above, however, faster rate of penetration.				
606 – 620	LIMESTONE, as above (582 – 593); Trace of sandy marl, light olive gray (5Y 7/1), soft, more prevalent below 615 ft bls.				
620 - 633	LIMESTONE, yellowish gray to light olive gray (5Y 7/1), silty phosphatic wackestone, moderately hard to soft, low to medium moldic macroporosity; trace marl, as above.				
633 – 640	LIMESTONE (60%), as above. MARL, (40%), yellowish gray to light olive gray (5Y 7/1), sandy, 10-20% very fine sand to silt-size phosphate, slightly sticky.				

Depth (ft bls)	Lithology					
640 – 650	MARL (60%), as above. LIMESTONE (40%), as above.					
650 – 660	LIMESTONE and MARL, interbedded, yellowish gray to light olive gray (5Y 7/1), limestone- wackestone, moderately hard to soft, as above; marl-sandy, phosphatic, more stiff than above.					
660 – 670	LIMESTONE with MARL, yellowish gray to light olive gray (5Y 7/1), interbedded; Limestone – silty wackestone, very fine grained, moderately hard to soft, phosphatic, low to medium moldic macroporosity. Marl – silty, 5% very fine grained phosphate, sticky, cohesive, low permeability.					
670 – 690	MARL with LIMESTONE, interbedded. Marl, yellowish gray to light olive gray (5Y 7/1), stiff to sticky, 5% very fine sand-size phosphate, low permeability. Limestone, yellowish gray (5Y 8/2) and lighter, very fine-grained wackestone, silty, 5-10% very fine sand-size phosphate, moderately hard to soft, low to medium moldic macroporosity.					
690 – 700	MARL with LIMESTONE, interbedded, yellowish gray (5Y 7/2); marl – sticky, 5-10% phosphate; Limestone, as above.					
700 - 720	LIMESTONE with MARL, yellowish gray (5Y 7/2), limestone-wackestone, sandy, <5% shell fragments, medium moldic macroporosity, <5% phosphate grains. Marl, as above.					
720 - 740	LIMESTONE with MARL, similar to above, except more limestone and less marl. Also trace of sandy, fossiliferous packestone with a medium moldic macroporosity.					
740 – 750	LIMESTONE, yellowish gray (5Y 7/2), sandy wackestone/packstone, moderately hard, medium moldic macroporosity; trace marl – from above?					
750 – 760	LIMESTONE, yellowish gray (5Y 7/2), sandy wackestone to packed wackestone, similar to above, except harder—moderately hard to hard, trace of marl – from above?					
760 – 770	LIMESTONE with MARL; Limestone – sandy wackestone to packed wackestone, as above, except slightly softer – moderately hard, also trace of shell fragments; Marl – yellowish gray (5Y 7/2), sticky, 5-10% phosphate grains.					

Depth (ft bls)	Lithology
770 – 785	MARL with minor LIMESTONE, yellowish gray (5Y 7/2), sticky, as above, slow drilling; limestone, as above.
785 – 810	<ul><li>MARL (80%), yellowish gray (5Y 7/2), as above.</li><li>LIMESTONE (20%), sandy wackestone, very fine grained allochems, moderately hard to hard, slightly phosphatic, low to medium moldic macroporosity.</li></ul>
810 - 824	MARL (75-80%), yellowish gray (5Y 7/2), as above, sandy, sticky, phosphatic. LIMESTONE (15-20%), as above.
824 - 837	LIMESTONE with MARL, limestone- yellowish gray (5Y 7/2), sandy, fossiliferous wackestone, slightly phosphatic, medium moldic macroporosity; marl – as above.
837 – 889	LIMESTONE, yellowish gray (5Y 8/1), wackestone, moderately hard to hard, <5% shell fragments, minor bivalve and gastropod molds/casts, medium to low moldic macroporosity, trace of clay/marl (YG 5Y 7/1).
889 – 896	MARL, yellowish gray to light olive gray (5Y 7/1), sticky, low permeability.
896 – 899	LIMESTONE, yellowish gray (5Y 8/1), wackestone, moderately hard to hard, <5% shell fragments, minor bivalve and gastropod molds/casts, medium to low moldic macroporosity, trace of clay/marl (YG 5Y 7/1).
899 – 915	<ul> <li>MARL (&gt;95%), yellowish gray (5Y 7/2), slightly sandy (5%) quartz sand,</li> <li>&lt;5% silt-size phosphate grains, sticky, but getting stiffer and more cohesive with depth.</li> <li>Limestone (&lt;5%), as above.</li> </ul>
915 – 919	LIMESTONE, white to light gray (N8 to N9), fossiliferous wackestone, very hard, medium moldic macroporosity.
919 – 940	MARL/CLAY and LIMESTONE, roughly equal percentages. Limestone, as above. Clay/marl, yellowish gray (5Y 7/2), slightly sandy, trace of bivalve shell fragments, sticky, cohesive.
940 – 950	LIMESTONE (70%), white to light gray (N7 to N9), wackestone, moderately hard to hard, abundant bivalve molds and shell fragments. Clay (30%), marly, as above.

Depth (ft bls)	Lithology
950 – 960	LIMESTONE, as above. Also some yellowish gray (5Y 7/2) fossiliferous wackestone, hard to moderately hard, with a trace of phosphate silt (<3%), low moldic macroporosity. Trace of clay, as above, 15-25% bivalve shell fragments.
960 – 970	CLAY (80-90%), yellowish gray (5Y 7/2) and (5Y 5/2), sticky, 5-10% silt- size phosphate, shell fragments. LIMESTONE (10-20%), as above.
970 – 980	<ul><li>CLAY (80%), yellowish gray to light olive gray (5Y 6/2), 10-15% silt-size phosphate, sticky to stiff.</li><li>Limestone (20%), yellowish gray (5Y 8/1) and light gray (N8), fossiliferous wackestone/packstone, moderately hard, medium moldic macroporosity.</li></ul>
980 – 990	CLAY (90-95%), yellowish gray (5Y 7/2) and light olive gray (5Y 5/2), stiff, very plastic, contains shell fragments, silt-size phosphate. Clay is becoming more yellowish gray with depth (5Y 7/2) and not quite as stiff (still pretty stiff though). LIMESTONE (5-10%), as above.
990 – 1000	CLAY (>95%), light olive gray (5Y 5/2) and yellowish gray (5Y 7/2), as above. LIMESTONE (<5%), as above.
1000 - 1020	CLAY, light olive gray (5Y 6/2), very sticky. Trace of limestone (1-2%), yellowish gray (5Y 7/2), slightly sandy fossiliferous wackestone, medium moldic macroporosity.
1020 - 1044	CLAY, as above, but <1% limestone (>99% clay).
1044 – 1050	LIMESTONE with SHELL BED (?), white (N9) to yellowish gray (5Y 8/1), fossiliferous wackestone to sandy molluscan wackestone. Very hard to hard, yellowish gray limestone is shelly, medium moldic macroporosity; interval contains 30 – 40% loose shell fragments.
1050 - 1060	LIMESTONE with SHELL, as above.
1060 – 1070	CLAY (60%), yellowish gray (5Y 7/2 to 5Y 6/2), stiff, locking up bit teeth. LIMESTONE (40%), white to light gray (N9 to N8), fossiliferous wackestone, moderately hard, medium moldic macroporosity.

1070 – 1080 CLAY (75%), pale greenish gray to pale olive (10Y 7/2), stiff, abundant shell fragments. LIMESTONE (25%), as above.

Depth (ft bls)

#### Lithology

- 1080 1085 LIMESTONE (60%), as above. CLAY (40%), as above.
- 1085 1095 LIMESTONE, light gray (N7 and yellowish gray (5Y 8/1), moderately hard, medium to high macroporosity, 5% fine phosphate grains.
- 1095 1100 LIMESTONE, very pale orange (10YR 8/2), soft, medium macroporosity, 5% phosphate fine grains.
- 1100 1105 LIMESTONE, yellowish gray (5Y 8/1), hard, high macroporosity.
- 1105 1110 LIMESTONE, very pale orange, soft, medium macroporosity.
- 1110 1115 LIMESTONE, pale yellowish brown (10YR 6/2), hard, high macroporosity.
- 1115 1122 LIMESTONE, very pale orange to pale yellowish brown (10YR 7/2), hard at top, soft in middle, moderately hard at bottom, high macroporosity, gastropod molds.
- 1122 1130 LIMESTONE, yellowish gray and light gray at top (5Y 8/1), grayish orange in middle (10YR 7/4), very pale orange at bottom (10YR 8/2), moderately hard to hard, high macroporosity, 3% phosphate fine grains.
- 1130 1143 LIMESTONE, dark greenish gray to 1135 (5GY 4/1), pale yellowish brown (10YR 6/2) to 1143, dolomitic.
- 1143 1148 LIMESTONE, matrix is medium light gray (N6) and grains are yellowish gray (5Y 8/10), hard, medium macroporosity, gastropod molds.
- 1148 1176 LIMESTONE, very pale orange (10YR 8/2), moderately hard, medium vuggy macroporosity, gastropod molds, crystalline bryzoan stems.
- 1176 1177 LIMESTONE, very pale orange (10YR 8/2), moderately hard, medium vuggy macroporosity, gastropod molds, crystalline bryzoan stems, abundant echinoids.
- 1177 1192 LIMESTONE, very pale orange (10YR 8/2), moderately hard, medium macroporosity, echinoids, foraminifera, bryzoan stems.
- 1192 1208 LIMESTONE, matrix is light gray (N7) and grains are yellowish gray (5Y8/1), moderately hard, medium macroporosity, echinoids and bryzoan stems.

#### Depth (ft bls)

#### Lithology

- 1208 1259 LIMESTONE, yellowish gray (5y 8/1), moderately hard to hard, medium to high macroporosity, echinoids, foraminifera.
- 1259 1270 LIMESTONE yellowish gray (5Y 8/1), soft to moderately hard, medium macroporosity, echinoids.
- 1270 1290 LIMESTONE (50%), grayish orange (10YR 7/4), soft, medium macroporosity, echinoids, foraminifera.
   LIMESTONE (50%), very pale orange (10YR 8/2), soft, medium macroporosity, foraminifera.
- 1290 1300 LIMESTONE (60%), grayish orange, as above. LIMESTONE (40%), very pale orange, as above.
- 1300 1306 LIMESTONE, yellowish gray (5Y 8/1) and grayish orange (10YR 7/4), moderately hard, hard lense at 1306, medium macroporosity, low macroporosity at lense.
- 1306 1331 LIMESTONE, grayish orange (10YR 7/4), soft to moderately hard, medium macroporosity, echinoids, foraminifera, coral.
- 1331 1336 LIMESTONE, grayish orange (10YR 7/4), soft, medium macroporosity, echinoids, foraminifera.
- 1336 1337 LIMESTONE, grayish orange (10YR 7/4) and yellowish gray (5Y 8/1), moderately hard to hard, medium macroporosity, foraminifera.
- 1337 1362 LIMESTONE, grayish orange (10YR 7/4), soft, medium macroporosity, echinoids, foraminifera.
- 1362 1377 LIMESTONE, grayish orange (10YR 7/4), soft, medium macroporosity.
- 1377 1382 LIMESTONE, moderate orange pink (5YR 8/4), hard, high moldic macroporosity.
- 1382 1393 LIMESTONE, grayish orange (10YR 7/4) and moderate orange pink (5YR 8/4), hard, high moldic macroporosity.
- 1393 1422 LIMESTONE, grayish orange (10YR 7/4), soft with hard lense at 1403, medium macroporosity, echinoids, mollusks molds.

Depth (ft bls)	Lithology
1422 – 1453 I	LIMESTONE, grayish orange (10YR 7/4), soft with hard lense at 1439, medium macroporosity, echinoids, mollusks molds.
1453 – 1482 I	LIMESTONE, grayish orange (10YR 7/4), soft and friable, medium macroporosity, foraminifera, coral.
1482 – 1484 I	LIMESTONE, very pale orange (10YR 8/2), moderately hard to hard, medium macroporosity.
1484 – 1485 I	LIMESTONE, very pale orange (10YR 8/2), moderately hard, medium macroporosity.
1485 – 1486 I	LIMESTONE, grayish orange (10YR 7/4), soft, medium macroporosity.
1486 – 1510 I	LIMESTONE, very pale orange (10Y 8/2), hard, high vuggy macroporosity, appears to be a substantial water source.
1510 – 1515 I	LIMESTONE, grayish orange (10YR 7/4), soft, medium macroporosity.
1515 – 1547	LIMESTONE, very pale orange (10YR 8/2) and grayish orange (10YR 7/4), soft, moderately high macroporosity, foraminifera.
1547 – 1562 I	LIMESTONE, grayish orange (10YR7/4), soft, medium macroporosity, foraminifera.
1562 – 1578 I	LIMESTONE, very pale orange (10YR 8/2), soft and hard, medium macroporosity.
1578 – 1603 I	LIMESTONE, very pale orange (10YR 8/2) and grayish orange (10YR 7/4), soft with hard lense at 1592, medium macroporosity.
1603 – 1608 I	LIMESTONE, very pale orange (10YR 8/2) and medium light gray (N6), moderately hard to hard, medium macroporosity.
1608 – 1626 I	LIMESTONE, grayish orange (10YR 7/4), soft to moderately hard, medium macroporosity.
1626 – 1635 I	LIMESTONE, very pale orange (10YR 8/2) and grayish orange (10YR 7/4), hard, low macroporosity.

1635 – 1640 Dolomitic LIMESTONE, dark gray (N3) and grayish orange (10YR 7/4), hard, medium macroporosity.

Depth (ft bls)	Lithology
1640 - 1652	2 LIMESTONE, grayish orange (10YR 7/4), soft, medium macroporosity.
1652 – 1657	7 LIMESTONE, gray black (N2) to med dark gray (N4), moderately hard, medium macroporosity, banded phosphate.
1657 – 1660	5 LIMESTONE, light olive gray (5Y 5/2) and yellowish gray (5Y 7/2), soft, medium macroporosity.
1666 – 167	1 LIMESTONE, very pale orange (10YR 8/2) and dusky yellowish gray (5Y 6/4), hard, low to medium macroporosity.
1671 – 1673	3 LIMESTONE, very pale orange (10YR 8/2) and grayish orange (10YR 7/4), moderately hard, medium macroporosity.
1673 – 1676	6 LIMESTONE, very pale orange (10YR 8/2), yellowish gray (5Y 8/1), and medium light gray (N6), hard, high vuggy macroporosity.
1676 – 1680	DLIMESTONE, grayish orange (10YR 7/4), soft, medium macroporosity.
1680 – 168	1 LIMESTONE, pale yellowish brown (10YR 6/2), hard, low macroporosity.
1681 – 1684	<ul> <li>LIMESTONE with interbedded CLAY</li> <li>LIMESTONE (70%), light brownish gray (5YR 6/1), soft, medium macroporosity</li> <li>CLAY (30%), light brownish gray (5YR 6/1) to yellowish gray (5Y 7/2)</li> </ul>
	stiff, banded.

- 1684 1688 LIMESTONE, grayish orange (10YR 7/4), soft, low to medium macroporosity.
- 1688 1693 Dolomitic LIMESTONE, dark gray (N3) to light gray (N7), hard, medium macroporosity.
- 1693 1700 LIMESTONE, grayish orange (10YR 7/4), soft, moderately high macroporosity.
- 1700 1701 Dolomitic LIMESTONE, dark gray (N3) to light gray (N7), hard, moderately high macroporosity.
- 1701 1710 LIMESTONE, very pale orange (10YR 8/2), pale yellowish brown (10YR 6/2), and medium light gray (N6), soft to hard, medium macroporosity

#### Depth (ft bls)

#### Lithology

- 1710 1711 LIMESTONE, medium gray (N5) and yellowish gray (5Y 8/1), hard, cherty low vuggy macroporosity.
- 1711 1715 LIMESTONE, very pale orange (10YR 8/2) to white (N9), hard, moderately high to high moldic and vuggy macroporosity, high yield water bearing source.
- 1715 1718 LIMESTONE, very pale orange (10YR 8/2), hard, medium macroporosity.
- 1718 1730 LIMESTONE, grayish orange (10YR 7/4) and medium gray (N5), hard, medium macroporosity.
- 1730 1733 LIMESTONE, very pale orange (10YR 8/2) and grayish orange (10YR 7/4), soft to moderately hard, medium macroporosity.
- Note: pilot hole was back filled with neat cement to 1489 ft bls before being reamed with a 15 inch bit to 1490

# Appendix B

Water Quality Laboratory Reports

# Analytical Report 337158

for

**Schlumberger Water Services** 

**Project Manager: Scott Manahan** 

**City of Hialeah** 

17-JUL-09





#### 10200 USA Today Way, Miramar, FL 33025 Ph:(305) 823-8500 Fax:(305) 823-8555

Xenco-Houston (EPA Lab code: TX00122): Texas (T104704215-08-TX), Arizona (AZ0738), Arkansas (08-039-0), Connecticut (PH-0102), Florida (E871002) Illinois (002082), Indiana (C-TX-02), Iowa (392), Kansas (E-10380), Kentucky (45), Louisiana (03054) New Hampshire (297408), New Jersey (TX007), New York (11763), Oklahoma (9218), Pennsylvania (68-03610) Rhode Island (LAO00308), USDA (S-44102)

Xenco-Atlanta (EPA Lab Code: GA00046): Florida (E87428), North Carolina (483), South Carolina (98015), Utah (AALI1), West Virginia (362), Kentucky (85) Louisiana (04176), USDA (P330-07-00105)

Xenco-Miami (EPA Lab code: FL01152): Florida (E86678), Maryland (330)

Xenco-Miramar (EPA Lab code: FL01246): Florida (E86349)

Xenco-Tampa Mobile (EPA Lab code: FL01212): Florida (E84900)

Xenco-Odessa (EPA Lab code: TX00158): Texas (T104704400-08-TX)

Xenco-Dallas (EPA Lab code: TX01468): Texas (T104704295-08-TX)

Xenco-Corpus Christi (EPA Lab code: TX02613): Texas (T104704370-08-TX)

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17-JUL-09



Project Manager: **Scott Manahan Schlumberger Water Services** 1567 Hayler Lane, Suite 202 Fort Myers, FL 33907

Reference: XENCO Report No: **337158 City of Hialeah** Project Address:

#### Scott Manahan:

We are reporting to you the results of the analyses performed on the samples received under the project name referenced above and identified with the XENCO Report Number 337158. All results being reported under this Report Number apply to the samples analyzed and properly identified with a Laboratory ID number. Subcontracted analyses are identified in this report with either the NELAC certification number of the subcontract lab in the analyst ID field, or the complete subcontracted report attached to this report.

Unless otherwise noted in a Case Narrative, all data reported in this Analytical Report are in compliance with NELAC standards. Estimation of data uncertainty for this report is found in the quality control section of this report unless otherwise noted. Should insufficient sample be provided to the laboratory to meet the method and NELAC Matrix Duplicate and Matrix Spike requirements, then the data will be analyzed, evaluated and reported using all other available quality control measures.

The validity and integrity of this report will remain intact as long as it is accompanied by this letter and reproduced in full, unless written approval is granted by XENCO Laboratories. This report will be filed for at least 5 years in our archives after which time it will be destroyed without further notice, unless otherwise arranged with you. The samples received, and described as recorded in Report No. 337158 will be filed for 60 days, and after that time they will be properly disposed without further notice, unless otherwise arranged with you. We reserve the right to return to you any unused samples, extracts or solutions related to them if we consider so necessary (e.g., samples identified as hazardous waste, sample sizes exceeding analytical standard practices, controlled substances under regulated protocols, etc).

We thank you for selecting XENCO Laboratories to serve your analytical needs. If you have any questions concerning this report, please feel free to contact us at any time.

Respectfully,

Tom Helton

Technical Director

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# Sample Cross Reference 337158



# Schlumberger Water Services, Fort Myers, FL

City of Hialeah

Sample Id	Matrix	Date Collected	Sample Depth	Lab Sample Id
TP-1 23 hrs	W	Jul-01-09 19:00		337158-001
TP-1 42 hrs	W	Jul-02-09 14:00		337158-002
TP-1	W	Jul-03-09 13:45		337158-003
TP-1	W	Jul-04-09 11:45		337158-004
TP-1 Z	W	Jul-05-09 14:00		337158-005
TP-1 Z	W	Jul-05-09 14:00		337158-006



### Schlumberger Water Services, Fort Myers, FL

City of Hialeah

Sample Id: TP-1 23 hrsMatrix: WATER% Moisture:Lab Sample Id: 337158-001Date Collected: Jul-01-09 19:00Date Received: Jul-06-09 11:15							
Analytical Method:         Hardness, Total by SM2340B         Prep Method:							
Date Analyzed: Jul-09-09 10:46 Analyst: ARP Seq Number: 764893			:		Tech	ARP	
Parameter	Cas Number	Result	PQL	MDL	Units	Flag	Dil
Hardness (CaCO3)	471-34-1	1010	0.500	0.500	mg/L		1
Analytical Method: Inorganic Anions by EPA 300 Prep Method:							
Date Analyzed: Jul-07-09 13:38 Analyst: NIB Seq Number: 764833		Date Prep	:		Tech	NIB	
Parameter	Cas Number	Result	PQL	MDL	Units	Flag	Dil
Chloride	16887-00-6	2410	10.5	2.58	mg/L		25



### Schlumberger Water Services, Fort Myers, FL

City of Hialeah

Sample Id: <b>TP-1 42 hrs</b> Lab Sample Id: <b>337158-002</b>	Matrix: WATER% Moisture:Date Collected: Jul-02-09 14:00Date Received: Jul-06-09 11:15								
Analytical Method: Hardness, Total by SM2340B					Prep Method:				
Date Analyzed: Jul-09-09 10:46 Se	Analyst: ARP eq Number: 764893	malyst: ARP Date Prep: mber: 764893			Tech: ARP				
Parameter	Cas Number	Result	PQL	MDL	Units	Flag	Dil		
Hardness (CaCO3)	471-34-1	992	0.500	0.500	mg/L		1		
Analytical Method: Inorganic Anions by EPA 300				Prep Method:					
Date Analyzed: Jul-07-09 13:38 Se	Analyst: NIB eq Number: 764833	Date Prep:			Tech: NIB				
Parameter	Cas Number	Result	PQL	MDL	Units	Flag	Dil		
Chloride	16887-00-6	2390	10.5	2.58	mg/L		25		



# Schlumberger Water Services, Fort Myers, FL

City of Hialeah

Sample Id: <b>TP-1</b>	Matrix: WATER			% Moistu	re:				
Lab Sample Id: <b>337158-003</b>	Date Collecte	Date Collected: Jul-03-09 13:45							
	Date Received: Jul-06-09 11:15								
Analytical Method: Hardness, Total by SM2340B					Prep Method:				
Date Analyzed: Jul-09-09 10:46 Se	Analyst: ARP eq Number: 764893	Date Prep	:	Tech: ARP					
Parameter	Cas Number	Result	PQL	MDL	Units	Flag	Dil		
Hardness (CaCO3)	471-34-1	946	0.500	0.500	mg/L		1		
Analytical Method: Inorganic Anions by EPA 300			Prep Method:						
Date Analyzed: Jul-07-09 13:38 Se	Analyst: NIB eq Number: 764833	Date Prep:			Tech: NIB				
Parameter	Cas Number	Result	PQL	MDL	Units	Flag	Dil		
Chloride	16887-00-6	2390	10.5	2.58	mg/L		25		



### Schlumberger Water Services, Fort Myers, FL

City of Hialeah

Sample Id: <b>TP-1</b> Lab Sample Id: <b>337158-004</b>	Matri Date Collecte Date Received	% Moisture:							
Analytical Method: Hardness, Total by SM2340B					Prep Method:				
Date Analyzed: Jul-09-09 10:46 Se	Analyst: ARP eq Number: 764893	umber: 764893 Date Prep:			Tech: ARP				
Parameter	Cas Number	Result	PQL	MDL	Units	Flag	Dil		
Hardness (CaCO3)	471-34-1	884	0.500	0.500	mg/L		1		
Analytical Method: Inorganic Anions by EPA 300				Prep Method:					
Date Analyzed: Jul-07-09 13:38 Se	Analyst: NIB eq Number: 764833	Date Prep:			Tech: NIB				
Parameter	Cas Number	Result	PQL	MDL	Units	Flag	Dil		
Chloride	16887-00-6	2300	10.5	2.58	mg/L		25		



# Schlumberger Water Services, Fort Myers, FL

City of Hialeah

Sample Id: <b>TP-1 Z</b> Lab Sample Id: <b>337158-005</b>	Matrix: WATER % Moisture: Date Collected: Jul-05-09 14:00 Date Received: Jul-06-09 11:15								
Analytical Method: Hardness, Total by SM2340B					Prep Method:				
Date Analyzed: Jul-09-09 10:46 Se	Analyst: ARP eq Number: 764893	alyst: ARP Date Prep: mber: 764893			Tech: ARP				
Parameter	Cas Number	Result	PQL	MDL	Units	Flag	Dil		
Hardness (CaCO3)	471-34-1	980	0.500	0.500	mg/L		1		
Analytical Method: Inorganic Anions by EPA 300				Prep Method:					
Date Analyzed: Jul-07-09 13:38 Se	Analyst: NIB eq Number: 764833	Date Prep:		Tech: NIB					
Parameter	Cas Number	Result	PQL	MDL	Units	Flag	Dil		
Chloride	16887-00-6	2410	10.5	2.58	mg/L		25		


## Schlumberger Water Services, Fort Myers, FL

City of Hialeah

Sample Id: <b>TP-1 Z</b>	Matri	x: WATER		% Moistu	re:		
Lab Sample Id: 337158-006	Date Collecte	d: <b>Jul-05-09</b>	9 14:00				
	Date Received	l: <b>Jul-06-09</b>	0 11:15				
Analytical Method: Alkalinity b	v SM2320B			Prep	Method:		
Date Analyzed: Jul-07-09 11:00	Analyst: OLA	Date Pren		1	Taab		
Se	eq Number: 764635	Date Trep			Tech.	. OLA	
Parameter	Cas Number	Result	PQL	MDL	Units	Flag	Dil
Alkalinity, Total (as CaCO3)		132	8.68	2.17	mg/L		1
Analytical Method: Color by SM	/12120B			Prep	Method:		
Date Analyzed: Jul-06-09 14:55	Date Prep	):		Tech	MSH		
Se	eq Number: 764534						
Parameter	Cas Number	Result	PQL	MDL	Units	Flag	Dil
Color		10	1.0	0.50	CU		1
Analytical Method: DOC by SM			Prep	Method:			
Date Analyzed: Jul-09-09 01:28	Analyst: MAB	Date Prep	):		Tech:	MAB	
Se	eq Number: 764878						
Parameter	Cas Number	Result	PQL	MDL	Units	Flag	Dil
Dissolved Organic Carbon	7440-44-0	2.47	0.500	0.548	mg/L		1
Analytical Method: Hardness, T	<b>Sotal by SM2340B</b>			Prep	Method:		
Date Analyzed: Jul-09-09 10:46	Analyst: ARP	Date Prep	):		Tech	ARP	
Se	eq Number: 764893						
Parameter	Cas Number	Result	PQL	MDL	Units	Flag	Dil
Hardness (CaCO3)	471-34-1	937	0.500	0.500	mg/L		1
Analytical Method: Inorganic A	nions by EPA 300			Prep	Method:		
Date Analyzed: Jul-07-09 13:38	Analyst: NIB	Date Prep	):		Tech:	NIB	
Se	eq Number: 764833						
Parameter	Cas Number	Result	PQL	MDL	Units	Flag	Dil
Fluoride	16984-48-8	0.744	0.740	0.183	mg/L		2
Chloride	16887-00-6	2310	21.0	5.15	mg/L mg/I	TT	50
Nitrate as N	1121-31-9 0 72 707	U	0.800	0.197	mg/L	U	2
Ortho-Phosphate	7723-14-0	U U	0.340	0.152	mg/L	U	2
Sulfate	14808-79-8	260	1.15	0.277	mg/L	U	5



## Schlumberger Water Services, Fort Myers, FL

	City	of Hialeah								
Sample Id: <b>TP-1 Z</b> Lab Sample Id: <b>337158-006</b>	Matriz Date Collected Date Received	x: WATER d: Jul-05-09 l: Jul-06-09	14:00 11:15	% Moisture:						
Angletical Matheds Matalanan I	CD MC L., CW (0204			Duar	Matha di C	W/2010A				
Analytical Method: Metals per 1	CP-MS DY SW 6020A			Prep	Method: S	w 3010A	L			
Date Analyzed: Jul-08-09 03:42 Seq	Analyst: ARP Number: 764731	Date Prep:	Jul-07-09	09:00	Tech	: RWA				
Parameter	Cas Number	Result	PQL	MDL	Units	Flag	Dil			
Barium	7440-39-3	0.017	0.010	0.002	mg/L		1			
Calcium	7440-70-2	128	10.0	2.50	mg/L	D	50			
Potassium	7440-09-7	37.1	0.500	0.170	mg/L	-	1			
Sodium	7440-23-5	1010	25.0	12.5	mg/L	D	50			
Strontium	7440-24-6	12.4	0.005	0.001	mg/L		1			
Analytical Method: Metals, Total	l by SW846 6010B			Prep Method: SW3010A						
Date Analyzed: Jul-08-09 12:39 Seq	Analyst: 4150 Number: 764783	Date Prep:	Jul-07-09	12:32	Tech	: 4150				
Parameter	Cas Number	Result	PQL	MDL	Units	Flag	Dil			
Iron	7439-89-6	U	0.100	0.021	mg/L	U	1			
Manganese	7439-96-5	U	0.050	0.001	mg/L	U	1			
Analytical Method: Nitrogen Am	monia by EPA 350.1			Prep	Method:					
Date Analyzed: Jul-07-09 10:35 Seq	Analyst: SHH Number: 764669	Date Prep:			Tech	: YAD				
Parameter	Cas Number	Result	PQL	MDL	Units	Flag	Dil			
Nitrogen, Ammonia (as N)	7664-41-7	0.172	0.100	0.020	mg/L		1			
Analytical Method: Phosphorus,	Total (Automated) by	EPA 365.4		Prep	Method:					
Date Analyzed: Jul-08-09 11:40 Seq	Analyst: MSH Number: 764826	Date Prep:			Tech	: MSH				
Parameter	Cas Number	Result	POL	MDL	Units	Flag	Dil			
Total Phosphorus (as P)	7723-14-0	U	0.100	0.037	mg/L	U	1			
Analytical Method: Silica by SM	4500-SiO2			Prep	Method:					
Date Analyzed: Jul-09-09 11:58 Seq	Analyst: 9999 Number: 765498	Date Prep:			Tech	: 99999				
Parameter	Cas Number	Result	PQL	MDL	Units	Flag	Dil			
Silica	7631-86-9	10.9	0.660	0.010	mg/L		1			



## Schlumberger Water Services, Fort Myers, FL

	Cit	y of Hialeal	n				
Sample Id: <b>TP-1 Z</b> Lab Sample Id: <b>337158-006</b>	Matri Date Collecte Date Received	x: WATER d: Jul-05-09 d: Jul-06-09	9 14:00 9 11:15	% Moistu	re:		
Analytical Method: Specific Co	onductance by EPA 120.1	_		Prep	Method:		
Date Analyzed: Jul-07-09 09:25	Analyst: MSH Seq Number: 764582	Date Prep	:		: MSH		
Parameter	Cas Number	Result	PQL	MDL	Units	Flag	Dil
Conductivity		5670	50.0	10.0	uS/cm		1
Analytical Method: Sulfide by	SM4500-S-F			Prep	Method:		
Date Analyzed: Jul-07-09 11:30	Analyst: OLA Seq Number: 764640	Date Prep	:		Tech	: OLA	
Parameter	Cas Number	Result	PQL	MDL	Units	Flag	Dil
Sulfide, total	105-05-2	2.40	5.00	1.00	mg/L	Ι	1
Analytical Method: TDS by SM	M2540C			Prep	Method:		
Date Analyzed: Jul-07-09 13:00	e Analyzed: Jul-07-09 13:00 Analyst: YAD Seq Number: 764766				Tech	: YAD	
Parameter	Cas Number	Result	PQL	MDL	Units	Flag	Dil
Total dissolved solids	TDS	4850	5.00	5.00	mg/L		1
Analytical Method: TOC by S	M 5310C			Prep	Method:		
Date Analyzed: Jul-08-09 22:06	Analyst: MAB Seq Number: 764872	Date Prep	:		Tech	: MAB	
Parameter	Cas Number	Result	PQL	MDL	Units	Flag	Dil
Total Organic Carbon	7440-44-0	1.31	0.500	0.210	mg/L		1
Analytical Method: Temperate	ure by EPA 170.1			Prep	Method:		
Date Analyzed: Jul-06-09 14:30	Analyst: MSH Seq Number: 764535	Date Prep	:		Tech	: MSH	
Parameter	Cas Number	Result	PQL	MDL	Units	Flag	Dil
Temperature		25.0	5.00	1.00	Deg C		1
Analytical Method: Turbidity	by EPA 180.1			Prep	Method:		
Date Analyzed: Jul-06-09 14:30	Analyst: MSH Seq Number: 764535	Date Prep	:		Tech	: MSH	
Parameter	Cas Number	Result	PQL	MDL	Units	Flag	Dil
Turbidity		1.21	1.00	0.100	NTU		1



#### Schlumberger Water Services, Fort Myers, FL

City of Hialeah												
Sample Id: <b>TP-1 Z</b> Lab Sample Id: <b>337158-006</b>	Matri Date Collecte Date Received	x: WATER d: Jul-05-09 d: Jul-06-09	% Moistu	re:								
Analytical Method: UV254 by S	SM20 5910B			Prep	Method:							
Date Analyzed: Jul-09-09 14:00 St	Analyst: 9999 eq Number: 765497	9999 Date Prep: 765497			Tech: 9999							
Parameter	Cas Number	Result	PQL	MDL	Units	Flag	Dil					
Absorbance		0.030	0.001	0.001	/cm		1					
Analytical Method: pH by SM4	500-Н			Prep	Method:							
Date Analyzed: Jul-06-09 13:30 St	Analyst: MSH eq Number: 764568	Date Prep	:		Tech	: MSH						
Parameter	Cas Number	Result	PQL	MDL	Units	Flag	Dil					
pH	PH	7.61		1.00	SU		1					





#### FLORIDA Flagging Criteria

- A Value reported is the mean (average) of two or more determinations. This code shall be used if the reported value is the average of results for two or more discrete and separate samples. These samples shall have been processed and analyzed independently. Do not use this code if the data are the result of replicate analysis on the same sample aliquot, extract or digestate.
- **B** Results based upon colony counts outside the acceptable range. This code applies to microbiological tests and specifically to membrane filter colony counts. The code is to be used if the colony count is generated from a plate in which the total number of coliform colonies is outside the method indicated ideal range. This code is not to be used if a 100 mL sample has been filtered and the colony count is less than the lower value of the ideal range.
- **F** When reporting species: F indicates the female sex. Otherwise it indicates RPD value is outside the acceptable range.
- **H** Value based on field kit determination; results may not be accurate. This code shall be used if a field screening test (i.e., field gas chromatograph data, immunoassay, vendor-supplied field kit, etc.) was used to generate the value and the field kit or method has not been recognized by the Department as equivalent to laboratory methods.
- **I** The reported value is between the laboratory method detection limit and the laboratory practical quantitation limit.
- J Estimated value. A "J" value shall be accompanied by a narrative justification for its use. Where possible, the organization shall report whether the actual value is less than or greater than the reported value. A "J" value shall not be used as a substitute for K, L, M, T, V, or Y, however, if additional reasons exist for identifying the value as estimate (e.g., matrix spiked failed to meet acceptance criteria), the "J" code may be added to a K, L, M, T, V, or Y. The following are some examples of narrative descriptions that may accompany a "J" code: .
  - J1: No known quality control criteria exist for the component;
  - J2: The reported value failed to meet the established quality control criteria for either precision or accuracy (the specific failure must be identified);
  - J3: The sample matrix interfered with the ability to make any accurate determination;
  - J4: The data are questionable because of improper laboratory or field protocols (e.g., composite sample was collected instead of a grab sample).
  - J5: The field calibration verification did not meet calibration acceptance criteria.
  - J6: QC protocol not followed.

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5200 Wurzhach, Sta 104 San Antonio TX 78229	(972) 481-9999	(972) 481-9998
2505 N Falkenburg Pd, Tampa EI 33610	(210) 509-3334	(201) 509-3335
5757 NW 158th St. Miami Lakes EL 33014	(813) 620-2000	(813) 620-2033
5757 NW 156th St, Mann Eakes, 1E 55014	(305) 823-8500	(305) 823-8555



J7: B/A results for Chlorophyll does not meet 1 - 1.7 ratio.

- **K** Off-scale low. Actual value is known to be less than the value given. This code shall be used if:
- **1.** The value is less than the lowest calibration standard and the calibration curve is known to be non-linear; or
- 2. The value is known to be less than the reported value based on sample size, dilution. This code shall not be used to report values that are less than the laboratory practical quantitation limit or laboratory method detection limit.
- L Off-scale high. Actual value is known to be greater than value given. To be used when the concentration of the analyte is above the acceptable level for quantitation (exceeds the linear range or highest calibration standard) and the calibration curve is known to exhibit a negative deflection.
- **M** When reporting chemical analyses: presence of material is verified but not quantified; the actual value is less than the value given. The reported value shall be the laboratory practical quantitation limit. This code shall be used if the level is too low to permit accurate quantification, but the estimated concentration is greater than the method detection limit. If the value is less than the method detection limit use "T" below.
- **N** Presumptive evidence of presence of material. This qualifier shall be used if:
- 1. The component has been tentatively identified based on mass spectral library search; or
- 2. There is an indication that the analyte is present, but quality control requirements for confirmation were not met (i.e., presence of analyte was not confirmed by alternative procedures).
- **O** Sampled, but analysis lost or not performed.
- **Q** Sample held beyond the accepted holding time. This code shall be used if the value is derived from a sample that was prepared or analyzed after the approved holding time restrictions for sample preparation or analysis.
- **T** Value reported is less than the laboratory method detection limit. The value is reported for informational purposes, only and shall not be used in statistical analysis.
- **U** Indicates that the compound was analyzed for but not detected. This symbolshall be used to indicate that the specified component was not detected. The value associated with the qualifier shall be the laboratory method detection limit. Unless requested by the client, less than the method detection limit values shall not be reported (see "T" above).
- **V** Indicates that the analyte was detected in both the sample and the associated method blank. Note: the value in the blank shall not be subtracted from associated samples.

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3737 INW 13001 SI, IVITALIII LAKES, FL 33014	(305) 823-8500	(305) 823-8555



- **Y** The laboratory analysis was from an unpreserved or improperly preserved sample. The data may not be accurate.
- **Z** Too many colonies were present (TNTC); the numeric value represents the filtration volume.
- ? Data are rejected and should not be used. Some or all of the quality control data for the analyte were outside criteria, and the presence or absence of the analyte cannot be determined from the data.
  - \* Not reported due to interference.

The following codes deal with certain aspects of field activities. The codes shall be used if the laboratory has knowledge of the specific sampling event. The codes shall be added by the organization collecting samples if they apply:

- **D** The sample result was reported from a dilution.
- **E** Indicates that extra samples were taken at composite stations.
- **R** Significant rain in the past 48 hours. (Significant rain typically involves rain in excess of 1/2 inch within the past 48 hours.) This code shall be used when the rainfall might contribute to a lower than normal value.
- ! Data deviate from historically established concentration ranges.

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(210) 509-3334	(201) 509-3335
(813) 620-2000	(813) 620-2033
(305) 823-8500	(305) 823-8555





Work Order #: 337158			Project ID:								
Lab Batch #: 764534	Sa	ample: 764534-	1-BKS	Matri	ix: Water						
<b>Date Analyzed:</b> 07/06/2009	Date Pre	pared: 07/06/20	)09	Analys	st: MSH						
Reporting Units: CU	Ba	atch #: 1	BLANK /	BLANK SPI	KE REC	COVERY S	STUDY				
Color by SM2120B		Blank Result [A]	Spike Added [B]	Blank Spike Result	Blank Spike %R	Control Limits %R	Flags				
Color		<0.50	36	35	97	80-120					
I ab Batch #• 764878	S.	mnler 764878	1-BKS	Motri	w. Water						
Date Analyzed: 07/09/2009	oa Date Prei	mple: 704878- pared: 07/09/20	)09	Analys	st: MAB						
<b>Reporting Units:</b> mg/L	Bate I Per	ntch #: 1	BLANK /	BLANK SPI	KE REC	COVERY S	STUDY				
DOC by SM5310 Analytes		Blank Result [A]	Spike Added [B]	Blank Spike Result [C]	Blank Spike %R [D]	Control Limits %R	Flags				
Dissolved Organic Carbon		<0.548	10.0	9.84	98	80-120					
Lab Patch # 7(492)		1 764826	1 DKG		W-4						
Lab Balcii #: 764826	Si Date Prei	ample: 704820- pared: 07/08/20	)09	Matri A nalve	<b>st</b> MSH						
Reporting Units: mg/L	Bate I Iej Ba	atch #: 1	BLANK /BLANK SPIKE RECOVERY ST								
Phosphorus, Total (Automated) by EPA Analytes	365.4	Blank Result [A]	Spike Added [B]	Blank Spike Result [C]	Blank Spike %R [D]	Control Limits %R	Flags				
Total Phosphorus (as P)		<0.037	6.82	7.00	103	80-120					
Lab Batch #: 764582 Date Analyzed: 07/07/2009 Reporting Units: uS/cm	Sa Date Prej Ba	ample: 764582- pared: 07/07/20 atch #: 1	1-BKS )09 <b>BLANK</b> /J	Matri Analy: BLANK SPI	ix: Water st: MSH KE REC	COVERY S	STUDY				
Specific Conductance by EPA 120.1 Analytes	l	Blank Result [A]	Spike Added [B]	Blank Spike Result [C]	Blank Spike %R [D]	Control Limits %R	Flags				
Conductivity		<10.0	101	102	101	80-120					
Lab Batch #: 764766 Date Analyzed: 07/07/2009	Sa Date Prej	ample: 764766- pared: 07/07/20	1-BKS 009	Matri Analys	ix: Water st: YAD						
Reporting Units: mg/L	Ba	atch #: 1	BLANK /	BLANK SPI	KE REC	COVERY S	STUDY				
TDS by SM2540C Analytes		Blank Result [A]	Spike Added [B]	Blank Spike Result [C]	Blank Spike %R [D]	Control Limits %R	Flags				

Blank Spike Recovery [D] = 100\*[C]/[B] All results are based on MDL and validated for QC purposes.

BRL - Below Reporting Limit





Work Order #: 337158	Project ID:							
Lab Batch #: 764872	Sample: 76487	2-1-BKS	-BKS Matrix: Water					
Date Analyzed: 07/08/2009	Date Prepared: 07/08	2009	Analys	st: MAB				
<b>Reporting Units:</b> mg/L	<b>Batch #:</b> 1	BLANK /	BLANK /BLANK SPIKE RECOVERY ST					
TOC by SM 5310C	Blank Result	Spike Added	Blank Spike	Blank Spike	Control Limits	Flags		
Analytes	[A]	[B]	Result [C]	%R [D]	%R			
Total Organic Carbon	<0.242	10.0	9.84	98	90-110			

Blank Spike Recovery [D] = 100\*[C]/[B] All results are based on MDL and validated for QC purposes. BRL - Below Reporting Limit





work Order #: 33/158							Proj	ect ID:			
Analyst: OLA	Da	ate Prepar	ed: 07/07/200	)9			Date A	nalyzed: (	07/07/2009		
Lab Batch ID:         764635         Sample:         764635	-BKS Batch #: 1 Matrix: Water										
Units: mg/L	BLANK /BLANK SPIKE / BLANK SPIKE DUPLICATE RECOVERY STUDY										
Alkalinity by SM2320B	Blank Sample Result [A]	Spike Added	Blank Spike Result	Blank Spike %R	Spike Added	Blank Spike Duplicate	Blk. Spk Dup. %R	RPD %	Control Limits %R	Control Limits %RPD	Flag
Analytes		[B]	[C]	[D]	[E]	Result [F]	[G]				
Alkalinity, Total (as CaCO3)	<2.17	100	100	100	100	98.0	98	2	89-106	20	
Alkalinity, Bicarbonate (as CaCO3)	<2.17	250	<8.68	0	250	<8.68	0	NC	80-117	20	J
Alkalinity, Carbonate (as CaCO3)	<2.17	250	<8.68	0	250	<8.68	0	NC	80-120	20	J
Analyst: NIB	Da	ate Prepar	ed: 07/07/200	)9			Date A	nalyzed: (	7/07/2009		
Lab Batch ID: 764833 Sample: 764833	1-BKS	Batch	n#: 1					Matrix: \	Water		
Lab Batch ID: 764833 Sample: 764833- Units: mg/L	1-BKS	Batch BLAN	n #: 1 K /BLANK \$	SPIKE / I	BLANK S	PIKE DUPI	LICATE 1	Matrix: V RECOVI	Water E <b>RY STUD</b>	Y	
Lab Batch ID: 764833 Sample: 764833- Units: mg/L Inorganic Anions by EPA 300	1-BKS Blank Sample Result [A]	Batch BLAN Spike Added	n #: 1 K /BLANK S Blank Spike Result	SPIKE / H Blank Spike %R	Spike Added	PIKE DUPI Blank Spike Duplicate Beguld Fel	Blk. Spk Dup. %R	Matrix: N RECOVI RPD %	Water ERY STUD Control Limits %R	Control Limits %RPD	Flag
Lab Batch ID: 764833 Sample: 764833- Units: mg/L Inorganic Anions by EPA 300 Analytes	1-BKS Blank Sample Result [A]	Batch BLAN Spike Added [B]	n #: 1 K /BLANK S Blank Spike Result [C]	SPIKE / I Blank Spike %R [D]	BLANK S Spike Added [E]	BIANK Blank Spike Duplicate Result [F]	LICATE I Blk. Spk Dup. %R [G]	Matrix: N RECOVI RPD %	Water ERY STUD Control Limits %R	Control Limits %RPD	Flag
Lab Batch ID: 764833       Sample: 764833         Units: mg/L       Inorganic Anions by EPA 300         Analytes       Fluoride	1-BKS Blank Sample Result [A] <0.092	Batch BLAN Spike Added [B] 5.00	n #: 1 K /BLANK S Blank Spike Result [C] 4.78	SPIKE / I Blank Spike %R [D] 96	Spike Added [E] 5	Blank Spike Duplicate Result [F] 4.66	Blk. Spk Dup. %R [G] 93	Matrix: N RECOVI RPD % 3	Water ERY STUD Control Limits %R 90-110	Y Control Limits %RPD 20	Flag
Lab Batch ID: 764833       Sample: 764833         Units: mg/L       Inorganic Anions by EPA 300         Analytes       Fluoride         Fluoride       Chloride	1-BKS Blank Sample Result [A]    <0.092	Batch BLAN Spike Added [B] 5.00 10.0	n #: 1 K /BLANK S Blank Spike Result [C] 4.78 9.56	SPIKE / I Blank Spike %R [D] 96 96	Spike Added [E] 5 10	Blank Spike Duplicate Result [F] 4.66 9.33	LICATE I Blk. Spk Dup. %R [G] 93 93	Matrix: N RECOVE RPD % 3 2	Water ERY STUD Control Limits %R 90-110 90-110	Control Limits %RPD 20 20	Flag
Lab Batch ID: 764833       Sample: 764833         Units: mg/L       Inorganic Anions by EPA 300         Analytes       Inorde         Fluoride       Inorde         Other ide       Inorde         Nitrite as N       Inorde	1-BKS  Blank Sample Result [A]   <	Batch           BLAN           Spike           Added           [B]           5.00           10.0           10.0	n #: 1 K /BLANK S Blank Spike Result [C] 4.78 9.56 9.38	SPIKE / I Blank Spike %R [D] 96 96 94	Spike Added [E] 5 10 10	Blank Spike Duplicate Result [F] 4.66 9.33 9.34	Blk. Spk           Dup.           %R           [G]           93           93           93	Matrix: N RECOVE RPD % 3 2 0	Water ERY STUD Control Limits %R 90-110 90-110 90-110	Y Control Limits %RPD 20 20 20 20	Flag
Lab Batch ID: 764833       Sample: 764833         Units: mg/L       Inorganic Anions by EPA 300         Analytes       Inorde         Fluoride       Inorde         Nitrite as N       Nitrate as N	1-BKS  Blank Sample Result [A]   <	Batch BLAN Spike Added [B] 5.00 10.0 10.0 4.52	n #: 1 K /BLANK S Blank Spike Result [C] 4.78 9.56 9.38 4.71	<b>SPIKE / I</b> Blank Spike %R [D] 96 96 96 94 104	Spike           Added           [E]           5           10           10           4.52	Blank Spike Duplicate Result [F] 4.66 9.33 9.34 4.72	Blk. Spk           Dup.           %R           [G]           93           93           93           104	Matrix: N RECOVE % 3 2 0 0 0	Water ERY STUD Control Limits %R 90-110 90-110 90-110 90-110	Y Control Limits %RPD 20 20 20 20 20	Flag
Lab Batch ID: 764833       Sample: 764833         Units: mg/L       Inorganic Anions by EPA 300         Analytes       Fluoride         Chloride       Nitrite as N         Nitrate as N       Ortho-Phosphate	Blank         Blank         Sample Result         [A]         <0.092	Batch           BLAN           Spike           Added           [B]           5.00           10.0           10.0           4.52           9.79	n #: 1 K /BLANK S Blank Spike Result [C] 4.78 9.56 9.38 4.71 9.18	Blank         Spike         %R         [D]           96         96         96         94         104         94	Spike           Added           [E]           5           10           4.52           9.79	Blank           Spike           Duplicate           Result [F]           4.66           9.33           9.34           4.72           9.22	Blk. Spk           Dup.           %R           [G]           93           93           93           93           93           93           93           93           93           93           93           93	Matrix: V RECOVE % 3 2 0 0 0 0	Water <b>CRY STUD</b> Control Limits %R 90-110 90-110 90-110 90-110 90-110	Control           Limits           %RPD           20           20           20           20           20           20           20           20           20           20           20           20           20           20           20           20           20           20	Flag

Relative Percent Difference RPD =  $200^{*}|(C-F)/(C+F)|$ Blank Spike Recovery [D] =  $100^{*}(C)/[B]$ Blank Spike Duplicate Recovery [G] =  $100^{*}(F)/[E]$ All results are based on MDL and Validated for QC Purposes





<b>Work Order #: </b> 337158	Project ID:											
Analyst: ARP	Da	ate Prepar	ed: 07/07/200	19			Date A	nalyzed: ()	7/08/2009			
Lab Batch ID: 764731         Sample: 533175-1-B	KS	Bate	<b>h #:</b> 1			Matrix: Water						
Units: mg/L		BLAN	K /BLANK S	SPIKE / E	BLANK S	K SPIKE DUPLICATE RECOVERY STUDY						
Metals per ICP-MS by SW 6020A	Blank Sample Result [A]	Spike Added	Blank Spike Result	Blank Spike %R	Spike Added	Blank Spike Duplicate	Blk. Spk Dup. %R	RPD %	Control Limits %R	Control Limits %RPD	Flag	
Analytes		[B]	[U]		[E]	Kesuit [F]	[G]					
Barium	< 0.002	0.100	0.108	108	0.1	0.107	107	1	75-125	25		
Calcium	< 0.050	5.00	5.60	112	5	5.43	109	3	75-125	25		
Potassium	< 0.170	5.00	5.23	105	5	5.08	102	3	75-125	25		
Sodium	< 0.250	5.00	5.45	109	5	5.20	104	5	75-125	25		
Strontium	< 0.001	0.100	0.103	103	0.1	0.105	105	2	75-125	25		
Analyst: 4150	Da	ate Prepar	ed: 07/07/200	19			Date A	nalyzed: 0	7/08/2009			
Lab Batch ID: 764783 Sample: 533209-1-B	KS	Bate	<b>h #:</b> 1					Matrix: V	Vater			
Units: mg/L		BLAN	K /BLANK S	SPIKE / E	BLANK S	SPIKE DUPI	LICATE ]	RECOVE	CRY STUD	Y		
Metals, Total by SW846 6010B	Blank Sample Result [A]	Spike Added [B]	Blank Spike Result [C]	Blank Spike %R [D]	Spike Added [E]	Blank Spike Duplicate Result [F]	Blk. Spk Dup. %R [G]	RPD %	Control Limits %R	Control Limits %RPD	Flag	
Iron	< 0.021	9.00	9.50	106	9	9.26	103	3	75-125	20		
Manganese	< 0.001	1.00	1.02	102	1	0.994	99	3	75-125	20		

Relative Percent Difference RPD =  $200^{*}|(C-F)/(C+F)|$ Blank Spike Recovery [D] =  $100^{*}(C)/[B]$ Blank Spike Duplicate Recovery [G] =  $100^{*}(F)/[E]$ All results are based on MDL and Validated for QC Purposes





Work Order #: 337158		Project ID:											
Analyst: SHH		Da	ate Prepar	ed: 07/07/200	)9		<b>Date Analyzed:</b> 07/07/2009						
Lab Batch ID: 764669	Sample: 764669-1-B	KS	Batcl	h#: 1			Matrix: Water						
Units: mg/L			BLANK /BLANK SPIKE / BLANK SPIKE DUPLICATE RECOVERY STUDY										
Nitrogen Ammonia by	EPA 350.1	Blank Sample Result [A]	Spike Added	Blank Spike Result	Blank Spike %R	Spike Added	Blank Spike Duplicate	Blk. Spk Dup. %R	RPD %	Control Limits %R	Control Limits %RPD	Flag	
Analytes			[ <b>B</b> ]	[C]	[D]	[E]	Result [F]	[G]			ľ		
Nitrogen, Ammonia (as N)		< 0.020	2.00	1.84	92	2	1.92	96	4	90-110	20		
Analyst: OLA		Da	ate Prepar	ed: 07/07/200	)9			Date A	nalyzed: ()	7/07/2009			
Lab Batch ID: 764640	Sample: 764640-1-B	KS	Batcl	<b>h #:</b> 1					Matrix: V	Vater			
Units: mg/L			BLAN	K /BLANK S	SPIKE / E	BLANK S	SPIKE DUPI	LICATE	RECOVE	ERY STUD	Y		
Sulfide by SM45 Analytes	00-S-F	Blank Sample Result [A]	Spike Added [B]	Blank Spike Result [C]	Blank Spike %R [D]	Spike Added [E]	Blank Spike Duplicate Result [F]	Blk. Spk Dup. %R [G]	RPD %	Control Limits %R	Control Limits %RPD	Flag	
Sulfide, total		<1.00	10.0	10.0	100	10	10.0	100	0	75-120	20		

Relative Percent Difference RPD =  $200^{*}|(C-F)/(C+F)|$ Blank Spike Recovery [D] =  $100^{*}(C)/[B]$ Blank Spike Duplicate Recovery [G] =  $100^{*}(F)/[E]$ All results are based on MDL and Validated for QC Purposes



# Form 3 - MS Recoveries

**Project Name: City of Hialeah** 



Work Order #: 337158						
Lab Batch #: 764635			Pr	oject ID:		
<b>Date Analyzed:</b> 07/07/2009 <b>Da</b>	te Prepared:	07/07/2009		Analyst:	OLA	
QC- Sample ID: 337158-006 S	Batch #:	1		Matrix:	Water	
Reporting Units: mg/L	MAT	RIX / MA	FRIX SPIKE	RECOV	/ERY STU	JDY
Alkalinity by SM2320B Analytes	Parent Sample Result [A]	Spike Added [B]	Spiked Sample Result [C]	%R [D]	Control Limits %R	Flag
Alkalinity, Bicarbonate (as CaCO3)	<8.68	20.0	<8.68	0	80-120	J
Alkalinity, Carbonate (as CaCO3)	<8.68	20.0	<8.68	0	80-120	J
Alkalinity, Total (as CaCO3)	132	50.0	180	96	80-120	1
Lab Batch #: 764833		1				<u> </u>
<b>Date Analyzed:</b> 07/07/2009 <b>Da</b>	te Prepared:	07/07/2009		Analyst:	NIB	
QC- Sample ID: 337134-001 S	Batch #:	1		Matrix:	Water	
Reporting Units: mg/L	MAT	RIX / MA	FRIX SPIKE	RECOV	/ERY STU	JDY
Inorganic Anions by EPA 300	Parent Sample Result [A]	Spike Added [B]	Spiked Sample Result [C]	%R [D]	Control Limits %R	Flag
Elucrida	<1.95	20.0	18.0	00	00.110	
Chlorida	171	100	261	90	90-110	
Nitrite os N	<2.00	100	01.2	90	90-110	
Nitrate as N	<2.00	40.0	91.5 43.3	91	90-110	
Ortho Phoenhate	<0.500	100	+3.5	103	90-110	1
Sulfate	202	100	387	05	90-110	1
Surate		100	507	)5	90-110	
Lab Batch #: 764731 Date Analyzed: 07/08/2009 Da	te Prepared:	07/07/2009		Analyst:	ARP	
OC- Sample ID: 337158-006 S	Batch #:	1		Matrix.	Water	
<b>Reporting Units:</b> mg/L	MAT	RIX / MA	<b>FRIX SPIKE</b>	RECOV	VERY STU	JDY
Metals per ICP-MS by SW 6020A Analytes	Parent Sample Result [A]	Spike Added [B]	Spiked Sample Result [C]	%R [D]	Control Limits %R	Flag
Barium	0.017	0.100	0.143	126	75-125	J
Calcium	141	5.00	140	0	75-125	J
Potassium	37.1	5.00	41.0	78	75-125	
Sodium	964	5.00	927	0	75-125	J
Strontium	12.4	0.100	11.8	0	75-125	J
Strontium	12.4	0.100	11.8	0	75-125	

Matrix Spike Percent Recovery [D] = 100\*(C-A)/BRelative Percent Difference [E] = 200\*(C-A)/(C+B)All Results are based on MDL and Validated for QC Purposes

BRL - Below Reporting Limit



# Form 3 - MS Recoveries

## **Project Name: City of Hialeah**



Work Order #: 337158						
Lab Batch #: 764731			Pr	oject ID:		
Date Analyzed: 07/08/2009	Date Prepared:	07/07/2009		Analyst:	ARP	
QC- Sample ID: 337191-001 S	Batch #:	1		Matrix:	Water	
Reporting Units: mg/L	MAT	TRIX / MAT	FRIX SPIKE	E RECOV	VERY STU	JDY
Metals per ICP-MS by SW 6020A	Parent Sample Result [A]	Spike Added [B]	Spiked Sample Result [C]	%R [D]	Control Limits %R	Flag
Analytes		[0]				
Barium	0.229	0.100	0.339	110	75-125	
Calcium	38.5	5.00	42.4	78	75-125	
Potassium	4.85	5.00	9.80	99	75-125	
Sodium	65.2	5.00	67.3	42	75-125	J
Strontium	0.322	0.100	0.420	98	75-125	
Lab Batch #: 764669						
Date Analyzed: 07/07/2009	Date Prepared:	07/07/2009		Analyst:	SHH	
QC- Sample ID: 337107-001 S	Batch #:	1		Matrix:	Water	
Reporting Units: mg/L	MAT	'RIX / MA'	FRIX SPIKE	E RECOV	VERY STU	JDY
Nitrogen Ammonia by EPA 350.1	Parent Sample Result	Spike Added	Spiked Sample Result [C]	%R [D]	Control Limits %R	Flag
Analytes	[]	[0]				
Nitrogen, Ammonia (as N)	0.828	2.50	2.65	73	90-110	J
Lab Batch #: 764826						
Date Analyzed: 07/08/2009	Date Prepared:	07/08/2009		Analyst:	MSH	
QC- Sample ID: 337158-006 S	Batch #:	1		Matrix:	Water	
Reporting Units: mg/L	MAT	TRIX / MA	FRIX SPIKE	E RECOV	VERY STU	JDY
Phosphorus, Total (Automated) by EPA 365.4 Analytes	Parent Sample Result [A]	Spike Added [B]	Spiked Sample Result [C]	%R [D]	Control Limits %R	Flag
Total Phoenhorus (as P)	<0.100	1.60	1.61	101	80-120	
rotar r nosphorus (as r)	\0.100	1.00	1.01	101	00-120	
Lab Batch #: 764640		07/07/0000			01.4	
Date Analyzed: 07/07/2009	Date Prepared:	07/07/2009		Analyst:	ULA	
QC- Sample ID: 337158-006 S	Batch #:	1		Matrix:	Water	
Reporting Units: mg/L	MAT	'RIX / MA'	FRIX SPIKE	E RECO	VERY STU	JDY
Sulfide by SM4500-S-F Analytes	Parent Sample Result [A]	Spike Added [B]	Spiked Sample Result [C]	%R [D]	Control Limits %R	Flag
Sulfide, total	2.40	5.00	6.41	80	75-120	

Matrix Spike Percent Recovery [D] = 100\*(C-A)/BRelative Percent Difference [E] = 200\*(C-A)/(C+B)All Results are based on MDL and Validated for QC Purposes

BRL - Below Reporting Limit



## Form 3 - MS / MSD Recoveries

#### **Project Name: City of Hialeah**



Work Order #: 337158 **Project ID:** Lab Batch ID: 764878 QC- Sample ID: 337158-006 S Matrix: Water Batch #: 1 Date Prepared: 07/09/2009 Analyst: MAB Date Analyzed: 07/09/2009 Reporting Units: mg/L MATRIX SPIKE / MATRIX SPIKE DUPLICATE RECOVERY STUDY Parent Spiked Sample Duplicate Spiked Control Spiked Control DOC by SM5310 Sample Spiked Sample Spike Result Sample Spike Dup. RPD Limits Limits Flag Result Added [C] %R Added Result [F] %R %R %RPD % Analytes [A] [B] [D] [E] [G] Dissolved Organic Carbon 2.47 5.00 6.55 82 5.00 6.74 85 3 80-120 20 Lab Batch ID: 764783 QC- Sample ID: 336994-001 S Batch #: 1 Matrix: Water Analyst: 4150 Date Prepared: 07/07/2009 Date Analyzed: 07/08/2009 Reporting Units: mg/L MATRIX SPIKE / MATRIX SPIKE DUPLICATE RECOVERY STUDY Parent Spiked Sample Spiked Duplicate Spiked Control Control Metals, Total by SW846 6010B Sample Spike Result Spiked Sample RPD Limits Limits Sample Spike Dup. Flag Result Added [C] %R Added Result [F] %R %R %RPD % Analytes [A] [**B**] [D] [E] [G] 0.223 Iron 9.00 9.49 103 9.00 9.44 102 1 75-125 20 99 1.00 99 0 Manganese 0.006 1.00 1.00 0.997 75-125 20 Lab Batch ID: 764872 Matrix: Water QC- Sample ID: 336507-025 S Batch #: 1 Date Prepared: 07/08/2009 Analyst: MAB Date Analyzed: 07/08/2009 Reporting Units: mg/L MATRIX SPIKE / MATRIX SPIKE DUPLICATE RECOVERY STUDY Parent Spiked Sample Spiked Duplicate Spiked Control Control TOC by SM 5310C Sample Spike Result Sample Spike Spiked Sample Dup. RPD Limits Limits Flag Result Added Added [C] %R Result [F] %R % %R %RPD Analytes [A] [**B**] [D] [E] [G] Total Organic Carbon 8.14 5.00 13.2 101 5.00 12.7 91 4 90-110 20

Matrix Spike Percent Recovery  $[D] = 100^{\circ}(C-A)/B$ Relative Percent Difference RPD =  $200^{\circ}[(C-F)/(C+F)]$  Matrix Spike Duplicate Percent Recovery [G] = 100\*(F-A)/E

ND = Not Detected, J = Present Below Reporting Limit, B = Present in Blank, NR = Not Requested, I = Interference, NA = Not ApplicableN = See Narrative, EQL = Estimated Quantitation Limit





Work Order #: 337158

Lab Batch #: 764635			Project I	D:	
Date Analyzed: 07/07/2009 Date Pro	epared: 07/0	07/2009	Analy	st: OLA	
<b>QC- Sample ID:</b> 337158-006 D	atch #: 1		Matr	ix: Water	
Reporting Units: mg/L	SAMPLE /	SAMPLE	DUPLIC	ATE REC	OVERY
Alkalinity by SM2320B Analyte	Parent Sample Result [A]	Sample Duplicate Result [B]	RPD	Control Limits %RPD	Flag
Alkalinity, Total (as CaCO3)	132	132	0	20	
I ah Ratch #• 764534	11		<u> </u>		1
Date Analyzed: 07/06/2009 Date Pre	epared: 07/0	6/2009	Analy	st: MSH	
<b>QC- Sample ID:</b> 337158-006 D <b>B</b>	atch #: 1		Matri	ix: Water	
Reporting Units: CU	SAMPLE /	SAMPLE	DUPLIC	ATE REC	OVERY
Color by SM2120B Analyte	Parent Sample Result [A]	Sample Duplicate Result [B]	RPD	Control Limits %RPD	Flag
Color	10	10	0	20	l
					1
Lab Batch #: 104070 Date Analyzed: 07/09/2009 Date Pr	enared: 07/0	19/2009	Analy	st: MAB	
	-purvus -	// _ ~ ~ /			
OC- Sample ID: 337158-006 D	atch #: 1		Matri	ix: Water	
QC- Sample ID: 337158-006 D B Reporting Units: mg/L	atch #: 1 SAMPLE	SAMPLE	Matri DUPLIC	ix: Water	OVERY
QC- Sample ID: 337158-006 D E Reporting Units: mg/L	Batch #: 1 SAMPLE / Parent Sample	SAMPLE	Matri DUPLIC	ix: Water ATE REC Control	OVERY
QC- Sample ID: 337158-006 D E Reporting Units: mg/L DOC by SM5310 Analyte	Batch #: 1 SAMPLE / Parent Sample Result [A]	/ SAMPLE Sample Duplicate Result [B]	Matr. DUPLIC	ix: Water ATE REC Control Limits %RPD	OVERY Flag
QC- Sample ID: 337158-006 D E Reporting Units: mg/L DOC by SM5310 Analyte Dissolved Organic Carbon	Batch #: 1 SAMPLE / Parent Sample Result [A] 2.47	/ SAMPLE Sample Duplicate Result [B] 2.27	Matr DUPLIC RPD	ix: Water ATE REC Control Limits %RPD 20	OVERY Flag
QC- Sample ID: 337158-006 D E Reporting Units: mg/L DOC by SM5310 Analyte Dissolved Organic Carbon Lab Batch #: 764731	Batch #: 1 SAMPLE / Parent Sample Result [A] 2.47	/ SAMPLE Sample Duplicate Result [B] 2.27	Matr DUPLIC RPD 8	ix: Water ATE REC Control Limits %RPD 20	OVERY Flag
QC- Sample ID: 337158-006 D E Reporting Units: mg/L DOC by SM5310 Analyte Dissolved Organic Carbon Lab Batch #: 764731 Date Analyzed: 07/08/2009 Date Pro	Batch #: 1 SAMPLE / Parent Sample Result [A] 2.47 2.47	/ SAMPLE Sample Duplicate Result [B] 2.27	Matr. DUPLIC. RPD 8 Analy	ix: Water ATE REC Control Limits %RPD 20 st: ARP	OVERY Flag
QC- Sample ID: 337158-006 D F Reporting Units: mg/L DOC by SM5310 Analyte Dissolved Organic Carbon Lab Batch #: 764731 Date Analyzed: 07/08/2009 Date Pro QC- Sample ID: 337191-001 D F	Batch #: 1 SAMPLE / Parent Sample Result [A] 2.47 2.47 2.47 2.47	/ SAMPLE Sample Duplicate Result [B] 2.27 17/2009	Matr DUPLIC. RPD 8 Analy Matr	ix: Water ATE REC Control Limits %RPD 20 st: ARP ix: Water	OVERY Flag
QC- Sample ID: 337158-006 D E Reporting Units: mg/L DOC by SM5310 Analyte Dissolved Organic Carbon Lab Batch #: 764731 Date Analyzed: 07/08/2009 Date Pro QC- Sample ID: 337191-001 D E Reporting Units: mg/L	Batch #: 1 SAMPLE, Parent Sample Result [A] 2.47 epared: 07/0 atch #: 1 SAMPLE,	/ SAMPLE Sample Duplicate Result [B] 2.27 7/2009	Matr DUPLIC RPD 8 Analy Matr DUPLIC	ix: Water ATE REC Control Limits %RPD 20 st: ARP ix: Water ATE REC	OVERY Flag
QC- Sample ID: 337158-006 D E Reporting Units: mg/L DOC by SM5310 Analyte Dissolved Organic Carbon Lab Batch #: 764731 Date Analyzed: 07/08/2009 Date Pro QC- Sample ID: 337191-001 D E Reporting Units: mg/L Metals per ICP-MS by SW 6020A Analyte	Batch #: 1 SAMPLE , Parent Sample Result [A] 2.47 2.47 2.47 2.47 2.47 2.47 2.47 2.47	/ SAMPLE Sample Duplicate Result [B] 2.27 97/2009 / SAMPLE Sample Duplicate Result [B]	Matr DUPLIC RPD 8 Analy Matri DUPLIC	ix: Water ATE REC Control Limits %RPD 20 st: ARP ix: Water ATE REC Control Limits %RPD	OVERY Flag OVERY Flag
QC- Sample ID: 337158-006 D E Reporting Units: mg/L DOC by SM5310 Analyte Dissolved Organic Carbon Lab Batch #: 764731 Date Analyzed: 07/08/2009 Date Pre QC- Sample ID: 337191-001 D E Reporting Units: mg/L Metals per ICP-MS by SW 6020A Analyte Barium	Batch #: 1 SAMPLE , Parent Sample Result [A] 2.47 Parent : 1 SAMPLE , Parent Sample Result [A] 0.229	/ SAMPLE Sample Duplicate Result [B] 2.27 7/2009 / SAMPLE Sample Duplicate Result [B] 0.226	Matri DUPLIC. RPD 8 Analy Matri DUPLIC. RPD	ix: Water ATE REC Control Limits %RPD 20 st: ARP ix: Water ATE REC Control Limits %RPD 25	OVERY Flag OVERY Flag
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QC- Sample ID: 337158-006 D E Reporting Units: mg/L DOC by SM5310 Analyte Dissolved Organic Carbon Lab Batch #: 764731 Date Analyzed: 07/08/2009 Date Pro QC- Sample ID: 337191-001 D E Reporting Units: mg/L Metals per ICP-MS by SW 6020A Analyte Barium Calcium Potassium	Batch #: 1 SAMPLE / Parent Sample Result [A] 2.47 epared: 07/0 atch #: 1 SAMPLE / Parent Sample Result [A] 0.229 38.5 4.85	/ SAMPLE Sample Duplicate Result [B] 2.27 17/2009 / SAMPLE Sample Duplicate Result [B] 0.226 37.4 4.78	Matr DUPLIC. RPD 8 Analy Matri DUPLIC. RPD 1 3 1	ix: Water ATE REC( Control Limits %RPD 20 st: ARP ix: Water ATE REC( Control Limits %RPD 25 25 25 25	OVERY Flag OVERY Flag
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Spike Relative Difference RPD 200 \* | (B-A)/(B+A) | All Results are based on MDL and validated for QC purposes. BRL - Below Reporting Limit





Work Order #: 337158

Lab Batch #: 764783			Project I	D:		
<b>Date Analyzed:</b> 07/08/2009 <b>Da</b>	te Prepared: 07/0	07/2009	Analy	st: 4150		
QC- Sample ID: 336994-001 D	<b>Batch #:</b> 1		Matr	ix: Water		
Reporting Units: mg/L	SAMPLE	/ SAMPLE	DUPLIC	ATE REC	OVERY	
Metals, Total by SW846 6010B	Parent Sample Result [A]	Sample Duplicate Result [B]	RPD	Control Limits %RPD	Flag	
Thinky to	0.223	0.227	2	20		
Iron Manganasa	0.225	0.227	2	20		
	0.000	0.000	0	20		
Lab Batch #: 764582 Date Analyzed: 07/07/2009 Da	te Prepared: 07/0	07/2009	Analy	st: MSH		
<b>QC- Sample ID:</b> 337158-006 D	<b>Batch #:</b> 1		Matr	ix: Water		
Reporting Units: uS/cm	SAMPLE	/ SAMPLE	DUPLIC	ATE REC	OVERY	
Specific Conductance by EPA 120.1 Analyte	Parent Sample Result [A]	Sample Duplicate Result [B]	RPD	Control Limits %RPD	Flag	
Conductivity	5670	5690	0	20		
L ah Ratch #. 764640			1		I	
<b>Date Analyzed:</b> 07/07/2009 <b>Da</b>	te Prepared: 07/0	07/2009	Analy	st: OLA		
<b>OC- Sample ID:</b> 337158-006 D	<b>Batch #:</b> 1		Matr	ix: Water		
Reporting Units: mg/L	SAMPLE	/ SAMPLE	DUPLIC	ATE REC	OVERY	
Sulfide by SM4500-S-F Analyte	Parent Sample Result [A]	Sample Duplicate Result [B]	RPD	Control Limits %RPD	Flag	
Sulfide, total	2.40	2.40	0	20		
Lab Batch #: 764766					1	
<b>Date Analyzed:</b> 07/07/2009 <b>Da</b>	te Prepared: 07/0	07/2009	Analy			
<b>QC- Sample ID:</b> 337134-010 D	Batch #: 1		Matrix: Water			
Reporting Units: mg/L	SAMPLE	/ SAMPLE	DUPLIC	ATE REC	OVERY	
TDS by SM2540C Analyte	Parent Sample Result [A]	Sample Duplicate Result [B]	RPD	Control Limits %RPD	Flag	
Total dissolved solids	384	408	6	30		

Spike Relative Difference RPD 200 \* | (B-A)/(B+A) |

All Results are based on MDL and validated for QC purposes.

BRL - Below Reporting Limit





Work Order #: 337158

Lab Batch #: 764872	Project ID:								
Date Analyzed: 07/08/2009	Date Pro	epared: 07/0	8/2009	Analy	st: MAB				
QC- Sample ID: 336507-025 D	E	atch #: 1		Matr	ix: Water				
Reporting Units: mg/L		SAMPLE	SAMPLE	DUPLIC	ATE REC	OVERY			
TOC by SM 5310C Analyte		Parent Sample Result [A]	Sample Duplicate Result [B]	RPD	Control Limits %RPD	Flag			
Total Organic Carbon		8.14	8.33	2	20				
Lab Batch #: 764535									
Date Analyzed: 07/06/2009	Date Pr	epared: 07/0	6/2009	Analy	st: MSH				
QC- Sample ID: 337158-006 D	E	atch #: 1		Matr	ix: Water				
Reporting Units: NTU		SAMPLE	SAMPLE	DUPLIC	ATE REC	OVERY			
Turbidity by EPA 180.1		Parent Sample	Sample		Control				
Analyte		Result [A]	Duplicate Result [B]	RPD	KIMITS %RPD	Flag			
Analyte		<b>Result</b> [A] 1.21	Duplicate Result [B] 1.25	<b>RPD</b>	20	Flag			
Analyte           Turbidity           Lab Batch #: 764568           Date Analyzed: 07/06/2009           QC- Sample ID: 337158-006 D	Date Pro	Result         [A]           1.21	Duplicate Result [B] 1.25 06/2009	Analy Matr	20 20 st: MSH ix: Water	Flag			
Analyte Turbidity Lab Batch #: 764568 Date Analyzed: 07/06/2009 QC- Sample ID: 337158-006 D Reporting Units: SU	Date Pro E	Result         [A]           1.21         1.21           epared:         07/0           satch #:         1           SAMPLE         1	Duplicate Result [B] 1.25 06/2009	Analy Matr DUPLIC	20 20 20 20 20 20 20 20 20 20 20 20 20 2	Flag			
Analyte Turbidity Lab Batch #: 764568 Date Analyzed: 07/06/2009 QC- Sample ID: 337158-006 D Reporting Units: SU pH by SM4500-H Analyte	Date Pro E	Result [A] 1.21 epared: 07/0 atch #: 1 SAMPLE Result [A]	Duplicate Result [B] 1.25 06/2009 / SAMPLE Sample Duplicate Result [B]	Analy Matr DUPLIC RPD	20 20 20 20 20 20 20 20 20 20 20 20 20 2	Flag OVERY Flag			

Spike Relative Difference RPD 200 \* | (B-A)/(B+A) | All Results are based on MDL and validated for QC purposes. BRL - Below Reporting Limit

Image: Instrument in transfer trans	om	'.xenco.c	WWW	Service and Quality	to Excellence in er from client comp	Committed	( <b>W</b> ), Liquid ( <b>L</b> ) alinguishment of these samples constitutes a va	), Solid(S), Water ( s document and re	Matrix: Air (A), Product (I Notice: Signature of thi
International and the second of the secon	vus ( <b>V</b> )	c (P), Vario	Clear (C), Plastic	Type: Glass Amb (A), Glass (	25 Cont	ious (V), Other_	ni VOA ( <b>40</b> ), 1L (1), 500ml ( <b>5</b> ), Tediar Bag ( <b>B</b> ), Va	<b>8</b> ), 32oz ( <b>32</b> ), 40m	Cont. Size: 4oz (4), 8oz (
Instrument         Instrum			Other (O)	;), None (NA), See Label (L), O	OH (Z), (Cool, <4C) (	H (A), ZnAc&Na(	12SO4 pH<2 ( <b>S</b> ), HNO3 pH<2 ( <b>N</b> ), Asbc Acid&NaC	/), HCI pH<2 (H), H	Preservatives: Various ()
Bit Production     Sender Vice Production     Production Production Production     Production Production     Production Production     Production Production     Production Production     Production Production     Production Production     Production Production     Production Production     Production Production     Production Production     Production Production     Production Production Production     Production Production     Production Product	unless otherwise until paid. Samples uested. Rush	d Conditions y of XENCO is hereby rec	ot XENCO terms an Intellectual Property ort is e-mailed unles yre-approved.	Upon signings this COC you accep agreed on writing. Reports are the will be held 30 days after final report Charges and Collection Fees are p	all I dal.		6) (//		5
Barner Charles Richards Ri		Temp:	Cooler	Total Containers per COC:	Date & Time	ials and Sign)	Date & Time Relinquisher to Un	itials and Sign)	Relinquished by (In
Sampler Name       Somplay Cly       Poly Name Location       Projection       Ball AL OF All Counting PL Transport Plants       Poly Name Location       Device Projection					Image: Section of the sectio	3 2 1 2 X 3 X 3 X 3 X 3 X 3 X 3 X 3 X 3 X 3 X	$ qoo \\  qoo \\ $	50/10/ 20/10/ 20/10/20 20/10/20 20/10/20 20/10/20 20/10/20 20/10/20 20/10/20	TP-1 23 MS
Company_City       Company_City       Company_City       Company_City       South Cadina 803-94000       Sou	Sample Clean-u	Addn: PAH above Hold Samples (Su	TAT ASAP 5h	EDB/DBCP Chlor.des Hardness	FL PRO DRO G SVOCs: Full-List OC Pesticides PC Metals: RCRA-8 F SPLP - TCLP (M	Preservatives VOCs Full-List VOCs PP TCL PAHs	Depth ft' In" m Matrix Composite Grab # Containers Container Size Container Type	Sampling Date	Sample ID
Image: Section of the sector of the secto	ps a	rchar	125		GRO DW CBs RCR Metal	BTE: DW	Signature		Sampler Name
Image: State       2505 Falkenburg RJ, Tampa, Fl 33569       813-520.2000       Philadephia/New Jersey 610-955-5649         Image: State       6017 Financial Drive, Norcoss, Georgia 30071       170-449-8800       South Carolina       South Caroli	are pre-approved a	mg/L W, mg irges will apply and a	24h 48h 3d 5d		MA EPH MA VP BN&AE TCL PP Herbicides OP Pe RA-4 Pb 13PP 23TA	X-MTBE EtOH Ox Appdx-1 Appdx-2	TO NO: LI Call for P.O. III Waste-Disp NPDES DW GA HSRA DOE DOD USACE OTHER: See Lab PM Included Call PM )	Y-CLEAN Land-F P AFCEE NAVY DAPP MDLS RLS	Reg Program: UST DF QAPP Per-Contract CL Special DLs ( GW DW (
Provenue       2505 Falkenburg Rd, Tampa, FI 33569       813-620-2000       Philadephila/New Jersey 610-955-5649         Provenue       6017 Financial Drive, Norcross, Georgia 30071       770-449-8800       South Carolina       803-543-8099       Other       Serial #: 251825       Page of         Company-City       Provenue       6017 Financial Drive, Norcross, Georgia 30071       770-449-8800       South Carolina       South Carolina       803-543-8099       Other       Serial #: 251825       Page of         Company-City       Provenue       234-206 - 0814       Lab Only:       337158       37158       251825       Page of         Provenue       Provenue       Project ID       Project ID       TAT: ASAP 5h 12h 24h 48h 3d 5d 7d 10d 21d Standard TAT is project specific.       158         Proj Statel:       AL, Eb, GA, LA, MS, NC, NJ       Proj Manager (PM)       It is typically 5-7 Working Days for level II and 10+ Working days for level III and IV data.       Project III         PA, SC, TIN, TX, UT Other       Fax No:       VO	as neede	/Kg S Hig are pre-ap	1 7d 10		H Appdx-2 esticides AL Appdx Pest. He	yg VOHs CALL	with Final Report	ng 🔲 Inc. Invoice	Invoice to Accounti Bill to: Cuote/Pricing:
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arrestore Lic       6017 Financial Drive, Norcess, Georgia 300 11, 770-449-8800       South Carolina 803-543-8099       Other       Serial #: 251826       Page         Name-Location       IP Previously done at XENCO       Project ID       TAT: ASAP 5h 12h, 24h, 48h, 3d 5d 7d 10d 21d Standard TAT is project specific.       South Carolina 803-543-8099       Other       South Carolina 803	8h 3 W, I appl	, H 2	S S	PH I E TC des 13PF	<u> </u> = EtC		E OTHER:	DOE DOD USAC	P AFCEE NAVY	Per-Contract Cl
ar transport Lit       017 Financial Drive, Norcross, Georgia 3001 770 449-8800       C Innauspirul new versey or u-so-overy         pany-city       (F1, M, S)       Sq2x6 - 051 H       Lab Only:       South Carolina 803-543-8099       Other       Serial #: 251826       Page         Name-Location       Innoice must have a P.O       Proviously done at XENCO       Project ID       TAT: ASAP 5h 12h 24h 48h 3 5d 7d 10d 21d Standard TAT is project specific.       South Carolina 803-543-8099       Innoice must have a P.O       It is typically 5-7 Working Days for level II and 10+ Working days for level III and IV data.         sc.       TN, TX, UT Other       Sc. TH, TX, UT Other       Sc. TH, TX, UT Other       Fax No:       Fax	3d ly ar ove	5		MA il P OF 23	H DH ppd	ISRA	DES DW GAH	I Waste-Disp NP	RY-CLEAN Land-Fi	Program: UST D
a target server LLC       Got 7 Financial Drive, Norcross, Georgia 30071 770-449-8800       Frinancup Induce with Final Report       Phone       South Carolina 803-543-8099       Other       Serial #: 251826       Page         Pany-City       CFT. Much       3:4 - 20.6 - 0.6 I H       Lab Only:       3:5 + 24h       48h 3d       5d       7d       10d       21d       Standard TAT is project specific.         Hame-Location       Proj. Manager (PM)       Fri ASAP 5h       12h       24h       48h       3d       5d       7d       10d       21d       Standard TAT is project specific.         Hame-Location       Fraine       Scott Manager (PM)       It is typically 5-7 Working Days for level II and 10+ Working days for level III and 10+ Working days for level III and IV data.         State:       AL.       Other       Fax No:       Page       Page <td< td=""><td>5d mg/Kg id are d as</td><td>โบณ</td><td>Cs P</td><td>VPH P Ap Pest</td><td>Oxyg &lt;-2</td><td>P.O.</td><td>Call for</td><td>9.0 No:</td><td></td><td>e/Pricing:</td></td<>	5d mg/Kg id are d as	โบณ	Cs P	VPH P Ap Pest	Oxyg <-2	P.O.	Call for	9.0 No:		e/Pricing:
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Notice: Signature of this document and relinquishment of these samples constitutes a valid purchase order from client company to Xenco Laboratories and its affiliates, subcontractors and assigns under Xenco's standard terms and conditions of service unless previolusly negotiated under a fully executed client contract.

Image:	Ico.com	www.xen	Service and Quality	to Excellence in S	Committed		lid(S), Water (W), Liquid (L)	Matrix: Air (A), Product (P), So
Image: The second se	Various (V)	Clear (C), Plastic (P),	Type: Glass Amb (A), Glass (	Cont.	ious (V), Other	. (1), 500ml (5), Tedlar Bag (B), Var	boz ( <b>32</b> ), 40ml VOA ( <b>40</b> ), 11	Cont. Size: 4oz (4), 8oz (8), 32
Image:		vre-approved. Vther ( <b>O</b> )	Charges and Collection Fees are pr	DH (Z). (Cool.<4C) (C	H (A), ZnAc&Na(	3), HNO3 pH<2 (N), Asbc Acid&NaO	1 pH<2 (H), H2SO4 pH<2 (	<sup>3</sup> reservatives: Various (V), HC
Barrier	ditions unless other ENCO until paid. Sa eby requested. Rusl	xt XENCO terms and Con Intellectual Property of XI xrt is e-mailed unless here	Upon signings this COC you accept agreed on writing. Reports are the l will be held 30 days after final repo	3/1 00/07			90// G	
Image: Image	0	Cooler Tem	Total Containers per COC:	Date 🎗 Time 🖉	tials and Sign)	Time Relinquished (Ini	and Sign) Date &	Relinquished by (Initials a
Sampler III       Signature       Signature <td></td> <td></td> <td></td> <td></td> <td></td> <td></td> <td></td> <td></td>								
Son Frienderson FT Types       Project Times       Bit Transition Project Times       Project Times       Son Frienderson Comparison       Project Times       Son Frienderson Comparison<	·				<b>F</b>	X 4 21 1 X	07/05 400	TP-12
Dispersive       2365 Flameburg RL times F13569       F13669       F13669       F13669<		i 			-96 <u>-</u>	W X 1 12 7 1	57/05 1400	TP-1 Z O
Image: Type Part Container Type       Sampler Name       Sampler					ř	シメートヤ	7/05 1400	7P-12 0
Sampler ID       Sampler ID <td> [</td> <td></td> <td></td> <td></td> <td>hie</td> <td></td> <td>7/05 (400</td> <td>TP-17 0</td>	[				hie		7/05 (400	TP-17 0
Provide work       2005 Falkenburg Rd. Tampa Fi 3369       819-820-2000       Philadephia/New Jersey 610-956649         Ford work work       6017 Financial Dirve. Norcess. Georgia 30071       770-449-8800       South Carolina 803-643-8009       Other       Serial #: 251827       Page         Proj Kame Location       Project ID       Project ID       Project ID       Norcess. Georgia 30071       170-449-8800       South Carolina 803-643-8009       Other       Serial #: 251827       Page         Proj Kame Location       Devices of Financial Dirve. Norcess. Georgia 30071       T70-449-8800       Iso Only:       South Carolina 803-643-8009       Other       Serial #: 251827       Page         Proj Kame Location       Project ID       Project ID       Nin Mark 2009       Int Xing Days for level II and V 448.       Serial #: 251827       Page         Proj Kame Location       Devices to DFM or       Fax No:       Fax No:       Int Xing Days for level III and V 448.       Serial #: 251827       Page         Proj Kame Location       DFM or Noice must Final Report       Invoice must Final Report       Int Xing Days for level III and V 448.       Serial #: 251827       Page         OukeP Fining:       P O No:       Fax No:       Call for P O       Is and At Yer Ar Yer A	Hold Samples (Su Sample Clean-u	FAT ASAP 5h Addn: PAH above	EDB/DBCP N Nitsatc, Nit TOC	FL PRO DRO SVOCs: Full-List OC Pesticides P Metals: RCRA-8 SPLP - TCLP (	Preservatives VOCs Full-List VOCs PP TCL PAHs	Depth ft' In" m Matrix Composite Grab # Containers Container Size Container Type	Sampling Date Time	Sample ID
Image: Program: USD Falkenburg Rt. Tampa, F13369       813-520-2000       Philadephia/New Jersey 610-955-664       Philadephia/New Jersey 610-955-664         For Jampe Location       B017 Friancial Drive, Norcoss, Georgia 30071       770-449-8800       South Carolina       803-543-609       Other       Serial #: 251827       Page         Projection       Projection       Projection       Projection       Projection       Projection       Projection       South Carolina       803-543-609       Other       Serial #: 251827       Page         Proj State:       A.L. (D. CA. LA. MS, N.C. N.I.       Proj. Manager (PM)       Nat. ASAP 5h 12h. 2th. 4th. 3d. 5d. 7d. 10d. 21d. Standard TAT is projection       Track SAP 5h 12h. 2th. 4th. 3d. 5d. 7d. 10d. 21d. Standard TAT is projection       South Carolina at Serial #: 251827       Page         Proj State:       A.L. (D. CA. LA. MS, N.C. N.I.       Proj. Manager (PM)       Track SAP 5h 12h. 2th. 4th. 3d. 5d. 7d. 10d. 21d. 11 and Verset Section.       Track SAP 5h 12h. 2th. 4th. 3d. 5d. 7d. 10d. 21d. 11 and Verset Section.       Track SAP 5h 12h. 2th. 4th. 3d. 5d. 7d. 10d. 21d. 11 and Verset Section.         Bill to::       Bill to::       Po No:       Call for P.O.       Fax No:       Fax No:       Section 4d. Section 4	irchai ips a	12h	n`1@	GRO DW CBs RCR Meta	BTE DW		Signature	Sampler Name
Image: Sold Falkenburg Rd, Tampa, Fl 33569       813-620-2000       Philadephia/New Jersey 610-955-5649         Prove Teary Sorver, UC       6017 Financial Drive, Norcross, Georgia 30071       770-449-8800       South Carolina       803-543-8099       Other       Serial #:       251827       Page         Company-City       Chird       F1. Mulers       239       -266-0814       Lab Only:       33       35       7158         Proj Vame-Location       Previously done at XENCO       Project ID       TAT: ASAP 5h 12h 24h 48h 3d 5d 7d 10d 21d Standard TAT is project specific.         Proj State:       AL, (5), GA, LA, MS, NC, NJ, (Proj. Manager (PM)       Tat: AsaP 5h 12h 24h 48h 3d 5d 7d 10d 21d Standard TAT is project specific.	rges will apply and are pre-approved) are pre-approved as needed	24h 48h 3d 5d 7d 10d 21d ma/LW. ma/Ka S Hinhest Hit	OP. Alkalinity Biearborate Corpon	MA EPH MA VPH BN&AE TCL PP Appdx-2 CALL Herbicides OP Pesticides RA-4 Pb 13PP 23TAL Appdx 1 Appdx 2 Ils VOCs SVOCs Pest. Herb. PCBs	X-MTBE EtOH Oxyg VOHs VOAs Appdx-1 Appdx-2 CALL Other:	Fax No: Fax No: Invoice must have a P.O Call for P.O. p NPDES DW GA HSRA USACE OTHER: Included Call PM )	Sb.com Inc. Invoice with Final Rep Inc. Invoice with Final Rep P.O No: EAN Land-Fill Waste-Dis CEE NAVY DOE DOD CEE NAVY DOE DOD CEE NAVY See Lab PN	PA. SC, TN, TX, UT Other e-Mail Results to WPM or SMEM AMA MA Invoice to Accounting E Invoice to Accounting E Invoice to Reg Program: Compared to the temperature of tempera
Image: Service Line       Image: Service Line<	ecific.	for level III and IV dat	Ievel II and 10d 21d Stand	5-7 Working Days for	It is typically		MS, NC, NJ, Proj. Manage	City of Italic
□ 2505 Falkenburg Rd, Tampa, Fl 33569 813-620-2000 □ Philadephia/New Jersey 610-955-5649 Press Serves Luc General Temps		128	400		Lab Only:	239-206-0814	(Ft. Myers)	Proi Name Location
2505 Falkenburg Rd, Tampa, FI 33569 813-620-2000     Philadephia/New Jersey 610-955-5649     O C C C C C C C C C C C C C C C C C	821 Page	serial #: 251	Other	rolina 803-543-8099	South Ca	eorgia 30071 770-449-8800	17 Financial Drive, Norcross, G	
			5649	hia/New Jersey 610-955-	Philadep	3569 813-620-2000	05 Falkenburg Rd, Tampa, Fl 3	

The second secon



## Prelogin / Nonconformance Report Sample Log-In

Client:	Schlumberger
Date / Time:	4.6.9 11:15
Lab ID#:	337158
Initials:	VWE

## Sample Receipt Checklist

⊭1 Temperature of cooler?	$\square$			40°C
#2 Shipping container in good condition?	Xes	No	None	
#3 Samples received on ice?	(Yes)	No	N/A	Blue/Water
#4 Custody Seals intact on sample container/cooler?	Yes	(NO)	- N/A	
#5 Custody SeasIs intact on sample bottles/containers	Yes		N/A	
#6 Chain of Custody present?	Yes	No		
#7 Sample instructions complete of Chain of Custody?	Yes	No		
#8 Any missing/extra samples?	Yes	No		
#9 Chain of custody signed when relinquished/ received?	(Ya)	No		
#10 Chain of Cusody agrees with sample label(s)?	Kes	No		
#11 Container label(s) legible and intact?	(Yes)	No		
#12 Sample matrix/ properties agree with Chain of Cusotdy?	Ker	No		
#13 Sample in proper conatiner/bottle?	Vest	No		
#14 Samples properly preserved?	YEST	No	N/A	
#15 Sample container(s) intact?		No		
#16 Sufficient sample amount for indicated test (s)?	(es)	No		
#17 All samples received within sufficient hold time?	Yes	Ma		
#18 Subcontract of sample(s)?	Yes	No		
#19 VOC samples have zero headspace?	Yes	(780)	N/A	
Nonconformance Docum	entation	)		

Contact:	Contacted by:	Date/Time:
Regarding:		
Corrective Action Take	n:	
Check all that Apply:	Client understands and would like to p	roceed with analysis
nander in einen eine eine eine eine eine ein	Cooling process had begun shortly a	iter sampling event

الإيدانية ستحجزت

	= 34 91 <sup>°</sup> - 33 55 - 33 55	<u>Abbrex</u> Gat and 32 amh 40mL - 32 W/w	10	9	¢e.	7	6	<u>с</u>	<u></u>	_بن	C4		Container Type/ Pres.	Tests	
	= 32oz = 16 oz 8oz Pl	$\frac{\text{fations}}{b = 0n}$ $\frac{b = 0n}{amb} = 32 \text{ o}$ $\frac{amb}{amb} = 40 \text{ mL}$						1					gal amb/		
	Plastic Plastic Plastic Bastic Ba	e galloi z. Ambi vials Wide	<b>m</b>		/								32 amb/		
	Bottle Piottle Diffe	n ambei Sr Month			1								32 amb/		4
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	- Steril ag . lic Bott	ide Mor ide Mor ide Mor		-		(	3						8 PE/ <b>NNO3</b>	metels	
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							>					••	16 PE/ HNO3	UU-254	
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# XENCO LABORATORIES Container Receipt Verification Form

Page 31 of 32

Reverse Osmosis Parameters						
Analyte	Analytical Method	Analytical Equipment	Container	Volume (mL)	Preservative	Maximun Holding Tir
Ammonia	350.1	Contract Laboratory	Р	250	Ice/H <sub>2</sub> SO <sub>4</sub>	28 d
Barium	200.7/200.8/6010/6020	Contract Laboratory	Р	250	lce/HNO3	180 d
Bicarbonate	310	Contract Laboratory	Р	250	Ice	7 d
Calcium		Contract Laboratory				
Carbonate	310	. Contract Laboratory	Р	250	Ice	7 d
Chloride	300	Contract Laboratory	Р	250	Ice	28 d
Color	110.2	Contract Laboratory	Р	250	Ice	48 hrs
Dissolved Organic Carbon HOU		Contract Laboratory				
Field Conductivity	120.1	Contract Laboratory	Р	250	Ice	7 d
Field pH	150.2	Contract Laboratory	Р	250	Ice	Immediat
Field Temperature		Contract Laboratory				
Fluoride	300	Contract Laboratory	Р	250	lce	28 d
Hydrogen Sulfide		Contract Laboratory				
Iron Atc.	200.7/200.8/6010/6020	Contract Laboratory	Р	250	Ice/HNO <sub>3</sub>	180 d
Manganese	200.7/200.8/6010/6020	Contract Laboratory	Р	250	Ice/HNO <sub>3</sub>	180 d
Nitrate/Nitrite	300	Contract Laboratory	Р	250	Ice	48 hrs
Orthophosphate	300	Contract Laboratory	Р	250	Ice/H <sub>2</sub> SO <sub>4</sub>	48 hrs
Potassium	200.7/200.8/6010/6020	Contract Laboratory	Р	250	Ice/HNO <sub>3</sub>	180 d
Silica FGS a		Contract Laboratory				
Sodium	200.7/200.8/6010/6020	Contract Laboratory	Р	250	Ice/HNO <sub>3</sub>	180 d
Strontium	200.7/200.8/6010/6020	Contract Laboratory	Р	250	Ice/HNO <sub>3</sub>	180 d
Sulfate	300	Contract Laboratory	Р	250	Ice	28 d
Total Alkalinity		Contract Laboratory				
Total Dissolved Solids	160.1	Contract Laboratory	Р	500	Ice	7 <b>d</b>
Total Hardness	130.2	Contract Laboratory	Р	250	Ice	180 d
Total Organic Carbon HOU	415.1/9060	Contract Laboratory	GA	500	Ice/H <sub>2</sub> SO <sub>4</sub>	28 d
Total Phosphate		Contract Laboratory				
Total Sulphides						
Turbidity	180.1	Contract Laboratory	Р	500	Ice	Immediat
UV-254 DUCODAS		Contract Laboratory			•	

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Project:Primary & Secondary TestingSite Location:Hialeah, FLMatrix:Drinking Water

Page 1 of 8 Report Printed: 07/23/09 Submission # 907000097 Order # 116139

Sample I.D.:	S/P	
Collected:	07/05/09	13:00
Received:	07/05/09	14:30
Collected by:	A.P./ L.G	

#### LABORATORY ANALYSIS REPORT

PARAMETER	RESULT	QC	UNITS	MDL	PQL	METHOD	DATE EXT.	DATE ANALY.	ANALYST
Coliform-Total (E-Coli)	A					9223B	07/06 10:50	07/07 10:50	AMC
Specific Conductance (Field)(grab)	5560		uS/cm	0.2	0.6	120.1	07/05 13:00	07/05 13:00	AP/LG
pH (field)	6.70		units	0.1	0.3	150.1	07/05 13:00	07/05 13:00	AP/LG
Temperature (Field)	22.1		Degree C	1	3	170.1	07/05 13:00	07/05 13:00	AP/LG
Turbidity (field)	1.4		NTU	0.1	0.3	180.1	07/05 13:00	07/05 13:00	AP/LG
Oxygen, Dissolved (Field)	7.9		mg/L	0.01	0.03	360.1	07/05 13:00	07/05 13:00	AP/LG
531.1 Carbamate Pesticides: 62-550.310	 )(4)(b) 	1	1	Dilution	Factor =	1			
Carbofuran	U	U	ug/L	0.45	1.35	531.1	07/13 18:28	07/13 18:28	RPV
Oxamyl (Vydate)	U	U	ug/L	0.41	1.23	531.1	07/13 18:28	07/13 18:28	RPV
Glyphosate	U	υ	ug/L	3.55	10 <b>.65</b>	547	07/07 13:56	07/07 13:56	RPV
549.2 Diquat: 62-550.310(4)(b)	r I	1	1	Dilution	Factor =	1			
Diquat	υ	υ	ug/L	0.4	1.2	549. <b>2</b>	07/08 10:30	07/08 12:26	RPV
Total Dissolved Solids (TDS)	3416		mg/L	1.00	3.00	EPA 160.1	07/06 16:06	07/07 16:06	LYR
Chloride	1430		mg/L	15.50	46.50	300.0	07/06 16:32	07/06 16:32	IMA
Fluoride	1.11		mg/L	0.080	0.240	300.0	07/06 16:32	07/06 16:32	IMA
Nitrate (as N)	υ	U	mg/L	0.010	0.030	300.0	07/06 16:32	07/06 16:32	IMA
Nitrate+Nitrite (as N)	υ	U	mg/L	0.012	0.036	300.0	07/06 16:32	07/06 16:32	ІМА
Nitrite (as N)	υ	U	mg/L	0.012	0.036	300.0	07/06 16:32	07/06 16:32	IMA
	t -	1	1	1			1	1	

Florida-Spectrum Environmental Services, Inc. 1460 W. McNab Road, Fort Lauderdale, FL 33309

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Pembroke Laboratory 528 Gooch Rd. Fort Mead, FL 33841 •

Big Lake Laboratory 415 B SW Park St. Okeechobee, FL 34972 www.flenviro.com Spectrum Laboratories 630 Indian St. Savannah, GA 31401

Project:Primary & Secondary TestingSite Location:Hialeah, FLMatrix:Drinking Water

Page 2 of 8 Report Printed: 07/23/09 Submission # 907000097 Order # 116139

Sample I.D.:	S/P	
Collected:	07/05/09	13:00
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Collected by:	A.P./ L.G	

PARAMETER	RESULT	QC	UNITS	MDL	PQL	METHOD	DATE EXT.	DATE ANALY.	ANALYST
Ortho-Phosphate (as P)	0.42		mg/L	0.080	0.240	300.0	07/06 16:32	07/06 16:32	IMA
Sulfate	460		mg/L	2.00	6.00	300.0	07/06 16:32	07/06 16:32	IMA
Alkalinity, Total (CaCO3) Endpoint 4.3	122		mg/L	0.1	0.3	310.1	07/08 10:38	07/08 10:38	DGK
Bicarbonate	122		mg/L	0.01	0.03	310.1	07/08 10:38	07/08 10:38	DGK
Carbonate	0.16		mg/L	0.01	0.03	310.1	07/08 10:38	07/08 10:38	DGK
Nitrogen (Ammonium, NH4+)	0.41		mg/L	0.02	0.06	350.CALC	07/15 14:41	07/15 14:41	RPV
Sulfide	3.61		mg/L	0.050	0.150	376.2	07/07 15:47	07/07 15:47	IMA
Color/pH (Lab)	U	U	Pt-Co	1.0	3.0	SM2120B	07/06 12:52	07/06 12:52	IMA
Odor (Lab) at 40 Degrees C	8		TON	1.0	3.0	SM2150B	07/06 12:53	07/06 12:53	IMA
Cyanide, Total	0.0043	I	mg/L	0.002	0.006	SM4500CN-E	07/07 09:40	07/07 12:31	MSG
Silica	10.8		mg/L	0.22	0.66	SM4500-SiO2	07/09 11:55	07/09 11:55	LYR
Organic Carbon, Dissolved	1.5		mg/L	0.302	0.906	SM5310C	07/09 10:02	07/09 10:02	MSG
Organic Carbon, Total	2.0		mg/L	0.302	0.906	SM5310C	07/09 10:02	07/09 10:02	MSG
MBAS Surfactants	0.12		mg/L	0.033	0.099	SM5540C	07/07 09:00	07/07 09:00	LR
Hardness, Total	792		mg/L	0.50	1.50	200.7	07/06	07/06 16:04	IMN
Aluminum	0.021		mg/L	0.0069	0.0207	200.7	07/06	07/06 14:19	IMN
Calcium	94.4		mg/L	0.0091	0.0273	200.7	07/06	07/06 14:18	IMN
Copper	U	U	mg/L	0.0001	0.0003	200.7	07/06	07/06 14:18	IMN

Project:Primary & Secondary TestingSite Location:Hialeah, FLMatrix:Drinking Water

Page 3 of 8 Report Printed: 07/23/09 Submission # 907000097 Order # 116139

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 07/05/09
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 Collected by:
 A.P./ L.G

PARAMETER	RESULT	QC	UNITS	MDL	PQL	METHOD	DATE EXT.	DATE ANALY.	ANALYST
Iron	0.023		mg/L	0.0007	0.0021	200.7	07/06	07/06 14:18	IMN
Magnesium	135		mg/L	0.3100	0.9300	200.7	07/06	07/06 16:04	IMN
Manganese	U	υ	mg/L	0.00006	0.00018	200.7	07/06	07/06 14:18	IMN
Potassium	59.5		mg/L	0.0012	0.0036	200.7	07/06	07/06 14:18	IMN
Silver	U	υ	mg/L	0.0001	0.0003	200.7	07/06	07/06 14:18	IMN
Sodium	958		mg/L	0.110	0.330	200.7	07/06	07/06 16:00	IMN
Strontium	11.3		mg/L	0.00030	0.00090	200.7	07/06	07/06 14:18	IMN
Zinc	0.002	I	mg/L	0.00099	0.00297	200.7	07/06	07/06 14:18	IMN
200.8 DW-10 Metals in Drinking Wate	r 62-550.310		 	Dilution	Factor =	1			
Arsenic	0.0015		mg/L	0.0002	0.0006	4.1.3/200.8	07/06 09:00	07/06 15:37	IMN
Barium	0.0100		mg/L	0.00004	0.00012	4.1.3/200.8	07/06 09:00	07/06 15:37	IMN
Cadmium	υ	U	mg/L	0.00008	0.00024	4.1.3/200.8	07/06 09:00	07/06 15:37	IMN
Chromium	0.0021		mg/L	0.0001	0.0003	4.1.3/200.8	07/06 09:00	07/06 15:37	IMN
Lead	U	U	mg/L	0.00006	0.00018	4.1.3/200.8	07/06 09:00	07/06 15:37	IMN
Nickel	υ	U	mg/L	0.0002	0.0006	4.1.3/200.8	07/06 09:00	07/06 15:37	IMN
Selenium	υ	U	mg/L	0.0009	0.0027	4.1.3/200.8	07/06 09:00	07/06 15:37	IMN
Antimony	U	U	mg/L	0.0002	0.0006	4.1.3/200.8	07/06 09:00	07/06 15:37	IMN
Beryllium	U	U	mg/L	0,00003	0.00009	4.1.3/200.8	07/06 09:00	07/06 15:37	IMN

Project:Primary & Secondary TestingSite Location:Hialeah, FLMatrix:Drinking Water

Page 4 of 8 Report Printed: 07/23/09 Submission # 907000097 Order # 116139

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 07/05/09
 14:30

 Collected by:
 A.P./ L.G

PARAMETER	RESULT	QC	UNITS	MDL	PQL	METHOD	DATE EXT.	DATE ANALY.	ANALYST
Thallium	υ	υ	mg/L	0.00000	0.00001	4.1.3/200.8	07/06 09:00	07/06 15:37	IMN
Mercury	υ	U	mg/L	0.0001	0.0003	245.1	07/10	07/10 12:02	EN
Ultraviolet Absorption Method	0.070		1/cm	0.009	0.027	5910.B	07/06	07/06 11:45	EN
504.1 EDB, DBCP: 62-550.310(4)(b)	1		1	Dilution	Factor = 1	L			
1,2-Dibromo-3-Chloropropane (DBCP)	U	U	ug/L	0.004	0.012	EPA 504.1 EC	D 07/0910:52	07/09 19:14	DŞ
Ethylene Dibromide (EDB)	U	υ	ug/L	0.007	0.021	EPA 504.1 EC	D 07/0910:52	07/09 19:14	D\$
508 Pesticides & PCBs: 62-550.310(4)	(b)	i	1	Dilution	Factor =	L			
Hexachlorocyclopentadiene	U	υ	ug/L	0.015	0.045	508	07/07 15:42	07/09 03:15	DŞ
Hexachlorobenzene	υ	υ	ug/L	0.006	0.018	508	07/07 15:42	07/09 03:15	DŞ
v-BHC (Lindane)	U	υ	ug/L	0.005	0.015	508	07/07 15:42	07/09 03:15	D\$
Heptachlor	U	υ	ug/L	0.002	0.006	508	07/07 15:42	07/09 03:15	DS
Heptachlor Epoxide	υ	υ	ug/L	0.002	0.006	508	07/07 15:42	07/09 03:15	D\$
Endrin	υ	υ	ug/L	0.005	0.015	508	07/07 15:42	07/09 03:15	DS
Methoxychlor	U	υ	ug/L	0.005	0.015	508	07/07 15:42	07/09 03:15	DS
Arochlor 1016	U	U	ug/L	0.1	0.3	508	07/07 15:42	07/09 03:15	DS
Arochlor 1221	υ	υ	ug/L	0.1	0.3	508	07/07 15:42	07/09 03:15	DS
Arochlor 1232	U	υ	ug/L	0.1	0.3	508	07/07 15:42	07/09 03:15	DS
Arochlor 1242	U	U	ug/L	0.1	0.3	508	07/07 15:42	07/09 03:15	DS

Project:Primary & Secondary TestingSite Location:Hialeah, FLMatrix:Drinking Water

Page 5 of 8 Report Printed: 07/23/09 Submission # 907000097 Order # 116139

Sample I.D.:	S/P	
Collected:	07/05/09	13:00
Received:	07/05/09	14:30
Collected by:	A.P./ L.G	

PARAMETER	RESULT	QC	UNITS	MDL	PQL	METHOD	DATE EXT.	DATE ANALY.	ANALYST
Arochlor 1248	U	υ	ug/L	0.1	0.3	508	07/07 15:42	07/09 03:15	D\$
Arochlor 1254	υ	υ	ug/L	0.1	0.3	508	07/07 15:42	07/09 03:15	DS
Arochlor 1260	U	U	ug/L	0.1	0.3	508	07/07 15:42	07/09 03:15	D\$
Toxaphene	υ	υ	ug/L	0.21	0.63	508	07/07 15:42	07/09 03:15	D\$
Chlordane	υ	υ	ug/L	0.03	0.09	508	07/07 15:42	07/09 03:15	D\$
515.3 Chlorophenoxy Herbicides 62-550	(Reg)	i	l	Dilutior	Factor =	1			
Dalapon	υ	U	ug/L	0.50	1.50	515.3	07/11 11:53	07/15 11:53	AC
2,4-D	υ	υ	ug/L	0.09	0.27	515.3	07/11 11:53	07/15 11:53	AC
Pentachlorophenol	U	υ	ug/L	0.020	0.060	515.3	07/11 11:53	07/15 11:53	AC
2,4,5-TP (silvex)	υ	υ	ug/L	0.14	0.42	515.3	07/11 11:53	07/15 11:53	AC
Dinoseb	U	U	ug/L	0.2	0.6	515.3	07/11 11:53	07/15 11:53	AC
Picloram	U	υ	ug/L	0.09	0.27	515.3	07/11 11:53	07/15 11:53	AC
524.2 Trihalomethanes: 62-550.310(3)	THMs	1	 	Dilution	n Factor =	1			
Bromodichloromethane	U	U	ug/L	0.08	0.24	524.2	07/06 22:48	07/06 22:48	MMD
Dibromochloromethane	U	U	ug/L	0.06	0.18	524.2	07/06 22:48	07/06 22:48	MMD
Tribromomethane (Bromoform)	U	U	ug/L	0.08	0.24	524.2	07/06 22:48	07/06 22:48	MMD
Trichloromethane (Chloroform)	U	U	ug/L	0.07	0.21	524.2	07/06 22:48	07/06 22:48	MMD
TOTAL Trihalomethanes	U		ug/L			524.2	07/06 22:48	07/06 22:48	MMD

Project:Primary & Secondary TestingSite Location:Hialeah, FLMatrix:Drinking Water

Page 6 of 8 Report Printed: 07/23/09 Submission # 907000097 Order # 116139

Sample I.D.:	S/P	
Collected:	07/05/09	13:00
Received:	07/05/09	14:30
Collected by:	A.P./ L.G	

PARAMETER	RESULT	QC	UNITS	MDL	PQL	METHOD	DATE EXT.	DATE ANALY.	ANALYST
524.2 Volatile Organics: 62-550.310(4)	(a)	1	1	Dilutior	Factor =	1			
Vinyl Chloride	υ	υ	ug/L	0,08	0.24	524.2	07/06 22:48	07/06 22:48	MMD
1,1-Dichloroethylene	υ	υ	ug/L	0,06	0.18	524.2	07/06 22:48	07/06 22:48	MMD
Dichloromethane (Methylene Chloride)	υ	υ	ug/L	0.14	0.42	524.2	07/06 22:48	07/06 22:48	MMD
Trans-1,2-Dichloroethylene	υ	υ	ug/L	0.09	0.27	524.2	07/06 22:48	07/06 22:48	MMD
Cis-1,2-Dichloroethylene	υ	υ	ug/L	0.12	0.36	524.2	07/06 22:48	07/06 22:48	MMD
1,1,1-Trichloroethane	υ	υ	ug/L	0.10	0.30	524.2	07/06 22:48	07/06 22:48	MMD
Carbon Tetrachloride	υ	υ	ug/L	0.10	0.30	524.2	07/06 22:48	07/06 22:48	MMD
Benzene	υ	υ	ug/L	0.06	0.18	524.2	07/06 22:48	07/06 22:48	MMD
1,2-Dichloroethane	υ	υ	ug/L	0.12	0.36	524.2	07/06 22:48	07/06 22:48	MMD
Trichloroethylene	υ	υ	ug/L	0.18	0.54	524.2	07/06 22:48	07/06 22:48	MMD
1,2-Dichloropropane	υ	υ	ug/L	0.06	0.18	524.2	07/06 22:48	07/06 22:48	MMD
Тоluene	υ	υ	ug/L	0.06	0.18	524.2	07/06 22:48	07/06 22:48	MMD
1,1,2-Trichloroethane	υ	υ	ug/L	0.13	0.39	524.2	07/06 22:48	07/06 22:48	MMD
Tetrachloroethylene	U	U	ug/L	0.14	0.42	524.2	07/06 22:48	07/06 22:48	MMD
Chlorobenzene	U	υ	ug/L	0.06	0.18	524.2	07/06 22:48	07/06 22:48	MMD
Ethylbenzene	υ	υ	ug/L	0.07	0.21	524.2	07/06 22:48	07/06 22:48	MMD
Xylenes (Total)	U	U	ug/L	0.16	0.48	524.2	07/06 22:48	07/06 22:48	MMD

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Project:Primary & Secondary TestingSite Location:Hialeah, FLMatrix:Drinking Water

Page 7 of 8 Report Printed: 07/23/09 Submission # 907000097 Order # 116139

 Sample I.D.:
 S/P

 Collected:
 07/05/09
 13:00

 Received:
 07/05/09
 14:30

 Collected by:
 A.P./ L.G

PARAMETER	RESULT	QC	UNITS	MDL	PQL	METHOD	DATE EXT.	DATE ANALY.	ANALYST
Styrene	U	U	ug/L	0.08	0.24	524.2	07/06 22:48	07/06 22:48	MMD
1,4-Dichlorobenzene (para)	U	υ	ug/L	0.09	0.27	524.2	07/06 22:48	07/06 22:48	MMD
1,2-Dichlorobenzene (ortho)	U	U	ug/L	0.07	0.21	524.2	07/06 22:48	07/06 22:48	MMD
1,2,4-Trichlorobenzene	U	υ	ug/L	0.12	0.36	524.2	07/06 22:48	07/06 22:48	' MMD
525.2 Semivolatile Organics: 62-550.31	0(4)(b)	 	1	Dilution	Factor =	l			
Di(2-Ethylhexyl)phthalate	U	U	ug/L	0.04	0.12	525.2	07/10 11:52	07/14 11:52	AC
Di(2-Ethylhexyl)adipate	U	U	ug/L	0.01	0.03	525.2	07/10 11:52	07/14 11:52	AC
Benzo(a)pyrene	U	U	ug/L	0.02	0.06	525.2	07/10 11:52	07/14 11:52	AC
Pentachlorophenol	U	υ	ug/L	0.02	0.06	525.2	07/10 11:52	07/14 11:52	AC
Alachlor	U	U	ug/L	0.04	0.12	525.2	07/10 11:52	07/14 11:52	AC
Atrazine	U	U	ug/L	0.04	0.12	525.2	07/10 11:52	07/14 11:52	AC
Simazine	υ	υ	ug/L	0.06	0.18	525.2	07/10 11:52	07/14 11:52	AC
2,3,7,8-TCDD (Dioxin)	See Attch.					1613	07/09	07/14 22:12	E87611
Endothall	υ	υ	mg/L	0.0046	0.0138	548.1	07/10 17:30	07/20 18:31	E84809
Gross Alpha	21.8 ± 5.1		pCi/L	6.9	20.7	EPA 900.0	07/09 06:34	07/10 15:24	E83033
Radium-226	5.2 ± 0.6		pCi/L	0.1	0.3	EPA 903.1	07/10 09:50	07/17 15:30	E83033
Radium-228	0.8U ± 0.5		pCi/L	0.8	2.4	EPA Ra-05	07/10 09:50	07/16 13:34	E83033
TEM Asbestos in Water 17-550.310		 		Dilution	n Factor =	1			

Project: Primary & Secondary Testing Site Location: Hialeah, FL Drinking Water Matrix:

Page 8 of 8 Report Printed: 07/23/09 Submission # 907000097 **Order** # 116139

Sample I.D.: S/P 07/05/09 Collected: 13:00 Received: 07/05/09 14:30 Collected by: A.P./ L.G

#### LABORATORY ANALYSIS REPORT

PARAMETER	RESULT	QC	UNITS	MDL	PQL	METHOD	DATE EXT.	DATE ANALY.	ANALYST
Asbestos (# of Fibers > 10u Detected)	See Attch.					TEM	07/02 12:30	07/03 11:00	E86772
Asbestos (Conc. of fibers > 10u)	See Attch.					TEM	07/02 12:30	07/03 11:00	E86772

Unless indicated, soil results are reported based on actual (wet) weight basis.

Analytes not currently NELAC certified denoted by ~. Work performed by outside (subcontract) labs denoted by Cert.ID in Analyst Field. Results relate only to this sample. QC=Qualifier Codes as defined by DEP 62-160

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U=Analyzed for but not detected.

Q=Sample held beyond accepted holding time. I=Value is between MDL and PQL.

J=Estimated value.

Authorized CSM Signature (954) 978-6400 Florida-Spectrum Environmental Services, Inc.

Certification # E86006

Florida Radiochemistry Services, Inc.

# Case Narrative ORDER#116139

**NOTE: \*\*** Gross Alpha:

Sample 0907050-01 had an elevated detection limit and/or counting error due to a low volume of sample used. The sample had high TDS (Total Dissolved Solids). The high TDS interferes with the sample counting efficiency. This is caused by the solids absorbing the sample activity (Sample self-absorption). The sample counting efficiency is decreased because of this. Therefore, the counting time was increased (the sample was counted over night or as long as possible) to help reduce the detection limit and counting error.



July 21, 2009

Service Request No: E0900482

Marie Castellanos Florida-Spectrum Environmental Services 1460 W. McNab Road Fort Lauderdale, FL 33309

#### Laboratory Results for: TCDD by 1613B/Hickeh

Dear Marie:

Enclosed are the results of the sample(s) submitted to our laboratory on July 9, 2009. For your reference, these analyses have been assigned our service request number E0900482.

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Analyses were performed according to our laboratory's NELAP-approved quality assurance program. The test results meet requirements of the current NELAP standards, where applicable, and except as noted in the laboratory case narrative provided.

All results are intended to be considered in their entirety, and Columbia Analytical Services, Inc. (CAS) is not responsible for use of less than the complete report. Results apply only to the items submitted to the laboratory for analysis and individual items (samples) analyzed, as listed in the report. In accordance to the NELAC 2003 Standard, a statement on the estimated uncertainty of measurement of any quantitative analysis will be supplied upon request.

Please contact me if you have any questions. My extension is 2960. You may also contact me via email at KVerschoor@caslab.com.

Respectfully submitted,

**Columbia Analytical Services, Inc.** 

Karen Verschoor Project Manager

Page 1 of

For a specific list of NELAP-accredited analytes, refer to the certifications section at

www.caslab.com. An Employee Owned Company | NELAP Accredited

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Page 1 of 22

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# Columbia Analytical Services<sup>\*\*</sup>

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# **Certificate of Analysis**

19408 Park Row, Suite 320, Houston, TX 77084 Phone (713)266-1599 Fax (713)266-0130 <u>www.caslab.com</u> An Employee Owned Company

#### COLUMBIA ANALYTICAL SERVICES, INC

Client:Florida-Spectrum Environmental ServicesProject:HickehSample Matrix:Drinking Water

Service Request No.: Date Received: E0900482 7/9/09

#### CASE NARRATIVE

All analyses were performed in adherence to the quality assurance program of Columbia Analytical Services, Inc. (CAS). This report contains analytical results for samples designated for Tier II. When appropriate to the method, method blank results have been reported with each analytical test.

#### Sample Receipt

One drinking water sample was received for analysis at Columbia Analytical Services on 7/9/09.

The following discrepancies were noted upon initial sample inspection: no custody seals on cooler(s). The exceptions are also noted on the cooler receipt and preservation form included in this data package.

The sample was received at 0°C in good condition and is consistent with the accompanying chain of custody form. The sample was stored in a refrigerator at 4°C upon receipt at the laboratory.

#### **Data Validation Notes and Discussion**

#### MS/DMS

EQ0900256: Laboratory Control Spike/Duplicate Laboratory Control Spike (LCS/DLCS) samples were analyzed and reported in lieu of an MS/DMS for this extraction batch. The batch quality control criteria were met.

#### Y flags – Labeled Standards

Samples that had recoveries of labeled standards outside the acceptance limits are flagged with 'Y' flags. In all cases, the signal-to-noise ratios are greater than 10:1, making these data acceptable.

#### **Detection Limits**

Detection limits are calculated for each congener in each sample by measuring the height of the noise level for each quantitation ion for the associated labeled standard. The concentration equivalent to 2.5 times the height of the noise is then calculated using the appropriate response factor and the weight of the sample. The calculated concentration equals the detection limit.

Lan Le, Ph.D. 2009.07.22 10:40:29 -05'00' For XL 07/22/09 . Approved by Date

Xiangqiu Liang, Laboratory Director
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Client:Florida-Spectrum Environmental ServicesProject:TCDD by 1613B/Hickeh

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### SAMPLE CROSS-REFERENCE

SAMPLE #	CLIENT SAMPLE ID	DATE	TIME
E0900482-001	116139	7/5/09	13:00

# Laboratory Certifications 2009-2010

STATE/PROGRAM	AGENCY	CERTIFICATION ID	EXP DATE
ARIZONA	AZ-DHS	AZ0725	05/26/10
ARKANSAS	ADEQ	08-056-0	06/16/09
CALIFORNIA	CA-ELAP	2452	02/28/11
FLORIDA/NELAP	FL-DOHS	E87611	06/30/10
HAWAII	HI-DOH	N/A	06/30/10
ILLINOIS/NELAP	IL-EPA	002122	10/06/09
LOUISIANA/NELAP	LELAP	03048	06/30/10
MAINE	ME-DOHS	2008031	06/05/10
MICHIGAN	MIDEQ	9971	06/30/10
MINNESOTA	MDH	048-999-427	03/25/10
NEVADA	NDEP	TX014112009A	07/31/09
New Jersey	NJDEP	TX008	06/30/10
NEW MEXICO	NMED-DWB	N/A	06/30/10
NEW YORK/NELAP	NY-DOH	11707	03/31/10
NFESC/NAVY	NFESC	N/A	01/09/10
OKLAHOMA	OKDEQ	D9925, 9962	08/31/09
OREGON/NELAP	ORELAP	TX200002-006	03/24/10
TENNESSEE	TNDEC	04016	06/30/10
TEXAS/NELAP	TCEQ	T104704216-09-TX	06/30/10
UTAH/NELAP	UTELCP	COLU2	06/30/10
SOIL IMPORT PERMIT	USDA	P330-09-00067	03/27/12
WASHINGTON/NELAP	WA-Ecology	C1855	11/14/09
WEST VIRGINIA	WVDEP	347	06/30/10

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# Abbreviations, Acronyms & Definitions

Cal	Calibration
Conc	CONCentration
Dioxin(s)	Polychlorinated dibenzo-p-dioxin(s)
EDL	Estimated Detection Limit
EMPC	Estimated Maximum Possible Concentration
Flags	Data qualifiers
Furan(s)	Polychlorinated dibenzofuran(s)
g	Grams
ICAL	Initial CALibration
ID	IDentifier
lons	Masses monitored for the analyte during data acquisition
L	Liter (s)
LCS	Laboratory Control Sample
DLCS	Duplicate Laboratory Control Sample
MB	Method Blank
MCL	Method Calibration Limit
MDL	Method Detection Limit
MRL	Method Reporting Limit
mL	Milliliters
MS	Matrix Spiked sample
DMS	Duplicate Matrix Spiked sample
NO	Number of peaks meeting all identification criteria
PCDD(s)	Polychlorinated dibenzo-p-dioxin(s)
PCDF(s)	Polychlorinated dibenzofuran(s)
ppb	Parts per billion
ppm	Parts per million
ppq	Parts per quadrillion
ppt	Parts per trillion
QA	Quality Assurance
QC	Quality Control
Ratio	Ratio of areas from monitored ions for an analyte
% <b>Rec.</b>	Percent Recovery
RPD	Relative Percent Difference
RRF	Relative Response Factor
RT	Retention Time
RRT	Relative Retention Time
SDG	Sample Delivery Group
S/N	Signal-to-Noise ratio
TEF	Toxicity Equivalence Factor
TEQ	Toxicity Equivalence Quotient

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## Data Qualifier Flags – Dioxin/Furans

- o **B** Indicates the associated analyte is found in the method blank, as well as in the sample.
- C Confirmation of the TCDF compound: When 2378-TCDF is detected on the DB-5 column, confirmation analyses are performed on a second column (DB-225). The results from both the DB-5 column and the DB-225 column are included in this data package. The results from the DB-225 analyses should be used to evaluate the 2378-TCDF in the samples. The confirmed result should be used in determining the TEQ value for TCDF.
- E Indicates an estimated value used when the analyte concentration exceeds the upper end of the linear calibration range.
- J Indicates an estimated value used when the analyte concentration is below the method reporting limit (MRL) and above the estimated detection limit (EDL).
- **K** EMPC When the ion abundance ratios associated with a particular compound are outside the QC limits, samples are flagged with a 'K' flag. A 'K' flag indicates an estimated maximum possible concentration for the associated compound.
- **U** indicates the compound was analyzed and not detected.
- Y Samples that had recoveries of labeled standards outside the acceptance limits are flagged with 'Y'. In all cases, the signal-to-noise ratios are greater than 10:1, making these data acceptable.
- ND Indicates concentration is reported as 'Not Detected.'
- **S** Peak is saturated; data not reportable.
- P Indicates chlorodiphenyl ether interference present at the retention time of the target compound.
- o **Q** Lock-mass interference by chlorodiphenyl ether compounds.

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	and the second	
SR# Unique ID FOA	70482	
First Leve	el - Data Processing - to be filled by person generating the fo	ms.
Date 07/15/09	Person 1 (001)	
Date	Person 2	
Date OC CL	Primary Data Reviewer	-4 0
<u> </u>	Secondary Data Reviewer	
Date 7/22/09	rel - Review - to be filled by person doing project compliance r	eview
7/22/09		
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Analytical Report

Client:	Florida-Spectrum Environmental Services	Service Request:	E0900482
Project:	TCDD by 1613B/Hickeh	Date Collected:	7/ 5/09 1300
Sample Matrix:	Drinking Water	Date Received:	7/ 9/09
Sample Name:	116139	Units:	pg/L
Lab Code:	E0900482-001	Basis:	NA

# Polychlorinated Dibenzodioxins and Polychlorinated Dibenzofurans by HRGC/HRMS

Analytical Method: Prep Method: Sample Amount: Data File Name: ICAL Date:	1613B Method 1060mL U132157 05/02/08	13BDate Analyzed:2thodDate Extracted:60mLInstrument Name:32157Blank File Name:/02/08Cal Ver. File Name:				: 7/14/09 2212 : 7/9/09 : E-HRMS-01 : DB-5 : U132137 : U132151				
Analyte Name		Result	Q	EDL	М	RL	Ion Ratio	RRT	Dilution Factor	
2,3,7,8-TCDD		ND	U	0.786	9	.43			1	
Labeled Compounds		Spike Conc.(pg)		Conc. Found (pg)	%Rec	Q	Control Limits	Io: Ra	n tio RRT	
13C-2,3,7,8-TCDD		2000		1329.467	66		25-164	0.7	79 1.008	
37Cl-2,3,7,8-TCDD		800		650.121	81		35-197	N.	A 1.009	

Comments:

,

## Analytical Report

Client:	Florida-Spectrum Environmental Services	Service Request:	E0900482
Project:	TCDD by 1613B/Hickeh	Date Collected:	NA
Sample Matrix:	Drinking Water	Date Received:	NA
Sample Name:	Method Blank	Units:	pg/L
Lab Code:	EQ0900256-01	Basis:	NA

# Polychlorinated Dibenzodioxins and Polychlorinated Dibenzofurans by HRGC/HRMS

Analytical Method: Prep Method: Sample Amount: Data File Name: ICAL Date:	1613B Method 1000mL U132137 05/02/08						с	Date Analyzed: Date Extracted: Instrument Name: GC Column: Blank File Name: al Ver. File Name:	7/13/09 1930 7/9/09 E-HRMS-01 DB-5 U132137 U132135
Analyte Name		Result Q	EDL	М	IRL	Ion Ratio	RRT	Dilution Factor	
2,3,7,8-TCDD		ND U	1.13	1	0.0			1	
Labeled Compounds		Spike Conc.(pg)	Conc. Found (pg)	%Rec	Q	Control Limits	Ior Rat	1 io RRT	
13C-2,3,7,8-TCDD		2000	557.524	28		25-164	0.7	9 1.008	
37Cl-2,3,7,8-TCDD		800	257.087	32	Y	35-197	N	A 1.009	

Comments:

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QA/QC Report

Client: Project: Sample Matrix:	Florida-Spectrum Enviro TCDD by 1613B/Hickeh Drinking Water	onmental h	Services				Serv. Dat	ice Request: te Analyzed:	E090048 7/13/09 7/14/09	82 -
	Polychlorinated E	Dibenzoo	Lab Contr lioxins and P	ol Sample So olychlorinat	ummary ed Dibenzof(	urans by HRG	C/HRMS			
Analytical Method: Prep Method:	1613B Method							Units: Basis:	pg/L NA	
-							Ext	raction Lot:	90735	
		Lab E	Control Sam Q0900256-02	ple 2	Duplicat	te Lab Control EQ0900256-03	l Sample	% Rec		RPD
Analyte Name	Re	esult	Expected	% Rec	Result	Expected	% Rec	Limits	RPD	Limit
2,3,7,8-TCDD	2	233	200	117	240	200	120	67 - 158	3	50

Comments:

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Lab Control Sample Summary

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## Analytical Report

Client:	Florida-Spectrum Environmental Services	Service Request:	E0900482
Project:	TCDD by 1613B/Hickeh	Date Collected:	NA
Sample Matrix:	Drinking Water	Date Received:	NA
Sample Name:	Lab Control Sample	Units:	pg/L
Lab Code:	EQ0900256-02	Basis:	NA

## Polychlorinated Dibenzodioxins and Polychlorinated Dibenzofurans by HRGC/HRMS

Analytical Method: Prep Method: Sample Amount: Data File Name: ICAL Date:	1613B Method 1000mL U132142 05/02/08					I Ca	Date Analyzed: Date Extracted: nstrument Name: GC Column: Blank File Name: il Ver. File Name:	7/13/09 2328 7/9/09 E-HRMS-01 DB-5 U132137 U132135
Analyte Name		Result Q	EDL	MRL	Ion Ratio	RRT	Dilution Factor	
2,3,7,8-TCDD		233	0.439	10.0	0.77	1.001	1	
Labeled Compounds		Spike Conc.(pg)	Conc. Found (pg)	%Rec Q	Control Limits	Ion Ratie	) RRT	
13C-2,3,7,8-TCDD		2000	1525.245	76	25-164	0.78	1.008	
37Cl-2,3,7,8-TCDD		800	739.000	92	35-197	NA	1.009	

Comments:

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# Analytical Report

Client:	Florida-Spectrum Environmental Services	Service Request:	E0900482
Project:	TCDD by 1613B/Hickeh	Date Collected:	NA
Sample Matrix:	Drinking Water	Date Received:	NA
Sample Name:	Lab Control Sample Dup	Units:	pg/L
Lab Code:	EQ0900256-03	Basis:	NA

## Polychlorinated Dibenzodioxins and Polychlorinated Dibenzofurans by HRGC/HRMS

Analytical Method: Prep Method: Sample Amount: Data File Name: ICAL Date:	1613B Method 1000mL U132159 05/02/08					Ir J Ca	Date Analyzed: Date Extracted: astrument Name: GC Column: Blank File Name: I Ver. File Name:	7/14/09 2347 7/9/09 E-HRMS-01 DB-5 U132137 U132151
Analyte Name		Result Q	EDL	MRL	Ion Ratio	RRT	Dilution Factor	
2,3,7,8-TCDD		240	0.564	10.0	0.78	1.001	1	
Labeled Compounds		Spike Conc.(pg)	Conc. Found (pg)	%Rec Q	Control Limits	Ion Ratio	RRT	
13C-2,3,7,8-TCDD		2000	1207.611	60	25-164	0.77	1.008	
37Cl-2,3,7,8-TCDD		800	602.869	75	35-197	NA	1.009	

Comments:

16



19408 Park Row, Suite 320, Houston, TX 77084 Phone (713)266-1599 Fax (713)266-0130 www.caslab.com

An Employee Owned Company

Image: State of the state o	3D DUE DATE Requested	Fax: (954) 978-2233 Fax: (912) 234-4815 Fax: (863) 285-7030 Fax: 6863) 285-7030	Plake Sampler Copy Rush Surcharges apply				Email:			vsis Required Field Tests	T P C C	1) 0 A 7. A 7. A							OD TRANSRER SICONTURES DATE/TIME	NJ K- 7/8/05					Le Brown CAS 7/9/09 1030
Image: Control of the state	NE CUSTODY RECOF	5309 Tel: (954) 978-6400 11 Tel: (912) 238-5050 11 Tel: (863) 285-8145 14077 Tal: (863) 232-8145	Yellow-Lab File Copy	Report to	Invoice to	Site ALLEN	Fax:	Compa	Signature	Number of Containers	Received & NELAC	Letter Letter Suffixes							Total SAMPLE CUSTODY AN	1 Kelinquished by:	ditional charge) I Received by:	2 Relinquished by:	H2SO4 2 Received by: Na2S203-H2O	Unpreserved 3 Relinquished by: NaOH	H4-NH4CL 3 Received by:
#     MS-HL-SAC       630 Indian Stre     630 Indian Stre       630 Indian Stre     538 Gooch Rot       1112 NW Park     0riter #       0.100 H     0riter #       1112 NW Park     0riter #       111     0riter #       112     0riter #       111     0riter #       111     0riter #       112     0riter #       112     0riter #       113     0riter #	O CHAIN O	b Road Ft Laud. FL 33 et Savannah, GA 3140 td Fort Meade, FL 338 sc. Observing ET 3	urn W/report							Matrix Bottle	DW SW Pres.	S SED HW BIO Combo SFA OIL Codes		)							Yes No (aa	A-ascorbic acid P-1	CHCL SI Cu-CuSO4 T.	H-HNO3 U- M-MCAB N-	Z.zinc acetate N
Image: Sample     Sample       Image: Sample     Sample <td>145-Husto</td> <td>Contract Contract Contrac</td> <td>Onginal-Rei</td> <td></td> <td>Purchase</td> <td></td> <td>Phone:</td> <td></td> <td></td> <td>Date Time</td> <td></td> <td></td> <td>1/19 1.300</td> <td></td> <td> </td> <td></td> <td></td> <td></td> <td></td> <td></td> <td>QA/QC Report Needed?</td> <td>idle Type</td> <td>gbottle D-125 mil</td> <td>r / S\$-8 oz soli jar</td> <td></td>	145-Husto	Contract Contrac	Onginal-Rei		Purchase		Phone:			Date Time			1/19 1.300		 						QA/QC Report Needed?	idle Type	gbottle D-125 mil	r / S\$-8 oz soli jar	
	# 2 2 1 0 · 1		EIMIUIIIRUAL	75		7-20-7	N' - WILLS	( ) MARA		Sample	<b>a</b>		111.139 7							l" (sign hore) >		Comments A-liter amber	C B-Bacterla ba	L-liter bottle hrs S4-4 or soil ja	hrs V-40 ml vial

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## Columbia Analytical Services, Inc. Cooler Receipt Form

Received:         07/09/09         Opened (Date/Time):         07/09/09         1030         By:         NAB	
<ol> <li>Samples were received via? □US Mail √Fedex □UPS □DHL □Courier □Hand Delivered</li> <li>Samples were received in: (circle) √Cooler □Box □Other □NA</li> <li>Were custody seals present on coolers? □Y √N If yes, how many and where?</li> <li>If present, were custody seals intact? □Y □N If present, were they signed and dated? □Y [</li> <li>Is shipper's air-bill filed? □NA □Y √N If not, record air bill number:797745488164</li> </ol>	ז  אר
5. Temperature of cooler(s) upon receipt (°C): 0	<u></u>
6. If applicable, list Chain of Custody numbers:	
7. Were custody papers properly filled our (ink, signed, etc.)? $\Box NA \forall Y$	]N
8. Packing material used: Inserts I Bubble Wrap Blue Ice Wet Ice Sleeves Other	
9. Were the correct types of bottles used for the tests indicated? $\mathbf{\nabla} \mathbf{Y}$	N[
Did all bottles arrive in good condition (unbroken)? Indicate in the table below.	]N
Sample ID Bottle Count Bottle Type Out of Temp Broken Initials	
10.       Were all bottle labels complete (i.e. analysis, ID, etc.)?       ✓Y       ✓         Did all bottle labels and tags agree with custody papers? Indicate in the table below.       ✓Y       ✓	]N ]N
Sample ID on Bottle Sample ID on COC Sample ID on Bottle Sample ID on CO	

11. Additional notes, discrepancies, and resolutions:

# 19

# Sample Acceptance Policy

## Custody Seals (desirable, mandatory if specified in SAP):

- ✓ On outside of cooler
- ✓ Seals intact, signed and dated

## Chain-of-Custody documentation (mandatory):

- ✓ Properly filled out in ink & signed by the client
- ✓ Sign and date the coc for CAS/HOU upon cooler receipt
- ✓ Coc must list method number
- ✓ If no coc was submitted with the samples, complete a CAS/HOU coc for the client

### Sample Integrity (mandatory):

- ✓ Sample containers must arrive in good condition (not broken or leaking)
- ✓ Sample IDs on the bottles must match the sample IDs on the coc
- ✓ The correct type of sample bottle must be used for the method requested
- The correct number of sample containers received must agree with the documentation on the coc
- ✓ The correct sample matrix must appear on the coc
- $\checkmark$  An appropriate sample volume or weight must be received

## Temperature Preservatives (varies by sample matrix):

- ✓ Aqueous and Non-aqueous samples must be shipped and stored cold, at 0 to  $6^{\circ}$ C
- ✓ Tissue samples must be shipped and stored frozen, at -20 to -10°C
- ✓ Air samples can be shipped and stored at ambient temperature, ~23°C
- ✓ The sample temperature must be recorded on the coc
- Notify a Project Chemist if any samples are outside the acceptance temperature or have compromised sample integrity – the client must decide re: replacement sample submittal or continue with the analysis

### Cooler Receipt Form, CRF (mandatory):

- ✓ Cooler receipt forms must be completed for each coc & SR#
- ✓ Sample integrity issues must be documented on the CRF
- ✓ A scan of the carrier and the airbill number must be recorded in CAS LIMS

## Sample Integrity Issues/Resolutions (mandatory):

- ✓ Sample integrity issues are documented on the CRF and given to the Project Chemist for resolution with the client
- Client resolution is documented in writing (typically email or on the CRF) and filed in the project folder(s)

Summary	
Request S	70
Service ]	

Folder #: Client Name: Project Name: Project Number: Report To:	E0900482 Florida-Spectrum Enviro TCDD by 1613B Hickeh Marie Castellanos	nmental Services	Project Chemist: Originating Lab: Logged By: Date Received: Internal Due Date: QAP:	Karen Verschoor HOUSTON NBROWN 7/ 9/09 LAB QAP
Phone Number: Cell Number: Fax Number: E-mail:	Fort Lauderdale, FL 333 Fort Lauderdale, FL 333 954-978-6400 mcastellanos@flenviro.co	60 Hio	Qualifier Set: Formset: Merged?: Report to MDL?? P.O. Number: EDD:	CAS Standard CAS Standard N Y No EDD Specified
			WAS	
CAS Samp No	Client Samp No.	Matrix Collected	1313P) Dioxins Fusars	

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7/5/09 1300

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116139

E0900482-001

1 . 1000 ml-Glass Bottle NM AMBER Tefton Liner Unpreserved

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Location: E-WIC01

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Page 1 of 1

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Pre	3 <b>p Run#:</b> 9073:	5				Prep WorkFle	:WO	OrgExtAq(365)				Status: Prepped		
Te	am: Semiv	voa GCMS/AKOD	JUR			Prep Meth	:poi	Method			Prep Da	te/Time: 7/9/09 0]	1:00 PM	
#	Lab Code	Client ID		#	Method /Test	-	M Hq	latrix	Amt. Ext.	Sample Description	_			
	E0900466-001	N10-20000001		02	1613B/Dioxins Furans	<u>-</u>	14 14	Vater	1080mL	clear colorless liquid				
7	E0900476-001	9750		10.	1613B/Dioxins Furans	4	M L	Vater	1050mL	clear colorless liquid				
m	E0900480-001	19G0259-01		10	1613B/Dioxins Furans	Š	M S	Vater	1060mL	clear colorless liquid				
4	E0900481-001	OUTFALL 004		.01	1613B/Dioxins Furans	4	7	Vastewaler	1010mL	fight orange cloudy liquit	-13			
5	E0900481-002	OUTFALL 005		-01	1613B/Dioxins Furans	~	2	Vastewater	1020mL	dark orange cloudy liquic	F			
۶	E0900482-001	116139		10.	1613B/Dioxins Furans	••	ц С	hinking Water	1060mL	clear colorless liquid				
7	E0900483-001	001		10	1613B/Dioxins Furans	2	7 V	Vater	1040mL	clear colorless liquid				
∞	EQ0900256-01	MB			1613B/Dioxins Furans			iquid	1000mL					
6	EQ0900256-02	LCS			1613B/Dioxins Furans		1	iquid	1000mL					
10	EQ0900256-03	DLCS			1613B/Dioxins Furans		-	iquid	1000mL					
Ξ	EQ0900256-04	MB			1613B/Dioxins Furans		7	iquid	1000mL					
12	EQ0900256-05	LCS			1613B/Dioxins Furans		ц Г	iquid	1000mL					
13	EQ0900256-06	DLCS			1613B/Dioxins Furans		Ţ	iquid	1000mL					
Spi	iking Solutions						-							
L	Name: 1613B A	Matrix Working Stan	ıdard		Inventory ID 10	0213		Logbook Ref: I	<b>D10-43-5B</b>			Expires On: 05/2	28/2019	
1	EQ0900256-02 1	100.00µL	EQ0900256-03	100.00µL	EQ0900256-05	100.00µL		EQ0900256-06	100.00µL					ł
	Name: 8290/16	13B Cleanup Worki	ing Standard		Inventory ID 10	0768		Logbook Ref: I	D10-51-3A/B			Expires On: 07/0	01/2014	
J	E0900466-001	100.00µL	EQ0900256-01	100.00µL	EQ0900256-02	100.00µL		EQ0900256-04	100.00µL	ЕQ0900256-05	100.00µL			י ו
L	Name: 1613B I	Labeled Working Sta	andard		Inventory ID 1	0911		Logbook Ref: I	D10-52-4B			Expires On: 07/	09/2014	
J	E0900466-001 [ E0900483-001 ] EQ0900256-06 ]	1,000.00µL 1,000.00µL 1,000.00µL	E0900476-001 EQ0900256-01	1,000.00µL 1,000.00µL	E0900480-001 EQ0900256-02	1,000.00µ1 1,000.00µ1		E0900481-001 EQ0900256-03	1,000.00µL 1,000.00µL	E0900481-002 EQ0900256-04	1,000.00µL 1,000.00µL	E0900482-001 EQ0900256-05	1,000.00µL 1,000.00µL	
	Name: 8290/16	513B Cleanup Worki	ing Standard		Inventory ID 1	0913		Logbook Ref: ]	D10-52-5A/B	×		Expires On: 07/	13/2014	
I	E0900476-001 EQ0900256-03	100.00µL 100.00µL	E0900480-001 EQ0900256-06	100.00µL 100.00µL	E0900481-001	100.00µL		E0900481-002	100.00µL	E0900482-001	100.00µL	E0900483-001	100.00µL	1

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Preparation Information Benchsheet

Preparation Theoremation Benchsheet

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Page 1

				Preparati	on Informu	ttion Ben	chshee	ji a		
Pren Run#:	90735			Prel	22 WorkFlow: Or	gExtAq(365)			Status: Prepped	•
Теат:	Semivoa GCMS	S/AKODUR		· 4	rep Method: Me	sthod		4	rep Date/Time: 7/9/09 01:00 PN	, M
Preparatio	n Materials									
Acetone 99.5%	s Minimum	C2-16-007 (7199)		Carbon, High Purity	C2-23	3-004 (9452)		Ethyl Acetate 99.9% Minimum Fr0Ac	C2-23-006 (9462)	
Glass Wool		C2-13-005 (7198)		Sulfuric Acid Reagent	Grade C2-24	4-003 (9461)		Dichloromethane (Methylene Chloride) 00 9% McCl3	C2-25-001 (9449)	
Methanol HR-	GC Grade MeOH	C2-18-004 (7220)		Nacotium Chloride Reage Nacri	ent Grade C1-1(	04-2 (3306)		Curotico NaCH Grade NaCH	C2-24-002 (9463)	
Sodium Sulfat	¢ Anhydrous	C2-19-006 (7201)		Tridecane (n-Tridecane	c-27	4-001 (9460)		Hexane (n-Hexane) 98.5%	C2-25-004 (9441)	
Reagent Grads Nonane (n-No Toluene 99.9%	e Na2SO4 nane) 99% 6 Minimum	C2-21-004 (94 <i>57</i> ) C2-25-003 (9446)		pH Paper 0-14	(100	8)		Silica Gel Reagent Grade	C2-27-007 (9456)	
Preparatio	n Steps									
Step:	Extraction	Step:	Acid Clean	Step:	Silica Gel Clean	St	tep:	Final Volume		
Started:	00:51 60/6/2	Started:	7/10/09 08:00	Started:	7/10/09 13:00	ι Ν	larted:	7/13/09 07:30		
Finished:	7/9/09 18:00	Finished:	7/10/09 08:00	Finished:	7/10/09 17:30	E (	inished	7/13/09 14:00		
By:	NBROWN	By:	NBROWN	By:	NBROWN	ά.	×	NBKOWN		
Comments:										
Reviewed By:			Date:							
Chain of Custo	ły									
Relinquished	By:			Date:			Extracts E	xamined		
Received By:	Nico	le Brown		Date:	04/16/09		Yes	No		
					Page 22	2 of 22				Page 2

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Preparation Mitorination Benchsheet

Page 2

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シートのノート		460 W McNa	h Road Fr I	anderdale DI	( 33300 ( <b>U</b>							E DATE F	kequested	
	<b>Orida</b>	40 Alt. 27 Sou	uth Babson	auuerdale FI Park, FL 338	L 33309 827	Tel: (95 Tel: (86	4) 978-640 3) 638-325	0 9 1 1 1 1	IX: (954) 978 IX: (863) 638	-2233				
ogged into	onmental	30 Indian Stre 28 Gooch Roe	et Savannal ad Fort Mea	h, GA 31401 de FL 33841		Tel: (91 Tel: (86	2) 238-505 3) 285-814	ierer	м. (863) 234- к (912) 234- к (863) 285-	-503/ 4815 7030	RUS	H RESER	VATION #	
sport to: Diversified Drilling -Lehi	iah	Original-Rel	turn w/repo	ort	Yellow-La	th File Co	py	Pinl	- Sampler C	0DV	<b>.</b> <b>.</b> <b>.</b> <b>.</b>	ish Surchar	, , ,	
voim to. Niversified Duilling 1					Address: 56	20 Lee St. 1	(ehiah EY	13071					Supply apple	
	ligh	Purchase Order #			Invoice to			1/200						-
oject Name d/or Number : Primary & Secondary Tesiı	ing				Site	to ree St, I	Lengh, FL	33971	1					Τ
oject gr: Andrea Jennings	Phone	: 239-368-6404			Fax: 239-36	laleah, FL 8-6716				ail:				
mpler Name: 1 7									5					
inted) AVGOUO PIFF	FERRED / 4	10 GAR	leno		Sampler Signature	Ü	Alle							
b Control Number	Date Sampled	Time Sampled	Matrix	Bottle	Number of	1 4 10		وأيترفنه	Door					
	•		WS WD	Pres.	Containers Received			aryon	Inpan	rea				
haded Areas For boratory Use Only			GW SED S EFF HW BIO SA OU	Combo Codes	& NELAC Letter Suffixes	pəyət Şee	សាទព	Auepuc	 		1-201	Ч Н Н Н Д		LOF
			A UIL X		# A-?	S StfA	ŅЧ	boəg		)	ආ<	Α Υ 	24	
11439 5/P	67-05-CD	13:00	MO		00	×	×			0	2		م ن	
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ial Comments: aive NELAC nrotocol' (sign have)					Total	SAMPLE	CUSTON	V AND TD V	NORED CITY					
verables:	OA/OC Band	et Noododo				1	celinquished	by: Arg		ATUKES		DATE	/ TIME	
inle Custody & Brand Commonstor	nday o Soos	, Dapaayi 11	Yes N	o (additi	onal charge)	I	cecived by:	1					14: 31 ( 14: 31	h
	<u>bottle Type</u> ter amber		Pres	<u>ervatives</u>		2	elinquished	py:	Y T		9	22	22	
odv Seals? V N   1-life	acteria pag/pottie 10 ml fer hoffle		A-ascorbic a	cid P-H <sub>3</sub> 1 S-H <sub>2</sub> 5	0°.	2	eceived by:							
S:soil 1. b. a. m	11 <b>Jar</b> 30 ml		H-Cusu,	T-Na U-Un	<sub>2</sub> S <sub>2</sub> O <sub>3</sub> -H <sub>2</sub> O preserved	В	elinquished	by:			.			
	) ml vial Ade mouth	- <i>F</i> . (	N-MCAB V-NaOH MNH-CT		20. c acetate	е В	eceived by:					1		1
er Charges - Weekend Charge X-on						8	ww.flen	Wiro.col		C	DC Pane			120

July 15, 2009



Accounts Payable Florida Spectrum Environmental Services, Inc. 1460 West McNab Road Ft. Lauderdale, FL 33309

## Re: **#907-030, Miami Beach #907-097, Hialeah**

Dear Sir or Madam:

Enclosed are the data reports for Transmission Electron Microscope (TEM) analyses for asbestos in water samples. A known volume of water was filtered and dried. Each filter was coated with a thin film of carbon by evaporative deposition under vacuum. The samples were then cleared in acetone leaving the particles attached to the carbon film.

The samples were examined in the TEM at magnifications of 15,000-20,000X. A variable number of grid openings from each prepared grid were examined in the TEM for mineral identification and structure counts. The analytical instrument used was a Philips 300 with electron diffraction capability. Elemental analysis was performed with a Link Analytical QX200 Energy Dispersive Spectrometer (EDS)/ AN10000 Multi-Channel Analyzer.

These results reflect only the concentration of particles extracted from the provided water and observed in the sample preparation. These results relate only to those items or parts tested. The samples were collected by the client and therefore we have no knowledge as to the manner in which they were obtained. Data interpretation is provided by the consultant who obtained the samples. This report shall not be reproduced except in full, and with the written approval of E. M. Analytical, Inc.

Sincerely,

Pat Blackwelder, Ph.D. President

PB/at Enclosures



C	lient:	Florida Sp 1460 W M	bectrum Env //cNab Rd	vironmental S Ft: Lauderdal	ervices, Inc. e, FL, 33309				
F	roject/Local	tion: #907-030	, Miami Bea	ch	, _,				
E	.M. Number	r: 09EF-1							
S	ample ID:	115804	-l		Dreservation	Tuno: Blain			
S	ampled By:	Not Provi	0e0 20015		Preservation	rype. Plain			
	ate Collecte	ea: 07/01/09(	moa.19						
L	ate issued:	07706/09							
<u>FILTER</u>				INSTRUME	NT				
Type: Mixed	l Cellulose Es	ster		TEM: Philip	s EM 300				
Filter Size:	47 mm	4000		Magnificatio	n: 17,000 X	V			
Effective Filt	er Area: f Eiltor Analys	13∠0 mm <sup>−</sup> red: 0.153 mm	2	FDS <sup>1</sup> Link A	Analytical OX 20	Ď			
Total Alea O	r ritter Analyz	-eu. 0.103 min	•	Multi-Channel Analyzer: AN10000					
Date Recei	ved: 07/	02/09@1230		Analyst:	PB		2		
Date Filtere	ed: 07/	02/09@1230		Grid Openi	ng Area:	0.0102 m	m⁴		
Date Analy:	zed: 07/	03/09@1100		# Grid Ope	nings Counted	: 15			
Volume Filt	ered: 50 i	ml		Analytical Sensitivity: 0.18 MFL					
Particulate	Loading:	<1 %		Analysis M	ethod: EPA-10	10.2			
Non-Asbes	tos Structur	es: organics							
Grid	Fiber	Fiber Length	SAED	EDS	Asbestos	Negative	Spectrum		
Opening	#	(μm)	1D	ID	Туре	#	#		
1-15	NFD								

NFD: No Fibers Detected

SAED: Select Area Electron Diffraction; EDS: Elemental Dispersive Spectroscopy.

Upper and lower limits of the Poissonian 95% confidence interval of the asbestos structure concentration: N/A - 0.5382.

Total Number of Asbestos Fibers Detected: 0; Total Asbestos Fiber Concentration: < 0.18 MFL.

RESULTS			
	Number of Asbestos Fibers >10µ Detected: Concentration of Asbestos Fibers >10µ:	0 <0.18 MFL	

MCL = 7 Millions (MFL) longer than 10 microns. Target Analytical Sensitivity = 0.20 MFL; MFL; Millions of Fibers per Liter; MCL: Maximum Contaminant Level. Preparation and analyses utilized guidelines from Federal Register 40 CFR Parts 114, 142, 143, National Primary Drinking Water Regulations; Final Rule. Additional procedures are from "Transmission Electron Microscopy by EPA, Method: Analytical Method for Determination of Asbestos Fibers in Water", EPA-600/4-83-043. These results reflect only the concentration of particles extracted from the provided water and observed in the sample preparation. These results relate only to those items or parts tested. The samples were collected by the client and therefore we have no knowledge as to the manner in which they were obtained. Data interpretation is provided by the consultant who obtained the samples. The total number of pages in this report (including this page) is14. This report shall not be reproduced except in full, and with the written approval of E. M. Analytical, Inc. This report meets the requirements of the NELAC Standard and EPA Method 100.2. Submitted sample volume was < than that recommended by EPA Method 100.2... If there are any questions contact the signalee at the phone number below.

Lab Director's Signature:

1 as de



CI	ient:	Florida Sp 1460 W M	bectrum Envi IcNab Rd., F	ronmental Se t. Lauderdale	ervices, Inc. e, FL, 33309		
Pr E. Sa Da Da Da	oject/Location: M. Number: ample ID: ampled By: ate Collected: ate Issued:	#907-030 09EF-2 115805 Not Provid 07/01/09@ 07/06/09	, Miami Beac ded @0930	h	Preservation 1	Гуре: Plain	
FILTER				INSTRUMEN	<u>TI</u>		
Type: Mixed Filter Size: Effective Filte Total Area of	Cellulose Ester 47 mm r Area: 132 Filter Analyzed:	0 mm <sup>2</sup> 0.153 mm	1 <sup>2</sup>	TEM: Philips Magnification Accelerating EDS: Link A Multi-Channe	s EM 300 n: 17,000 X Voltage: 100 k\ nalytical QX 200 el Analyzer: AN	V ) 10000	
Date Receiv Date Filtered Date Analyz Volume Filte Particulate L Non-Asbesto	ed: 07/02/0 d: 07/02/0 ed: 07/03/0 ered: 50 ml .oading: <1 0 os Structures:	9@1230 9@1230 9@1100 % organics, fit	perglass	Analyst: Grid Openin # Grid Ope Analytical S Analysis Me	PB ng Area: nings Counted: Sensitivity: ethod: EPA-10	0.0102 mr 15 0.18 MFL 0.2	n²
Grid	Fiber Fit	per Length	SAED 1D	EDS ID	Asbestos Type	Negative #	Spectrum #
1-15	" NFD	(min)			21		

NFD: No Fibers Detected

SAED: Select Area Electron Diffraction; EDS: Elemental Dispersive Spectroscopy.

Upper and lower limits of the Poissonian 95% confidence interval of the asbestos structure concentration: N/A - 0.5382.

Total Number of Asbestos Fibers Detected: 0; Total Asbestos Fiber Concentration: <0.18 MFL.

RESULTS		
	Number of Asbestos Fibers >10µ Detected: Concentration of Asbestos Fibers >10µ:	0 <0.18 MFL

MCL = 7 Millions (MFL) longer than 10 microns. Target Analytical Sensitivity = 0.20 MFL; MFL: Millions of Fibers per Liter; MCL: Maximum Contaminant Level. Preparation and analyses utilized guidelines from Federal Register 40 CFR Parts 114, 142, 143, National Primary Drinking Water Regulations; Final Rule. Additional procedures are from "Transmission Electron Microscopy by EPA, Method: Analytical Method for Determination of Asbestos Fibers in Water", EPA-600/4-83-043. These results reflect only the concentration of particles extracted from the provided water and observed in the sample preparation. These results relate only to those items or parts tested. The samples were collected by the client and therefore we have no knowledge as to the manner in which they were obtained. Data interpretation is provided by the consultant who obtained the samples. The total number of pages in this report (including this page) Is14. This report shall not be reproduced except in full, and with the written approval of E. M. Analytical, Inc. This report meets the requirements of the NELAC Standard and EPA Method 100.2. Submitted sample volume was < than that recommended by EPA Method 100.2... If there are any questions contact the signatee at the phone number below.

Lab Director's Signature:

.Oh



C	Client: Florida Spectrum Env 1460 W McNab Rd.				ervices, Inc. e, FL, 33309		
F E S S C C	Project/Locati E.M. Number: Sample ID: Sampled By: Date Collected Date Issued:	on: #907-030 09EF-3 115806 Not Provid 07/01/09@ 07/06/09	, Miami Bea ded @0950	ach			
FILTER				INSTRUME	T		
Type: Mixed Filter Size: Effective Filt Total Area o	l Cellulose Est 47 mm er Area: f Filter Analyze	er 1320 mm <sup>2</sup> ed: 0.153 m <b>n</b>	1 <sup>2</sup>	TEM: Philip Magnificatio Accelerating EDS: Link A Multi-Chann	s EM 300 n: 17,000 X g Voltage: 100 k Analytical QX 20 el Analyzer: AN	V 0 10000	
Date Received:07/02/09@1230Date Filtered:07/02/09@1230Date Analyzed:07/03/09@1100Volume Filtered:50 mlParticulate Loading:1 %Non-Aspestos Structures:organics			Analyst: Grid Openi # Grid Ope Analytical S Analysis M	PB ng Area: nings Counted Sensitivity: ethod: EPA-10	0.0102 m : 15 0.18 MFL 00.2	m²	
Grid	Fiber	Fiber Length	SAED	EDS	Asbestos	Negative #	Spectrum #
Opening	#	(μm)	טו	U.	i ype	#	IT.
1-15	NFD						

NFD: No Fibers Detected

SAED: Select Area Electron Diffraction; EDS: Elemental Dispersive Spectroscopy. Upper and lower limits of the Poissonian 95% confidence interval of the asbestos structure concentration: N/A - 0.5382.

Total Number of Asbestos Fibers Detected: 0; Total Asbestos Fiber Concentration: <0.18 MFL.

RESULTS			
	Number of Asbestos Fibers >10µ Detected: Concentration of Asbestos Fibers >10µ:	0 <0.18 MFL	

MCL = 7 Millions (MFL) longer than 10 microns. Target Analytical Sensitivity = 0.20 MFL; MFL: Millions of Fibers per Liter; MCL: Maximum Contaminant Level. Preparation and analyses utilized guidelines from Federal Register 40 CFR Parts 114, 142, 143, National Primary Drinking Water Regulations; Final Rule. Additional procedures are from "Transmission Electron Microscopy by EPA, Method: Analytical Method for Determination of Asbestos Fibers in Water", EPA-600/4-83-043. These results reflect only the concentration of particles extracted from the provided water and observed in the sample preparation. These results relate only to those items or parts tested. The samples were collected by the client and therefore we have no knowledge as to the manner in which they were obtained. Data interpretation is provided by the consultant who obtained the samples. The total number of pages in this report (including this page) is14. This report shall not be reproduced except in full, and with the written approval of E. M. Analytical, Inc. This report meets the requirements of the NELAC Standard and EPA Method 100.2. Submitted sample volume was < than that recommended by EPA Method 100.2... If there are any questions contact the signatee at the phone number below.

Lab Director's Signature:



C	Client:	Florida S 1460 W M	pectrum En McNab Rd	vironmental S Ft. Lauderdal	ervices, Inc. e. FL, 33309		
F	Project/Locat E.M. Numbe	tion: #907-030 r: 09EF-4	), Miami Bea	ach.			
S S [ [	Sample ID: Sampled By: Date Collecte Date Issued:	115807 Not Provi ed: 07/01/09( 07/06/09	ded @1005		Preservation	Type: Plain	
FILTER				INSTRUME	NT		
Type: Mixed Cellulose Ester Filter Size: 47 mm Effective Filter Area: 1320 mm <sup>2</sup> Total Area of Filter Analyzed: 0.153 mm <sup>2</sup>			TEM: Philip Magnificatio Accelerating EDS: Link / Multi-Chann	s EM 300 m: 17,000 X g Voltage: 100 k Analytical QX 20 rel Analyzer: AN	V 0 110000		
Date Received:07/02/09@1230Date Filtered:07/02/09@1230Date Analyzed:07/03/09@1100Volume Filtered:50 mlParticulate Loading:1 %Non-Asbestos Structures:organics			Analyst: Grid Openi # Grid Ope Analytical \$ Analysis M	PB ing Area: enings Counted Sensitivity: lethod: EPA-10	0.0102 m 15 0.18 MFL 00.2		
Grid Opening	Fiber #	Fiber Length (μm)	SAED ID	EDS ID	Asbestos Type	Negative #	Spectrum #
. °	NFD						

NFD: No Fibers Detected

SAED: Select Area Electron Diffraction; EDS: Elemental Dispersive Spectroscopy.

Upper and lower limits of the Poissonian 95% confidence interval of the asbestos structure concentration: N/A - 0.5382.

Total Number of Asbestos Fibers Detected: 0; Total Asbestos Fiber Concentration: <0.18 MFL.

RESULTS			
	Number of Asbestos Fibers >10µ Detected: Concentration of Asbestos Fibers >10µ:	0 <0.18 MFL	:

MCL = 7 Millions (MFL) longer than 10 microns. Target Analytical Sensitivity = 0.20 MFL; MFL: Millions of Fibers per Liter; MCL: Maximum Contaminant Level. Preparation and analyses utilized guidelines from Federal Register 40 CFR Parts 114, 142, 143, National Primary Drinking Water Regulations; Final Rule. Additional procedures are from 'Transmission Electron Microscopy by EPA, Method: Analytical Method for Determination of Asbestos Fibers in Water', EPA-600/4-83-043. These results reflect only the concentration of particles extracted from the provided water and observed in the sample preparation. These results relate only to those items or parts tested. The samples were collected by the client and therefore we have no knowledge as to the manner in which they were obtained. Data interpretation is provided by the consultant who obtained the samples. The total number of pages in this report (including this page) Is14. This report shall not be reproduced except in full, and with the written approval of E. M. Analytical, Inc. This report meets the requirements of the NELAC Standard and EPA Method 100.2. Submitted sample volume was < than that recommended by EPA Method 100.2... If there are any questions contact the signatee at the phone number below.

Lab Director's Signature:



C	Client:	Florida Spectrum Environmental Services, Inc. 1460 W McNab Rd., Ft. Lauderdale, FL, 33309					
F S S C D	Project/Location: E.M. Number: Sample ID: Sampled By: Date Collected: Date Issued:	#907-030 09EF-5 115808 Not Provid 07/01/09@ 07/06/09	, Miami Beac ded @1015	h	Preservation	Type: Plain	
<u>FILTER</u>				INSTRUMEN	<u>TI</u>		
Type: Mixed Filter Size: Effective Filt Total Area of	l Cellulose Ester 47mm er Area: 1320 f Filter Analyzed:	mm <sup>2</sup> 0.153 mm	2	TEM: Philips Magnification Accelerating EDS: Link A Multi-Channe	s EM 300 n: 17,000 X Voltage: 100 k nalytical QX 200 el Analyzer: AN	V ) 10000	
Date Received:07/02/09@1230Date Filtered:07/02/09@1230Date Analyzed:07/03/09@1100Volume Filtered:50 mlParticulate Loading:1 %Non-Asbestos Structures:organics				Analyst: Grid Openir # Grid Oper Analytical S Analysis Me	PB ng Area: nings Counted: ensitivity: ethod: EPA-10	0.0102 m 15 0.18 MFL 0.2	m²
Grid	Fiber Fibe	r Length	SAED	EDS	Asbestos	Negative	Spectrum
Opening	# (	μm)	ID	ID	Туре	#	#
1-15	NFD						

NFD: No Fibers Detected

SAED: Select Area Electron Diffraction; EDS: Elemental Dispersive Spectroscopy.

Upper and lower limits of the Poissonian 95% confidence interval of the asbestos structure concentration: N/A - 0.5382.

Total Number of Asbestos Fibers Detected: 0; Total Asbestos Fiber Concentration: <0.18 MFL.

RESULTS			
	Number of Asbestos Fibers >10µ Detected: Concentration of Asbestos Fibers >10µ:	0 <0.18 MFL	

MCL = 7 Millions (MFL) longer than 10 microns. Target Analytical Sensitivity = 0.20 MFL; MFL: Millions of Fibers per Liter; MCL: Maximum Contaminant Level. Preparation and analyses utilized guidelines from Federal Register 40 CFR Parts 114, 142, 143, National Primary Drinking Water Regulations; Final Rule. Additional procedures are from "Transmission Electron Microscopy by EPA, Method: Analytical Method for Determination of Asbestos Fibers in Water", EPA-600/4-83-043. These results reflect only the concentration of particles extracted from the provided water and observed in the sample preparation. These results relate only to those items or parts tested. The samples were collected by the client and therefore we have no knowledge as to the manner in which they were obtained. Data interpretation is provided by the consultant who obtained the samples. The total number of pages in this report (including this page) is14. This report shall not be reproduced except in full, and with the written approval of E. M. Analytical, Inc. This report meets the requirements of the NELAC Standard and EPA Method 100.2. Submitted sample volume was < than that recommended by EPA Method 100.2.. If there are any questions contact the signatee at the phone number below.

Lab Director's Signature:

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#### Asbestos in Water Report/Count Sheet

(	Client:	Florida S 1460 W I	Florida Spectrum Environmental Services, Inc. 1460 W McNab Rd., Ft. Lauderdale, FL. 33309				
	Project/Location: E.M. Number: Sample ID: Sampled By:	#907-030 09EF-6 115809 Not Provi	#907-030, Miami Beach 09EF-6 115809				
	Date Collected: Date Issued:	07/01/09	@1030				
<u>FILTER</u>				INSTRUME	<u>NT</u>		
Type: Mixe Filter Size: Effective Fil Total Area c	d Cellulose Ester 47 mm ter Area: 132 of Filter Analyzed:	0 mm <sup>2</sup> 0.153 mr	n²	TEM: Philip Magnificatio Accelerating EDS: Link A Multi-Chann	s EM 300 n: 17,000 X g Voltage: 100 k Analytical QX 20 lel Analyzer: AN	V 0 110000	
Date Received:07/02/09@1230Date Filtered:07/02/09@1230Date Analyzed:07/03/09@1100Volume Filtered:50 mlParticulate Loading:1 %Non-Aspestos Structures:organics			Analyst: Grid Openi # Grid Ope Analytical S Analysis M	PB ng Area: enings Counted Sensitivity: ethod: EPA-10	0.0102 m I: 15 0.18 MFL 00.2	m²	
Grid Opening	Fiber Fit #	er Length (μm)	SAED 1D	EDS ID	Asbestos Type	Negative #	Spectrum #
1-15	NFD						

NFD: No Fibers Detected SAED: Select Area Electron Diffraction; EDS: Elemental Dispersive Spectroscopy. Upper and lower limits of the Poissonian 95% confidence interval of the asbestos structure concentration: N/A - 0.5382.

Total Number of Asbestos Fibers Detected: 0; Total Asbestos Fiber Concentration: <0.18 MFL.

RESULTS			
	Number of Asbestos Fibers >10 $\mu$ Detected: Concentration of Asbestos Fibers >10 $\mu$ :	0 <0.18 MFL	

MCL = 7 Millions (MFL) longer than 10 microns. Target Analytical Sensitivity = 0.20 MFL; MFL: Millions of Fibers per Liter; MCL: Maximum Contaminant Level. Preparation and analyses utilized guidelines from Federal Register 40 CFR Parts 114, 142, 143, National Primary Drinking Water Regulations; Final Rule. Additional procedures are from "Transmission Electron Microscopy by EPA, Method: Analytical Method for Determination of Asbestos Fibers in Water", EPA-600/4-83-043. These results reflect only the concentration of particles extracted from the provided water and observed in the sample preparation. These results relate only to those items or parts tested. The samples were collected by the client and therefore we have no knowledge as to the manner in which they were obtained. Data interpretation is provided by the consultant who obtained the samples. The total number of pages in this report (including this page) is14. This report shall not be reproduced except in full, and with the written approval of E. M. Analytical, Inc. This report meets the requirements of the NELAC Standard and EPA Method 100.2. Submitted sample volume was < than that recommended by EPA Method 100.2.. If there are any questions contact the signatee at the phone number below.

Lab Director's Signature:



	Client:	Florida S 1460 W	pectrum En McNab Rd	ivironmental S Ft. Lauderdal	ervices, Inc. le. FL 33309		
	Project/Location E.M. Number: Sample ID: Sampled By: Date Collected:	1400 W #907-030 09EF-7 115810 Not Prov 07/01/09	#907-030, Miami Beach 09EF-7 115810 Not Provided Preservation Type: Plain 07/01/09@1045				
	Date Issued:	07/06/09					
<u>FILTER</u> Type: Mixed Cellulose Ester Filter Size: 47 mm Effective Filter Area: 1320 mm <sup>2</sup> Total Area of Filter Analyzed: 0.153 mm <sup>2</sup>				TEM: Philip Magnificatio Accelerating EDS: Link / Multi-Chann	n: 17,000 X n: 17,000 X Voltage: 100 k Analytical QX 20 nel Analyzer: Analyz	(V 0 110000	
Date Received:07/02/09@1230Date Filtered:07/02/09@1230Date Analyzed:07/03/09@1100Volume Filtered:50 mlParticulate Loading:1 %Non-Asbestos Structures:organics			Analyst: Grid Openi # Grid Ope Analytical \$ Analysis M	PB ing Area: enings Counted Sensitivity: ethod: EPA-10	0.0102 m I: 15 0.18 MFL 00.2	m²	
Grid Opening	Fiber F #	iber Length (μm)	SAED ID	EDS ID	Asbestos Type	Negative #	Spectrum #
1-15	NFD						

NFD: No Fibers Detected SAED: Select Area Electron Diffraction; EDS: Elemental Dispersive Spectroscopy. Upper and lower limits of the Poissonian 95% confidence interval of the asbestos structure concentration: N/A - 0.5382.

Total Number of Asbestos Fibers Detected: 0; Total Asbestos Fiber Concentration: <0.18 MFL.

RESULTS			
	Number of Asbestos Fibers >10 $\mu$ Detected: Concentration of Asbestos Fibers >10 $\mu$ :	0 <0.18 MFL	

MCL = 7 Millions (MFL) longer than 10 microns. Target Analytical Sensitivity = 0.20 MFL; MFL: Millions of Fibers per Liter; MCL: Maximum Contaminant Level. Preparation and analyses utilized guidelines from Federal Register 40 CFR Parts 114, 142, 143, National Primary Drinking Water Regulations; Final Rule. Additional procedures are from "Transmission Electron Microscopy by EPA, Method: Analytical Method for Determination of Asbestos Fibers in Water", EPA-600/4-83-043. These results reflect only the concentration of particles extracted from the provided water and observed in the sample preparation. These results relate only to those items or parts tested. The samples were collected by the client and therefore we have no knowledge as to the manner in which they were obtained. Data interpretation is provided by the consultant who obtained the samples. The total number of pages in this report (including this page) is14. This report shall not be reproduced except in full, and with the written approval of E. M. Analytical, Inc. This report meets the requirements of the NELAC Standard and EPA Method 100.2. Submitted sample volume was < than that recommended by EPA Method 100.2.. If there are any questions contact the signatee at the phone number below.

Lab Director's Signature:



	Client: Florida Spectrum Env 1460 W McNab Rd.				vironmental Services, Inc. Ft. Lauderdale, FL, 33309				
	Project/Loca	tion: #907-030	), Miami Bea	ich					
	E.M. Numbe	r: 09EF-8							
-	Sample ID:	115811			<b>—</b>				
	Sampled By:	Not Provi	ded Adda		Preservation	Type: Plain			
	Date Collecte	ea: 07/01/09/ 07/06/09	@1100						
	Date Issued.	01100/03							
<u>FILIER</u>				INSTRUME	<u>NI</u>				
Type: Mixe	d Cellulose Es	ster		TEM: Philip	os EM 300				
Filter Size:	47 mm	$1320 \text{ mm}^2$		Acceleration: 17,000 A					
Total Area o	of Filter Analy	zed: 0.153 mr	n <sup>2</sup>	EDS: Link	EDS: Link Analytical QX 200				
(old) for a				Multi-Channel Analyzer: AN10000					
Date Rece	ived: 07/	02/09@1230		Analyst:	PB		_		
Date Filter	ed: 07/	02/09@1230		Grid Opening Area: 0.0102 mm <sup>2</sup>					
Date Analy	/zed: 07/	03/09@1100		# Grid Ope	enings Counted	: 15			
Volume Fil	tered: 50	ml		Analytical Sensitivity: 0.18 MFL					
Particulate Loading: 1 %				Analysis M	ethod: EPA-10	00.2			
Non-Asbes	stos Structure	es: organics							
Grid	Fiber	Fiber Length	SAED	EDS	Asbestos	Negative	Spectrum		
Opening	#	(μm)	ID	ID	Туре	#	#		
1-15	NFD								

NFD: No Fibers Detected SAED: Select Area Electron Diffraction; EDS: Elemental Dispersive Spectroscopy. Upper and lower limits of the Poissonian 95% confidence interval of the asbestos structure concentration: N/A - 0.5382.

Total Number of Asbestos Fibers Detected: 0; Total Asbestos Fiber Concentration: <0.18 MFL.

RESULTS			
	Number of Asbestos Fibers >10 $\mu$ Detected: Concentration of Asbestos Fibers >10 $\mu$ :	0 <0.18 MFL	

MCL = 7 Millions (MFL) longer than 10 microns. Target Analytical Sensitivity = 0.20 MFL; MFL: Millions of Fibers per Liter; MCL: Maximum Contaminant Level. Preparation and analyses utilized guidelines from Federal Register 40 CFR Parts 114, 142, 143, National Primary Drinking Water Regulations; Final Rule. Additional procedures are from "Transmission Electron Microscopy by EPA, Method: Analytical Method for Determination of Asbestos Fibers in Water", EPA-600/4-83-043. These results reflect only the concentration of particles extracted from the provided water and observed in the sample preparation. These results relate only to those items or parts tested. The samples were collected by the client and therefore we have no knowledge as to the manner in which they were obtained. Data interpretation is provided by the consultant who obtained the samples. The total number of pages in this report (including this page) is14. This report shall not be reproduced except in full, and with the written approval of E. M. Analytical, Inc. This report meets the requirements of the NELAC Standard and EPA Method 100.2. Submitted sample volume was < than that recommended by EPA Method 100.2.. If there are any questions contact the signatee at the phone number below.

Lab Director's Signature:

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(	Client: Florida Spec 1460 W McN			Environmental Services, Inc. Rd., Ft. Lauderdale, FL, 33309			
	Project/Location E.M. Number: Sample ID: Sampled By: Date Collected: Date Issued:	Asbestos 09EF-WE N/A N/A N/A@ 07/06/09	in Water 3	Water Preservation Type: N/A			
FILTER				INSTRUME	T		
Type: Mixed Cellulose Ester Filter Size: 47 mm Effective Filter Area: 1320 mm <sup>2</sup> Total Area of Filter Analyzed: 0.153 mm <sup>2</sup>			TEM: Philips EM 300 Magnification: 17,000 X Accelerating Voltage: 100 kV EDS: Link Analytical QX 200 Multi-Channel Analyzer: AN10000				
Date Received:N/A@Date Filtered:07/02/09@1230Date Analyzed:07/03/09@1100Volume Filtered:50 mlParticulate Loading:<1 %			Analyst: PB Grid Opening Area: 0.0102 mm <sup>2</sup> # Grid Openings Counted: 15 Analytical Sensitivity: 0.18 MFL Analysis Method: EPA-100.2				
Grid Opening	Fiber F #	ber Length (μm)	SAED ID	EDS ID	Asbestos Type	Negative #	Spectrum #
1-15	NFD						

NFD: No Fibers Detected

SAED: Select Area Electron Diffraction; EDS: Elemental Dispersive Spectroscopy.

Upper and lower limits of the Poissonian 95% confidence interval of the asbestos structure concentration: N/A - 0.5382.

Total Number of Asbestos Fibers Detected: 0; Total Asbestos Fiber Concentration: <0.18 MFL.

RESULTS			
	Number of Asbestos Fibers >10µ Detected: Concentration of Asbestos Fibers >10µ:	0 <0.18 MFL	

MCL = 7 Millions (MFL) longer than 10 microns. Target Analytical Sensitivity = 0.20 MFL; MFL: Millions of Fibers per Liter; MCL: Maximum Contaminant Level. Preparation and analyses utilized guidelines from Federal Register 40 CFR Parts 114, 142, 143, National Primary Drinking Water Regulations; Final Rule. Additional procedures are from "Transmission Electron Microscopy by EPA, Method: Analytical Method for Determination of Asbestos Fibers in Water", EPA-600/4-83-043. These results reflect only the concentration of particles extracted from the provided water and observed in the sample preparation. These results relate only to those items or parts tested. The samples were collected by the client and therefore we have no knowledge as to the manner in which they were obtained. Data interpretation is provided by the consultant who obtained the samples. The total number of pages in this report (including this page) is14. This report shall not be reproduced except in full, and with the written approval of E. M. Analytical, Inc. This report meets the requirements of the NELAC Standard and EPA Method 100.2. Submitted sample volume was N/A than that recommended by EPA Method 100.2.. If there are any questions contact the signatee at the phone number below.

Lab Director's Signature:

Appendix C

**Geophysical Logs** 

FIELD:HIALEAHSTATE:FLORIDACOUNTRY:FLORIDADate Logged:8–JUNE–2009Well Location:LAT: 25* 55.447'LONG: 80* 22.195'Elevations:K.B. 5 ftD.F. 5 ftG.L. 5 ftAPI Number:Job Number:	HNGS*, ECS*, PSP* FMI CMR COMPANY: CITY OF HIALEAH WELL: HIALEAH UPPER FLORIDAN TEST WELL	"A Mark of Schlumberger Using the following logs: AIT*, MCFL*, FMI* CNT*. TLD*. DSI*	Integrated Wireline Log Well Composite	GEOFRAME PROCESSED INTERPRETATION
FOLD HERE The well name, lo	cation and borehole referen	ce data were furnished b	y the customer	
All interpretations are opinions based on infere correctness of any interpretation, and we will n costs, damages oe expenses incurred or susta These interpretations are also subject to Claus Ser. Oder # B2EO00014 OP Vers.	ences from electrical or other measure not, except in the case gross or willf ained by anyone resulting from any se 4 of our General Terms and Con .: 16C0–147 Process Date	arements and we cannot, and do al negligence on our part, be liat interpretations made by any of o ditions as set out in our Price So e: Sept–2009 Center: S	o not guarantee the accuracy or ole or responsible for any loss, our officers, agents or employees. chedule.	Log Analyst: N. Cla
Mud and Borehole Measurements:			1	
Rm @ Measured Temperature: 1.6	ohm.m @ 85 c BHT: 78	3 degF	Bitsize: 15 in	
Rmf @ Measured Temperature: 1.6	ohm.m @ 85 G Type Flu	uid in Hole:	FGM:	

Remarks:

Rmc @ Measured Temperature: 1.6 ohm.m @ 85

Depth reference is drill floor (5 ft above ground surface).

Severe washout (cavity) at 1160–1260 ft compromising log quality.

Certain log measurements are unattainable in PVC casing (above 1088 ft).

Mud density: 8.34 lbm/gal

FGM:

			Large Pore Poro.
		K (Carb. Analysis)	Small Pore Porosity
		RT < RI	Capillary Bound Flu
	Very Small Pores	RT > RI	Neutron Poro. (LS)           0         (ft3/ft3)         1
		BH Fluid Resi.	MR Free Fluid Poro.
Uran.	Bin 2	0.1 (ohm.m) 1000	0 (ft3/ft3) 1

							Irr Water
							Water
				]			
			Large Pore Poro.				
		K (Carb. Analysis)	Small Pore Porosity				Dolomite
		RT < RI	Capillary Bound Flu			MR Flow Profile	
I			Neutron Poro, (LS)				
	Very Small Pores	RT > RI	0 (ft3/ft3) 1			Salinity from ECS	Pyrite
Uran.	Bin 2	BH Fluid Resi.	MR Free Fluid Poro.			Carb. An. Flow Prof	Quartz
		Carb. Anal. K	MR 3 ms. Poro.	-		Spinner Flow Prof.	· · · · · · · · · · · · · · · · · · ·
Washout	Bin 3	0.01 (mD) 100	0 (ft3/ft3) 1	-	L	0 (bbl/d) 200000	Carbonate Apatite
$\frac{\text{Tot. GR}}{0}$	Bin 4	MR SDR Hyd. Cond. 0.01 (mD) 100	MR Total Porosity 0 (ft3/ft3) 1		Static Norm. Horizontal Scale: 1 : 23.562	Salinity from ECS	Bound Water
HD2_PPC1	Bin 5	Micro_Resistivity	Density Porosity	-	Straight Image	Spinner Fluid Velo.	Kaolinite
5 <sub>(in)</sub> 35 Bitsize	Dino	0.1 (ohm.m) 1000	0 (ft3/ft3) 1	MR T2 Dist	0 120 240 360 Resistive FMI Image Conductive	0 (ft/min) 150	
5 (in) 35	Bin 6	0.1 (ohm.m) 1000	0 (ft3/ft3) 1				Illite
K+Th GR	Bin 7	90 in. DOI Resi.	ELAN Total Poro.	T2Cutoff Short	N	BH Fluid Temp.	ELAN Mat.+Pore Vol.
ິ (gAPI) <sup>ວບ</sup> SP		Sonic. Frac. Aper.	o         (ft3/ft3)         1           Macro Poro. (Carb.         1	0.3 (ms) 3000 MR T2 Log Mean	- W 🔆 E	MR SDR Flow Profile	· (V/V) 0
<sup>0</sup> (mV) <sup>35</sup>	Very Large Pores	0.001 ( in ) 10	0 (ft3/ft3) 1	0.3 (ms) 3000	S	0 () 1	Pyrite Dry Wt.
мп 1:500	MR Total Porosity	Carb. Anal. Tran.	Macro Poro. (FMI)	T2CCutoff Long	True Dip	Spinner Norm. Flow	O.5 ( lbf/lbf ) 0
100 150 200 250 300 300 300 400							
500							



Appendix D

**APT Data (Attached Digitally)**